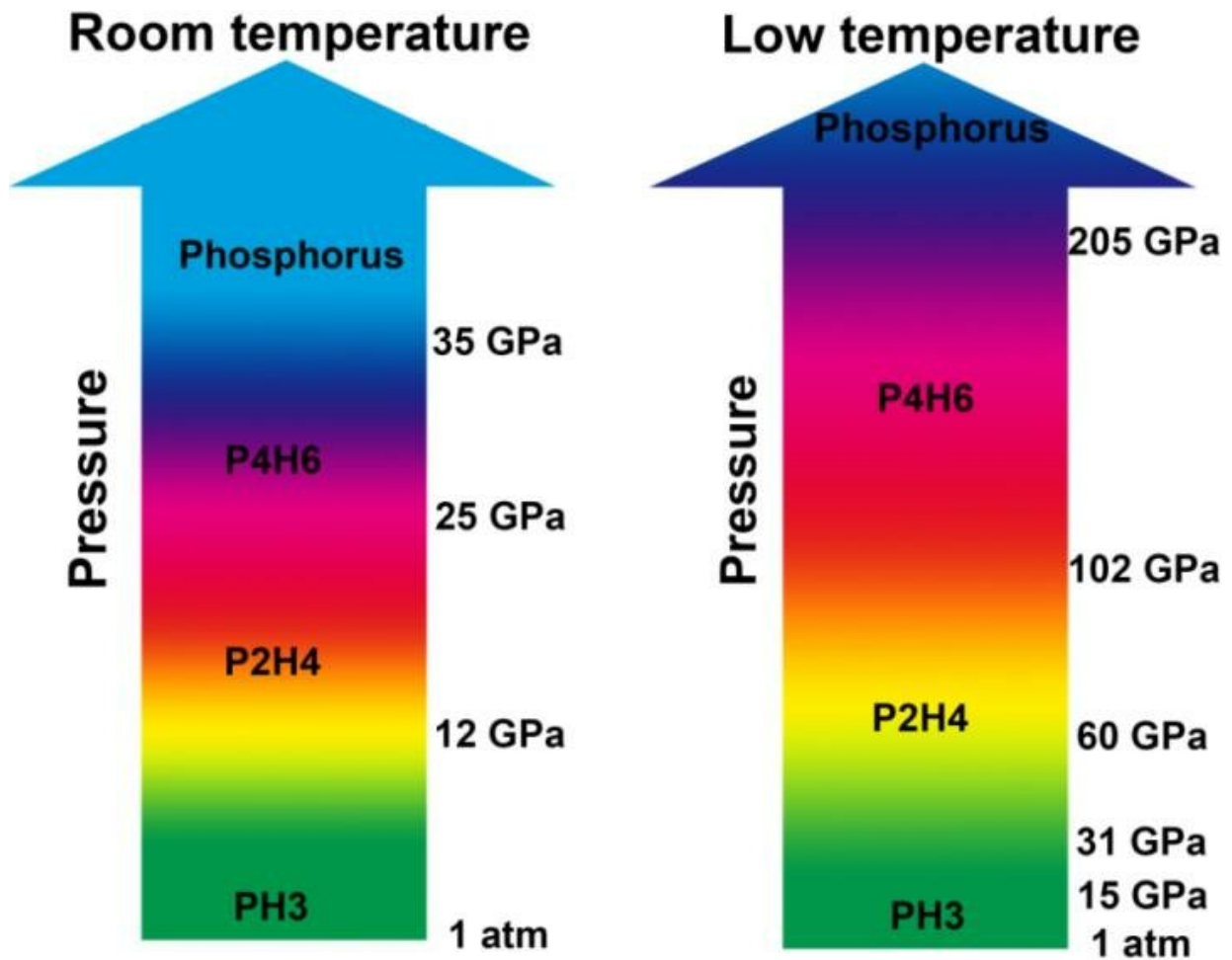


Uncovering the superconducting phosphine: P₂H₄ and P₄H₆

March 19 2019



The high pressure phase diagrams of PH₃ at room temperature and low temperature. Credit: ©Science China Press

High-Tc superconductors have become a hot topic in physics since superconducting mercury was first reported more than a century ago. Dense hydrogen was predicted to metalize and become a superconductor at high pressure and room temperature. However, no widely accepted experimental work has been reported yet. In 2004, Ashcroft predicted hydrogen-dominant hydrides could become a high-Tc superconductor at high pressure, due to chemical precompression. Later, Drozdov et al. observed the superconductive transition of H₂S at 203 K and 155 GPa, which broke the highest T_c record. Very recently, LaH₆ was reported to show superconducting behavior at ~260K. Motivated by these efforts, extensive investigations on hydrides system have been reported.

PH₃, a typical hydrogen-rich hydride, has attracted a great deal of research interest because of its superconducting transition at high pressure. However, structural information was not provided, and the origin of the superconducting transition remains puzzling. Although a series of theoretical works suggest possible structures, the PH₃ phase under compression has remained unknown and no relevant experimental studies have been reported.

In a recent research article published in *National Science Review*, a collaborative of scientists has presented their results on the studies of stoichiometric evolutions of PH₃ under high pressure. They found that PH₃ is stable below 11.7 GPa and then starts to dehydrogenate through two dimerization processes at [room temperature](#) and pressures up to 25 GPa. Two resulting phosphorus hydrides, P₂H₄ and P₄H₆, were verified experimentally and can be recovered to ambient pressure. Under further compression above 35 GPa, the P₄H₆ directly decomposed into elemental phosphorus. Low temperature can greatly hinder polymerization/decomposition under high [pressure](#), and retain P₄H₆ up to at least 205 GPa. "Our findings suggested that P₄H₆ might be responsible for superconductivity at high pressures," said Dr. Lin Wang, the corresponding author of the article.

To determine the possible [structure](#) of P₄H₆ at [high pressure](#), structural searches were performed. Theoretical calculations revealed two stable structures with space group Cmcm (above 182 GPa) and C2/m (above 182 GPa). Phonon dispersions calculations of the two structures do not give any imaginary frequencies. Therefore, this verifies their dynamic stabilities. The superconducting T_c of the C2/m structure at 200 GPa was estimated to be 67 K. "All of these findings confirmed P₄H₆ might be the corresponding superconductor, which is helpful for shedding light on the superconducting mechanism." Dr. Wang added.

More information: Ye Yuan et al, Stoichiometric evolutions of PH₃ under high pressure: implication for high T_c superconducting hydrides, *National Science Review* (2019). [DOI: 10.1093/nsr/nwz010](https://doi.org/10.1093/nsr/nwz010)

Provided by Science China Press

Citation: Uncovering the superconducting phosphine: P₂H₄ and P₄H₆ (2019, March 19)
retrieved 26 April 2024 from
<https://phys.org/news/2019-03-uncovering-superconducting-phosphine-p2h4-p4h6.html>

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