

# Rich defects boosting the oxygen evolution reaction

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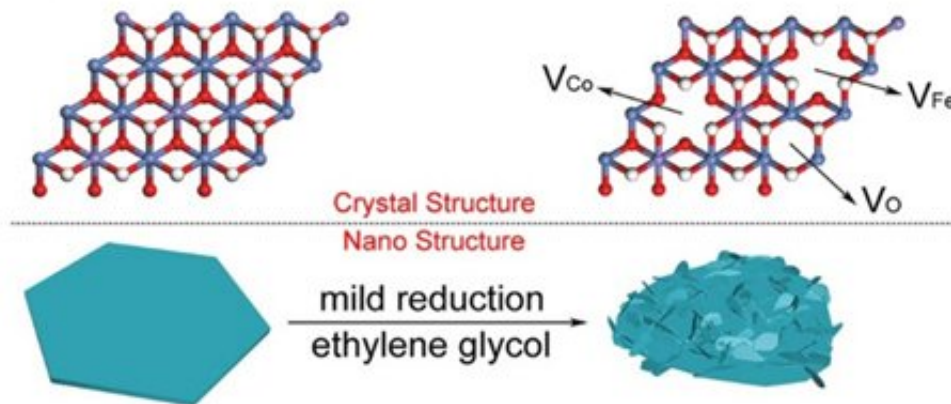


Illustration of the solvothermal reaction of pristine CoFe LDHs by using ethylene glycol. Credit: ©Science China Press

The oxygen evolution reaction (OER) with sluggish reaction kinetics and large over-potential is the severe reaction in water splitting that seems promising for energy storage and conversion. However, it is still the bottleneck reaction of the water-splitting system because of the slow kinetics and large over-potential during the anodic polarization process. Therefore, it is crucial to develop highly efficient OER catalysts which can lower the over-potential effectively and accelerate reaction kinetics.

At present, CoFe double metal oxides or hydroxides have been proved

by many studies to be efficient catalysts for catalyzing OER. However, the performance of the corresponding bulk catalysts is still unsatisfactory in practical applications. Based on this, it is of considerable significance to achieve the simultaneous improvement of the apparent activity and intrinsic activity of CoFe-based catalysts through material nanostructure engineering and electronic structure regulation.

Recently, Professor Shuangyin Wang's group from Hunan University, based on the strategy of defect engineering, used a mild reducing agent-ethylene glycol as a solvent in the solvothermal reduction of bulk CoFe LDHs to achieve defects construction. This treatment facilitated the formation of anion and cation defects (O, Co, and Fe), and the bulk CoFe LDHs were *in situ* exfoliated, and a three-dimensional hierarchical structure was formed due to the intercalation effect of large size ethylene glycol during the solvothermal process.

After further morphology and electronic structure characterization, the authors found that the defect-rich structure significantly increased the intrinsic activity of the material, and the resulting 3-dimensional hierarchical structure promoted the [mass transfer](#) in the catalytic process, ultimately achieving effective OER performance.

Compared with the conventional two-dimensional material exfoliation or defect construction methods, this method breaks through the bottleneck of scale-up exfoliation of two-dimensional material, and the exfoliation of two-dimensional [catalyst](#) and in-situ development of three-dimensional [structure](#) are realized by a simple one-step solvothermal method. It provides a new direction for the large-scale preparation and application of OER catalysts.

**More information:** Peng Zhou et al, Single-crystalline layered double hydroxides with rich defects and hierarchical structure by mild reduction

for enhancing the oxygen evolution reaction, *Science China Chemistry* (2019). [DOI: 10.1007/s11426-019-9511-x](https://doi.org/10.1007/s11426-019-9511-x)

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