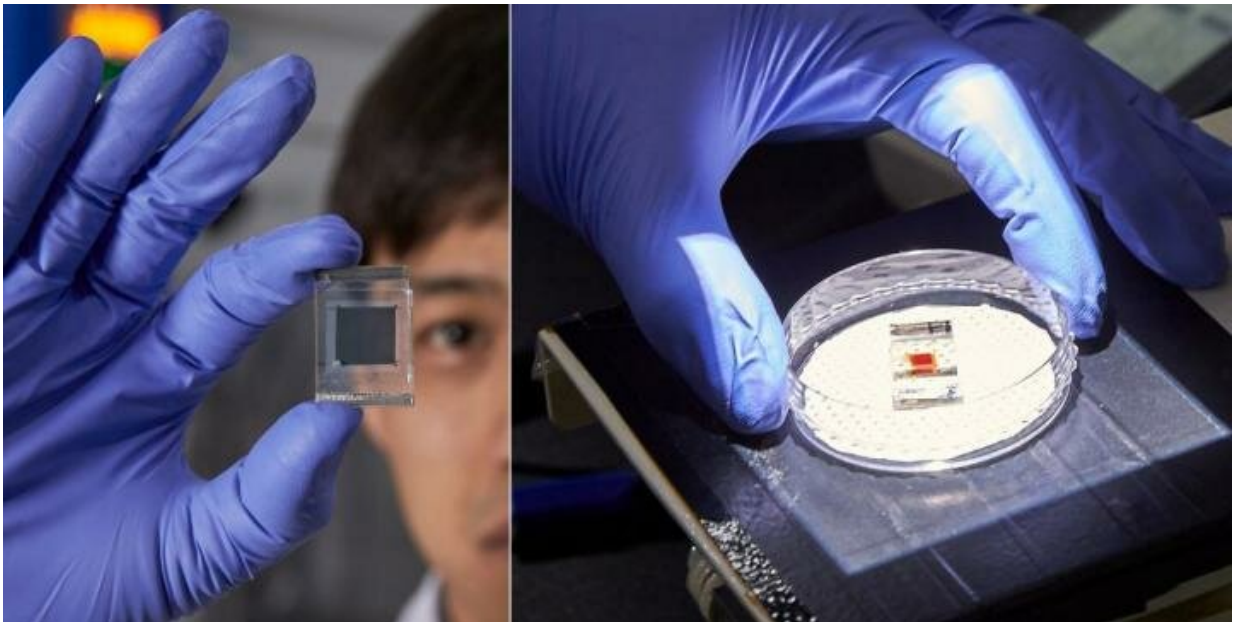


# New class of solar cells, using lead-free perovskite materials

January 21 2019, by Joo Hyeon Heo

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Lead-free Perovskite flim (left) and dye-sensitized organic solar cells (right).  
Credit: UNIST

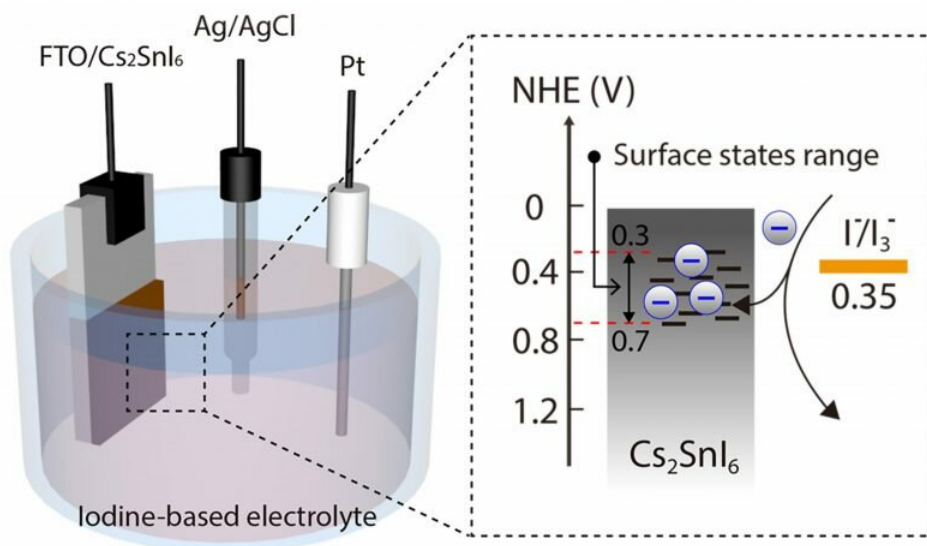
Lead-based perovskites are promising materials for low-cost and high-efficiency solar cells. However, the intrinsic instability and the toxicity of lead (Pb) have raised serious concerns of the viability of Pb-based perovskites, hindering large-scale commercialization of solar cells and similar devices based on these materials. As an alternative solution, Pb-free perovskites were recently proposed to counter the toxicity of lead-

based perovskites, yet it is of little use due to lower efficiencies.

A recent study, led by Professor Tae-Hyuk Kwon in the School of Natural Science at UNIST, represents a major step toward the development of a new generation of solar cells using lead-free perovskites. With its promising electronic properties, the new [perovskite](#) material has been demonstrated to function as a charge regenerator with dye-sensitized solar cells, thus enhancing both the overall efficiency and stability. Published in the November 2018 issue of *Advanced Materials*, the findings will open new possibilities for the application of lead-free perovskites in solar cells.

Among the various alternatives to lead, the research team used the vacancy-ordered double perovskite ( $\text{Cs}_2\text{SnI}_6$ ). Despite their promising outlook, the [surface](#) states of  $\text{Cs}_2\text{SnI}_6$  and their function remain largely unclear. Thus, a comprehensive study is necessary to clarify these features of  $\text{Cs}_2\text{SnI}_6$  for the future design of  $\text{Cs}_2\text{SnI}_6$ -based devices.

Through this work, the team examined the charge transfer mechanism of  $\text{Cs}_2\text{SnI}_6$  with the aim of clarifying the function of its surface state. For this purpose, a three-electrode system was developed to observe charge transfer through the surface state of  $\text{Cs}_2\text{SnI}_6$ . Cyclic voltammetry and Mott-Schottky analyses were also used to probe the surface state of  $\text{Cs}_2\text{SnI}_6$ , whose potential is related to its bandgap.



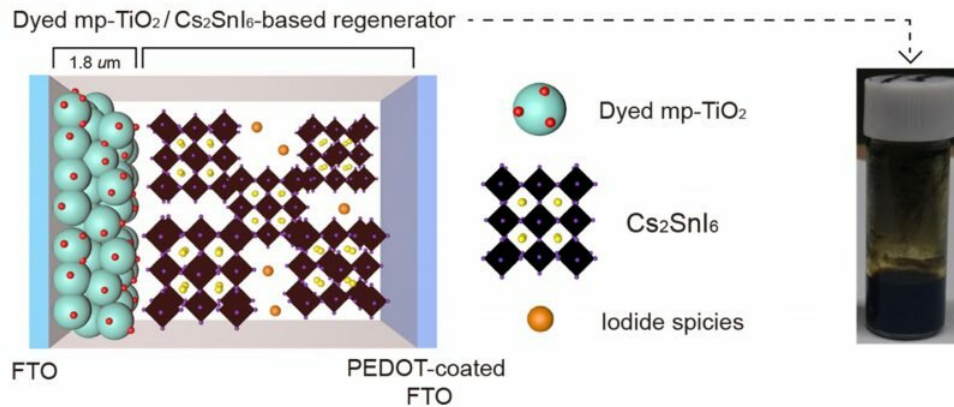
Above is the 3-electrode system for the observation of charge transfer through the surface state of  $\text{Cs}_2\text{SnI}_6$ . Credit: Ulsan National Institute of Science and Technology

Their analysis demonstrated that the surface state of  $\text{Cs}_2\text{SnI}_6$  is highly redox active and can be effectively charged/discharged in the presence of iodide redox mediators. Besides, the preparation of a charge regenerator system based on  $\text{Cs}_2\text{SnI}_6$  confirmed that charge transfer occurred through the surface state of  $\text{Cs}_2\text{SnI}_6$ .

"In case of  $\text{Cs}_2\text{SnI}_6$ , charge transfer occurred through the surface state of  $\text{Cs}_2\text{SnI}_6$ ," says HyeonOh Shin in the Combined MS./Ph.D in Chemistry at UNIST. "This will aid in the design of future electronic and energy devices, using Pb-free perovskites."

Based on this strategy, the research team engineered hybrid solar cells, using a  $\text{Cs}_2\text{SnI}_6$ -based charge regenerator for organic dye-sensitized solar

cells (DSSCs). Such solar [cells](#) generate [electric current](#) in the process where the oxidized organic dye returns to its original state.



Credit: Ulsan National Institute of Science and Technology

"Due to a high volume of electrical charges in organic dyes that show high connectivity with the surface state of Cs<sub>2</sub>SnI<sub>6</sub>, more electric current were generated," says Byung-Man Kim in the Department of Chemistry at UNIST, another lead author of this study. "Consequently, Cs<sub>2</sub>SnI<sub>6</sub> shows efficient charge transfer with a thermodynamically favorable charge acceptor level, achieving a 79% enhancement in the photocurrent density compared with that of a conventional liquid electrolyte."

This study has attracted considerable attention among researchers, as it examined the charge transfer mechanism of Cs<sub>2</sub>SnI<sub>6</sub> with the aim of clarifying the function of its surface state. Their results suggest that the surface state of Cs<sub>2</sub>SnI<sub>6</sub> is the main charge [transfer](#) pathway in the presence of a redox mediator and should be considered in future designs

of  $\text{Cs}_2\text{SnI}_6$ -based devices.

**More information:** HyeonOh Shin et. al., "Surface State-Mediated Charge Transfer of  $\text{Cs}_2\text{SnI}_6$  and Its Application in Dye-Sensitized Solar Cells," *Advanced Energy Materials*, (2018).

Provided by Ulsan National Institute of Science and Technology

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