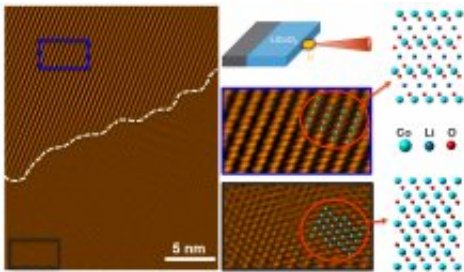


# Scientists study impact of lithium dendrites on cathode materials

December 5 2018

---



Researchers imaged the structural and chemical evolutions of  $\text{LiCoO}_2$  cathodes upon Li dendrite contact at an atomic scale. A spontaneous, irreversible conversion reaction leads to the formation of Co metal and  $\text{Li}_2\text{O}$ , with  $\text{CoO}$  as a metastable reaction intermediate. Credit: Environmental Molecular Sciences Laboratory

Rechargeable lithium (Li) batteries are the most widely used energy storage devices found in consumer electronics and electric vehicles today. There are challenges, though, to optimizing their capacity, recyclability, and stability. During a Li battery's charge-discharge cycles, highly localized Li dendrites – strands of Li that grow inside batteries – can form and hurt the battery's performance.

"Lithium dendrites are fine strands, like whiskers, that can come in contact with [cathode materials](#) and cause a chain of irreversible, spontaneous chemical reactions," said Yingge Du, a researcher from Pacific Northwest National Laboratory (PNNL) who recently led a study

to visualize this phenomena. "It can lead to performance decay, short circuiting, and safety hazards," he added.

Du and his team sought to pinpoint the exact failure mechanism. To do so, they needed to precisely control how and when the Li dendrites came into contact with the [cathode](#) materials. The team used in situ [transmission electron microscopy](#) (TEM) to directly image the structural and chemical evolutions of lithium cobalt oxide ( $\text{LiCoO}_2$ ) thin film cathodes on Li [dendrite](#) contact. " $\text{LiCoO}_2$  is still one of the most widely used cathode materials," said Du, "and making it into well-defined single-crystalline form by pulsed [laser deposition](#) can unleash the power of advanced electron microscopy." Using this approach, researchers discovered an unexpected Li propagation pathway and detailed [reaction](#) steps that lead to the cathode failure.

Their study, "Direct Visualization of Li Dendrite Effect on  $\text{LiCoO}_2$  Cathode by In Situ TEM," was recently published in *Small* and featured on its cover.

To meet the increasing demands for large-scale [energy storage devices](#), there is an urgent need for researchers to develop safer, rechargeable batteries with higher energy output. Li metal has been regarded as an ideal, high-capacity anode material. However, its use is greatly hampered by the formation of Li dendrites that can penetrate the separator – a permeable membrane placed between a battery's anode and cathode. Once Li dendrites are in direct contact with cathode materials, a chain of irreversible, spontaneous chemical reactions can occur, leading to performance decay, short circuiting, and safety hazards.

While significant efforts have been devoted to detect, understand, and prevent the Li dendrite formation within the electrolytes, little is known regarding the detailed reaction pathways involving Li metal and cathode materials. Bridging this knowledge gap may provide detection and design

principles that are critical to future energy storage solutions.

In this study, Du and his team of researchers at PNNL studied Li dendrites and their effect on cathode materials. They sought to understand the detailed reaction pathways, which could lead to the development of better Li batteries.

Using pulsed-laser deposition, Du and his team fabricated well-defined, epitaxial  $\text{LiCoO}_2$  thin films with controlled crystallographic orientations to serve as the model cathode materials. A Li metal tip was used to mimic the Li dendrite inside a TEM to study its reaction with prepared  $\text{LiCoO}_2$  samples.

Advanced microscopy and spectroscopy techniques – including scanning transmission electron spectroscopy, nanobeam diffraction, and electron-energy loss spectroscopy – allowed such reactions to be investigated with high spatial and temporal resolution. In combination with density functional theory calculations, researchers elucidated the reaction steps, intermediates, and final products at an unprecedented level. An unexpected Li diffusion direction perpendicular to the Li-containing planes was found, which tore the  $\text{LiCoO}_2$  crystal apart, generating large amount of grain boundaries and antiphase boundaries. While Co metal and  $\text{Li}_2\text{O}$  were found to be the final products of the full conversion reaction,  $\text{CoO}$  was identified as a metastable intermediate at the reaction front as a result of facile phase transition from  $\text{LiCoO}_2$ .

"The reaction steps and intermediates revealed provide a clear failure mechanism for the  $\text{LiCoO}_2$  cathodes caused by Li dendrites, and may also offer insights into the over-discharge processes in cathodes," remarked Du.

Continuing this work, Du's team intends to fabricate all-solid-state batteries by multi-step deposition processes using pulsed-laser deposition

to better understand the ion transport processes across the well-defined interfaces.

**More information:** Zhenzhong Yang et al. Direct Visualization of Li Dendrite Effect on  $\text{LiCoO}_2$  Cathode by In Situ TEM, *Small* (2018). [DOI: 10.1002/sml.201803108](https://doi.org/10.1002/sml.201803108)

Provided by Environmental Molecular Sciences Laboratory

Citation: Scientists study impact of lithium dendrites on cathode materials (2018, December 5) retrieved 24 May 2024 from <https://phys.org/news/2018-12-scientists-impact-lithium-dendrites-cathode.html>

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.