

# High-tech material in a salt crust

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Dr. Jesus Gonzalez-Julian (right) and Apurv Dash (left). Credit: Forschungszentrum Jülich / Regine Panknin

MAX phases are viewed as promising materials for the future, for example, in the power, aerospace and medical implants industries. A



new method developed by scientists from Forschungszentrum Jülich now makes it possible to produce this material class on an industrial scale for the first time. They report that a crust of salt protects the raw material from oxidation at a production temperature of more than 1,000 degrees Celsius—and can then simply be washed off with water. The method, which was recently published in the journal *Nature Materials*, can also be applied to other high-performance materials.

MAX phases unite the positive properties of ceramics and metals. They are heat-resistant and lightweight like ceramics, yet less brittle, and can be plastically deformed like metals. Furthermore, they are the material basis of MXenes, a largely unexplored class of compound that are similar to the graphene and have extraordinary electronic properties.

"In the past, there was no suitable <u>method</u> for producing MAX phases in powder form, which would be advantageous for further industrial processing. This is why MAX phases have not played any practical role in industrial applications so far," explains Junior Professor Dr. Jesus Gonzalez-Julian, young investigators group leader at Forschungszentrum Jülich.

## The salt strategy

MAX phases are produced at temperatures higher than 1,000 degrees Celsius. At such high temperatures, the <u>materials</u> would normally react with atmospheric oxygen and oxidize, which is why they are usually produced in a vacuum or in a protective atmosphere of argon. The Jülich method is astonishingly simple by comparison: the researchers encapsulate the raw material with a <u>salt</u>, potassium bromide, which melts during the production process. A vacuum or argon atmosphere for additional protection is no longer needed.





Schematic representation of the process. Credit: Forschungszentrum Jülich / Apurv Dash

"A bath of molten salt thus protects the material and prevents it from coming in contact with <u>atmospheric oxygen</u>," explains Apurv Dash, lead author of the study published in *Nature Materials* and doctoral researcher at Forschungszentrum Jülich.

At the same time, the salt acts as a separating agent. The components no longer bond together to form a compact solid, and allow the direct production of fine-grained powders. This is important because it avoids an additional long, energy-intensive milling process. As a positive side effect, the salt bath also reduces the synthesis temperature necessary to form the desired compound, which will additionally cut energy and production costs.

## With just salt and water

Methods using molten salt have been used for the powder production of non-oxide ceramics for some time. However, they require a protective argon atmosphere instead of atmospheric air, which increases both the complexity and production costs.





Solidified salt crust after cooling. Credit: Forschungszentrum Jülich / Tobias Schlößer

"Potassium bromide, the salt we use, is special because when pressurized, it becomes completely impermeable at room temperature. "We have now demonstrated that it is sufficient to encapsulate the <u>raw</u> <u>materials</u> tightly enough in a salt pellet to prevent contact with oxygen—even before the melting point of the salt is reached at 735 degrees Celsius. A protective atmosphere is thus no longer necessary," explains Apurv Dash.

As with many scientific discoveries, a little bit of luck factored into the development of the method—vacuum furnaces are scarce because they are so expensive and they take a lot of effort to clean. To produce his powder, the Jülich doctoral researcher therefore resorted to testing a



normal air furnace successfully.

The new method is not limited to this material. The researchers have already produced a multitude of MAX phases and other <u>high-</u> <u>performance materials</u>, such as titanium alloys for bioimplants and aircraft engineering. Next, the scientists are planning to investigate industrial processes with which these powders can be processed further.

**More information:** Apurv Dash et al, Molten salt shielded synthesis of oxidation prone materials in air, *Nature Materials* (2019). DOI: 10.1038/s41563-019-0328-1

### Provided by Forschungszentrum Juelich

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