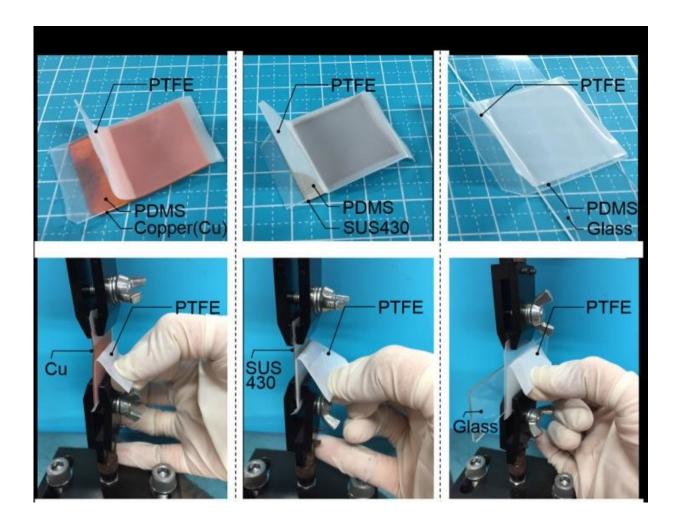


## Give it the plasma treatment: Strong adhesion without adhesives

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Photograph of three-layer assemblies of (a) PTFE/PDMS/Cu, (b) PTFE/PDMS/SUS430, and (c) PTFE/PDMS/glass. When the PTFE or PDMS sheet was jerked and shaken, no peeling occurred at the interfaces of PTFE/PDMS, PDMS/Cu, PDMS/SUS430, and PDMS/glass. This shows that PJtreated PDMS could be used as an alternative to strong adhesives to adhere



fluoropolymers to other types of materials. Credit: Osaka University

Polymers containing plastics are essential in modern life. Being lightweight, strong and unreactive, a vast range of technologies depend on them. However, most polymers do not adhere naturally to other materials, so they need adhesives or corrosive chemical treatments to enable attachment. This is a problem in areas like food and medicine, where contamination must be avoided at all costs.

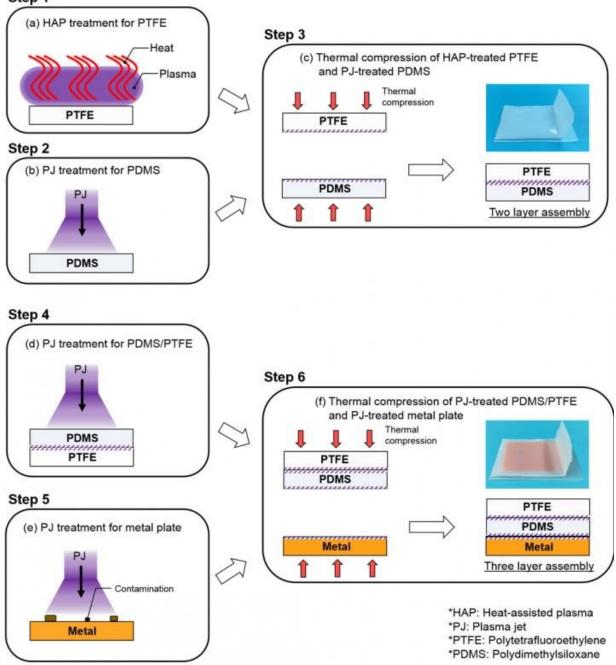
A clean way to make industrial polymers adhesive is urgently needed. Now, a team at Osaka University has achieved just that. They have developed a suite of plasma treatments to allow vulcanized rubber and the plastic PTFE (polytetrafluoroethylene) to adhere to one another or to other materials. The method activates the polymers' surface chemistry, as described in a study in *Scientific Reports*.

"If you spray PTFE with a plasma of helium at 200 degrees, it can adhere to unvulcanized rubber—this is a technique we developed earlier in our lab," says study lead author Yuji Ohkubo. "But vulcanized rubber presents a greater challenge. In our latest study, we customized a new plasma treatment for vulcanized <u>silicone rubber</u>, making it adhere strongly to PTFE for the first time."

The silicone in question was PDMS (polydimethylsiloxane), a wellknown resin. While the key breakthrough in PTFE adhesion was the heatassisted plasma treatment, the trick with PDMS is to bombard the surface with a <u>plasma jet</u>, by forcing nitrogen/air plasma through a small hole. The jet breaks the silicon-carbon bonds on the surface and converts them to silanol (Si-OH).



Step 1



Preparation procedure for a two-layer assembly such as PTFE/PDMS (steps 1-3) and a three-layer assembly such as PTFE/PDMS/Cu (steps 1-6). PFA/PDMS, PFA/PDMS/glass, PTFE/PDMS/glass, and PTFE/PDMS/SUS430 assemblies were prepared in the same way. Credit: Osaka University



Being more reactive than the original silicone surface, these silanol groups can bond with PTFE. Under <u>high pressure</u>, <u>hydrogen bonds</u> form between silanol and the oxygen-containing functional groups on the treated PTFE. Strong covalent bonds (C-O-Si, where C comes from PTFE and Si from silicone) further stitch the two polymers together, even with no adhesive.

Uniting the two materials allows each to enjoy the benefits of the other—the chemical resistance, dirt-repellent and slide-ability of PTFE, and the elasticity of silicone. Opaque PTFE can also be replaced by PFA (perfluoroalkoxy alkane) if transparency is needed. And that's not all—when the reverse side of the PDMS is also <u>plasma</u>-jetted, it can bond to copper and even glass. Like an extremely strong double-sided tape, this three-layer sandwich allows the fluoropolymers to adhere cleanly to other useful materials.

"PDMS is widely used in medicine, for example in microfluidic chips," explains co-author Katsuyoshi Endo. "There could be huge benefits in making both PTFE and PDMS more versatile for medical and food technologies through adhesive-free adhesion. Combined with the lack of any need for <u>volatile chemicals</u>, we hope our method will broaden the horizons for polymers in high technology."

**More information:** Yuji Ohkubo et al, Adhesive-free adhesion between heat-assisted plasma-treated fluoropolymers (PTFE, PFA) and plasma-jet-treated polydimethylsiloxane (PDMS) and its application, *Scientific Reports* (2018). DOI: 10.1038/s41598-018-36469-y

Provided by Osaka University

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