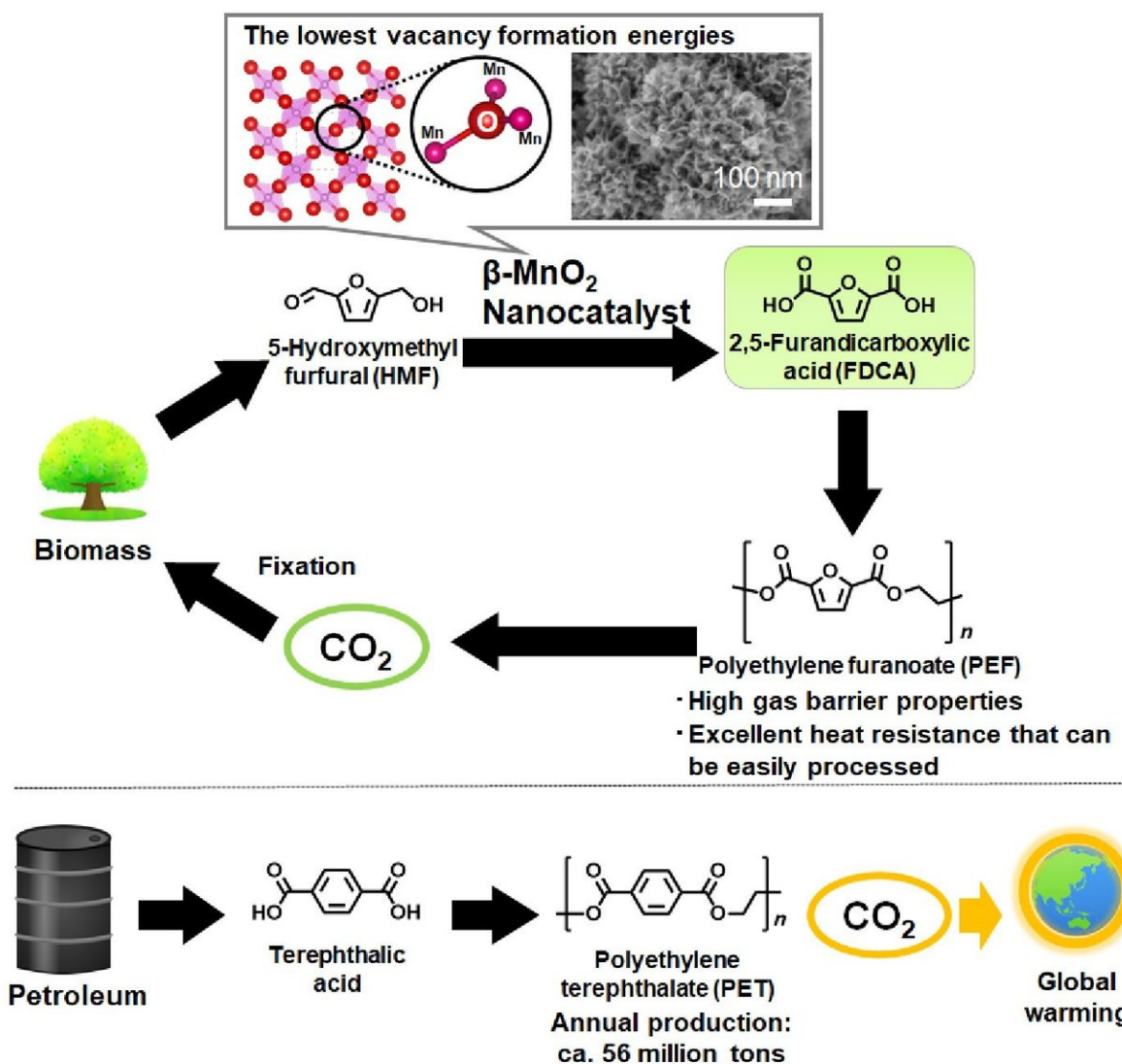


# Green catalysts with Earth-abundant metals accelerate production of bio-based plastic

January 8 2019



Replacing fossil based PET, known as raw material of soft drink bottles, with bio-based PEF largely contributes reduction of CO<sub>2</sub> emissions. Credit: Keigo

Kamata of Tokyo Institute of Technology

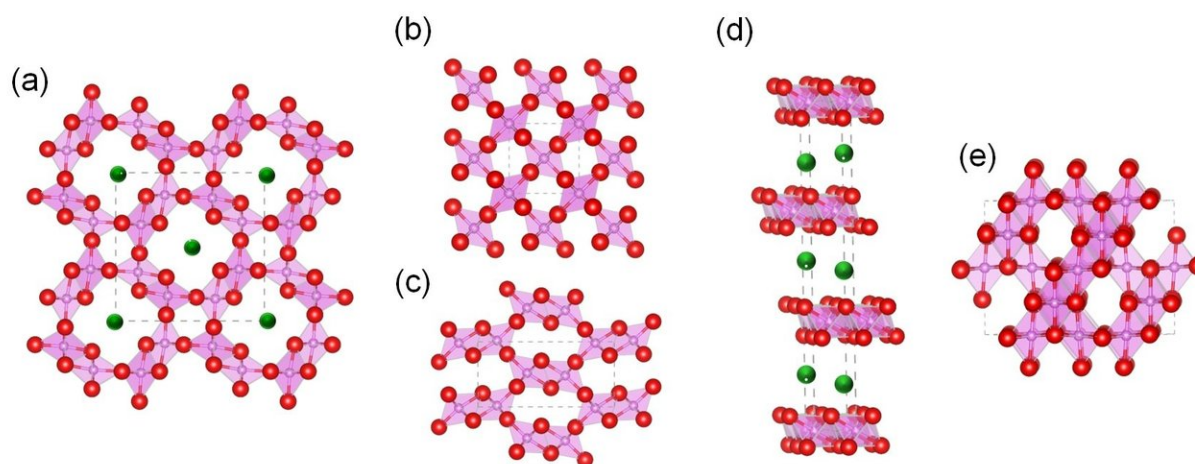
Scientists at Tokyo Institute of Technology (Tokyo Tech) have developed and analyzed a novel catalyst for the oxidation of 5-hydroxymethyl furfural, which is crucial for generating new raw materials that replace the classic non-renewable ones used for making many plastics.

It should be no surprise to most readers that finding an alternative to non-[renewable natural resources](#) is a key topic in current research. Some of the raw materials required for manufacturing many of today's plastics involve non-renewable fossil resources, coal and natural gas, and a lot of effort has been devoted to finding sustainable alternatives.

2,5-Furandicarboxylic acid (FDCA) is an attractive raw material that can be used to create polyethylene furanoate, which is a bio-polyester with many applications.

One way of making FDCA is through the [oxidation](#) of 5-hydroxymethyl furfural (HMF), a compound that can be synthesized from cellulose. However, the necessary oxidation reactions require the presence of a catalyst, which helps in the intermediate steps of the reaction so that the final product can be achieved.

Many of the catalysts studied for use in the oxidation of HMF involve precious metals; this is clearly a drawback because these metals are not widely available. Other researchers have found out that manganese oxides combined with certain metals (such as iron and copper) can be used as catalysts. Although this is a step in the right direction, an even greater finding has been reported by a team of scientists from Tokyo Tech: manganese dioxide ( $\text{MnO}_2$ ) can be used by itself as an effective [catalyst](#) if the crystals made with it have the appropriate structure.



Structures of (a)  $\alpha$ - $\text{MnO}_2$ , (b)  $\beta$ - $\text{MnO}_2$ , (c)  $\gamma$ - $\text{MnO}_2$ , (d)  $\delta$ - $\text{MnO}_2$ , and (e)  $\lambda$ - $\text{MnO}_2$ . Pink, green, and red spheres represent Mn, K, and O atoms, respectively. Of these,  $\beta$ - $\text{MnO}_2$  is the most promising one as a catalyst for oxidation reactions because of the disposition and characteristics of its oxygen atoms. Credit: Keigo Kamata of Tokyo Institute of Technology

The team, which includes Associate Professor Keigo Kamata and Professor Michikazu Hara, worked to determine which  $\text{MnO}_2$  crystal structure would have the best catalytic activity for making FDCA and why. They inferred through computational analyses and the available theory that the structure of the crystals was crucial because of the steps involved in the oxidation of HMF. First,  $\text{MnO}_2$  transfers a certain amount of oxygen atoms to the substrate (HMF or other by-products) and becomes  $\text{MnO}_{2-\delta}$ . Then, because the reaction is carried out under an oxygen atmosphere,  $\text{MnO}_{2-\delta}$  quickly oxidizes and becomes  $\text{MnO}_2$  again. The energy required for this process is related to the energy required for the formation of oxygen vacancies, which varies greatly with the [crystal structure](#). In fact, the team calculated that active oxygen sites had a lower (and thus better) vacancy formation energy.

To verify this, they synthesized various types of  $\text{MnO}_2$  crystals, as shown in Figure, and then compared their performance through numerous analyses. Of these crystals,  $\beta\text{-MnO}_2$  was the most promising because of its active planar oxygen sites. Not only was its vacancy formation energy lower than that of other structures, but the material itself was proven to be very stable even after being used for oxidation reactions on HMF.

The team did not stop there, though, as they proposed a new synthesis method to yield highly pure  $\beta\text{-MnO}_2$  with a [large surface area](#) in order to improve the FDCA yield and accelerate the oxidation process even further. "The synthesis of high-surface-area  $\beta\text{-MnO}_2$  is a promising strategy for the highly efficient oxidation of HMF with  $\text{MnO}_2$  catalysts," states Kamata.

With the methodological approach taken by the team, the future development of  $\text{MnO}_2$  catalysts has been kick-started. "Further functionalization of  $\beta\text{-MnO}_2$  will open up a new avenue for the development of highly efficient catalysts for the oxidation of various biomass-derived compounds," concludes Hara. Researches such as this one ensure that renewable [raw materials](#) will be available to mankind to avoid all types of shortage crises.

**More information:** Eri Hayashi et al, Effect of  $\text{MnO}_2$  Crystal Structure on Aerobic Oxidation of 5-Hydroxymethylfurfural to 2,5-Furandicarboxylic Acid, *Journal of the American Chemical Society* (2019). [DOI: 10.1021/jacs.8b09917](https://doi.org/10.1021/jacs.8b09917)

Provided by Tokyo Institute of Technology

Citation: Green catalysts with Earth-abundant metals accelerate production of bio-based plastic (2019, January 8) retrieved 1 May 2024 from <https://phys.org/news/2019-01-green-catalysts->

[earth-abundant-metals-production.html](http://earth-abundant-metals-production.html)

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.