

Solar-powered cup brews better brine treatments

April 10 2019



The 3-D-cup-shaped solar evaporator at increasing time intervals (l-r) at 0, 24, 48, 72, 96 and 120 hours. Credit: King Abdullah University of Science and Technology

A new distillation device can better recycle contaminants produced by a desalination plant to dramatically reduce waste.

When <u>sea water</u> is purified using reverse osmosis membranes, a liquid brine containing salts and other impurities is often disposed of directly into the environment. Strict regulations are now pushing desalination

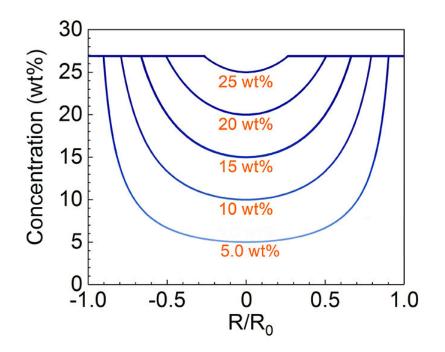


plants to adopt zero liquid discharge principles to treat this <u>waste water</u>. Current technologies, however, rely on expensive electrical or thermal treatments to concentrate the brine.

A more <u>sustainable approach</u> is now being investigated by Peng Wang and colleagues from the Water Desalination and Reuse Centerat KAUST. The team is developing solar photothermal distillation devices that can convert sunlight into heat. When placed upon a typical sample of liquid brine, these photothermal devices can evaporate water with 90 percent energy efficiency.

One problem with using solar distillation devices to isolate clean water from brine is that as liquid evaporates there is a sharp increase in <u>salt</u> concentrations in the effluent. This can cause a thick crust to form on top of the photothermal material, blocking sunlight absorption and reducing the distillation rate dramatically.





The ratio (R/Ro) of the radius of the salt-uncovered area (R) to the radius of the photothermal disk (Ro) relative to brine concentration. Credit: King Abdullah University of Science and Technology

Wang and colleagues looked at how salt precipitated on flat, disc-shaped membranes to better understand crust formation. Solar distillation experiments showed that the multilayered silica–carbon–silica structure of their membranes efficiently evaporated pure water, but not brine. After a few hours, salt crystals appeared at the disk edges, eventually spreading to cover most of the membrane, with the exception of the very center.

"When we analyzed these images, we focused on the central areas that



are always uncovered even with different salt concentrations," says Wang. "They helped us realize that <u>structural design</u> may be the key to higher performance, rather than material synthesis."

Using a simple model, the researchers deduced that the inner diameter of the crust ring could be predicted from the brine's salt concentration. They hypothesized that by folding the flat membrane into a new shape—a 3-D cup with a circular base the exact size of the salt-free zone—they could induce salts to precipitate on the cup's side above the liquid interface.

Tests with the 3-D cup produced remarkable results: with optimized height/base ratios, the device recovered nearly 100 percent of <u>brine</u> as solid salt and could run for days without maintenance.

"The insights from this work are quite valuable as we move forward and assess how 3-D solar evaporator performs with real desalination brines," says Wang.

More information: Yusuf Shi et al. Solar Evaporator with Controlled Salt Precipitation for Zero Liquid Discharge Desalination, *Environmental Science & Technology* (2018). DOI: 10.1021/acs.est.8b03300

Provided by King Abdullah University of Science and Technology

Citation: Solar-powered cup brews better brine treatments (2019, April 10) retrieved 25 April 2024 from <u>https://phys.org/news/2019-04-solar-powered-cup-brews-brine-treatments.html</u>

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