

June 6 2019, by Bob Yirka

Researchers find a way to produce freestanding films of perovskite oxides

b a C Film PDMS H₂O Sr3Al2O6 Silicon Substrate As-grown Released Transferred d f 2 unit cells thick 4 unit cells thick 5 nm Si e g 2 unit cells thick 4 unit cells thick 1 μm 2 µm

Growth and transfer of ultrathin freestanding SrTiO3 films. a, Schematic of a film with an SAO buffer layer. b, The sacrificial SAO layer is dissolved in water



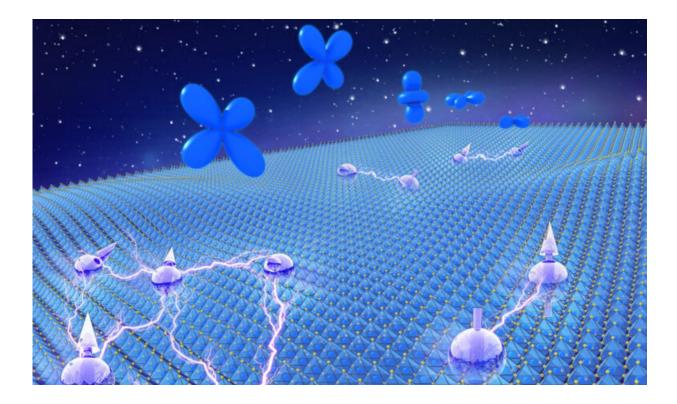
to release the top oxide films with the mechanical support of PDMS. c, New heterostructures and interfaces are formed when the freestanding film is transferred onto the desired substrate. d, e, Atomically resolved cross-sectional (d) and low-magnification plan-view (e) HAADF images of a two-unit-cell freestanding STO film transferred to a silicon wafer and a holey carbon TEM grid, respectively. f, g, Atomically resolved cross-sectional (f) and low-magnification planview (g) HAADF images of a representative four-unit-cell freestanding STO film, showing the excellent flexibility of ultrathin freestanding films. Credit: *Nature* (2019). DOI: 10.1038/s41586-019-1255-7

A team of researchers from Nanjing University in China, the University of Nebraska and the University of California in the U.S. has found a way to produce free-standing films of perovskite oxide. In their paper published in the journal *Nature*, the group describes the process they developed and how well it worked when tested. Yorick Birkhölzer and Gertjan Koster from the University of Twente have published a News and Views <u>piece</u> on the work done by the team in the same journal issue.

Birkhölzer and Koster point out that many <u>new materials</u> are made by going to extremes—making them really big or really small. Making them small has led to many recent discoveries, they note, including a technique to make graphene. One area of research has focused on ways to produce transition-metal oxides in a thinner format. It has been slow going, however, due to their crystalline nature. Unlike some materials, transition-metal oxides do not naturally form into layers with a top layer that can be peeled off. Instead, they form in strongly bonded 3-D structures. Because of this, some in the field have worried that it might never be possible to produce them in desired forms. But now, the researchers with this new effort have found a way to produce two transition-metal oxides (perovskite oxides strontium titanate and bismuth ferrite) in a thin-film format.



The process developed by the researchers involved using <u>molecular</u> <u>beam epitaxy</u> to apply a buffer layer onto a substrate followed by a layer of perovskite. Once the sandwich of materials was made, the researchers used water to dissolve the buffer <u>layer</u>, allowing the perovskite to be removed and placed onto other substrates. The researchers report that their process worked so well they were able to extract <u>films</u> of <u>perovskite</u> near the theoretical limit—one square unit cell (with approximately 0.4-nanometer sides).



"Through our successful fabrication of ultrathin perovskite oxides down to the monolayer limit, we've created a new class of two-dimensional materials," says Xiaoqing Pan, professor of materials science & engineering and Henry Samueli Endowed Chair in Engineering at UCI. "Since these crystals have strongly correlated effects, we anticipate they will exhibit qualities similar to graphene that will be foundational to next-generation energy and information technologies." Credit: Xiaoqing Pan / UCI



Birkhölzer and Koster point out that the work done by the combined Chinese and American team demonstrated that it is possible to produce at least some transition-metal oxides in a thin film format. Their research also allayed fears that such a film would collapse, making it unusable.

More information: Dianxiang Ji et al. Freestanding crystalline oxide perovskites down to the monolayer limit, *Nature* (2019). <u>DOI:</u> <u>10.1038/s41586-019-1255-7</u>

Press release

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Citation: Researchers find a way to produce free-standing films of perovskite oxides (2019, June 6) retrieved 2 May 2024 from <u>https://phys.org/news/2019-06-free-standing-perovskite-oxides.html</u>

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