

# Making it easier to capture carbon dioxide in the cement industry

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Cement is one of the world's commonest building materials, and its manufacture accounts for seven per cent of our greenhouse gas emissions. We have to do something about this. This photo was taken in Shanghai. Credit: Thinkstock

Cement manufacture accounts for as much as seven percent of global greenhouse gas emissions. A new hybrid technology makes it easier and less expensive to capture and purify  $\text{CO}_2$  produced by the industry. And the technology can be retrofitted to existing plant.

Researchers at SINTEF have now upgraded the method used to capture greenhouse gases generated during the [cement](#) manufacturing process. The hope is that this technology can be utilised by cement factories and in other [industrial processes](#) in coastal areas and along European rivers, because the method is based on the liquefaction of CO<sub>2</sub>, allowing it to be transported by ship. It can also be retrofitted to existing plant.

Normally, the flue gases emitted from a cement factory contain about 20 percent CO<sub>2</sub>. In order to transport and/or store the CO<sub>2</sub> from these gases, they must first be scrubbed. The current minimum requirement is approx. 95 percent purity. However, this process requires large amounts of heat and is thus energy-demanding.

The team of researchers is now proposing that the sector should utilise a hybrid technology that makes CO<sub>2</sub> capture easier, more energy efficient, and better suited to CO<sub>2</sub> transport by ship.

### **More than 99 percent pure**

"We propose a promising approach using a membrane filter combined with an in-house developed system involving forced concentration of the CO<sub>2</sub> by liquefaction," explains SINTEF researcher David Berstad. "We achieve this by cooling it down under pressure," he says.

The membrane alone concentrates the CO<sub>2</sub> gases to approximately 70 percent purity. It is currently standard to employ an additional membrane step to scrub the gas and achieve the necessary purity of 95 percent. However, the research team's method involving CO<sub>2</sub> liquefaction is less energy-demanding, and also results in an even purer gas.

Measurements made by Berstad and his research colleague Stian Trædal during experiments with the new system carried out on a laboratory rig

in Trondheim show levels of purity of at least 99 percent.

"The best results that we have seen are about 99.8 percent, but it is theoretically possible to achieve even higher levels of purity," says Berstad. "The purer the gas the better, because it then requires less capacity to transport and/or store the CO<sub>2</sub> in gas or liquefied form.

### **Electrical energy "replaces" steam**

Another advantage of this process is that, unlike other systems, it uses electrical energy to cool and compress the gas instead of steam to regenerate solvents (chemicals that bind to the CO<sub>2</sub> – ed. note) as is the case using conventional CO<sub>2</sub> capture technologies.

Thus the hybrid CO<sub>2</sub> capture process has no need of steam, which only very few industrial facilities in Norway and Europe have access to without having to construct an additional steam production plant.

### **Ship transport instead of pipelines**

Another important benefit is that the liquefied CO<sub>2</sub> can be transported by ship. This involves making CO<sub>2</sub> liquefaction an integral part of the capture process, preparing the CO<sub>2</sub> for transport by ship, which is the means stipulated in the plans for Norway's full-scale CCS project.

"If CO<sub>2</sub> is not to be transported by ship, but under high pressure in a pipeline, the liquefied CO<sub>2</sub> is pumped to the required pressure at a temperature of minus 50 degrees before later being heated up again," says Berstad.

### **Multiple lab rig experiments**

The CO<sub>2</sub> condensation experiments carried out by the SINTEF

researchers using their laboratory rig were carried out in autumn 2018, resulting in the report "Experimental investigation of CO<sub>2</sub> liquefaction for CO<sub>2</sub> capture from cement plants."

"During the experiments the lab rig behaved as expected, so now we know that we can be confident that it works and can force the limits of purity even further," says Berstad.

In future, the rig will be made more versatile to enable it to be used for other experiments such as the separation of syngas mixtures, which is relevant in a number of different systems being used for emissions-free hydrogen production.

"We have constructed the rig to be as versatile as possible so that it can meet a number of different experimental needs," says Berstad. "As well as carbon capture and storage applications, we want to use it to find out exactly what level of purity is needed for the liquefaction of CO<sub>2</sub>."

Besides, we intend to use it to see how we can efficiently remove and liquefy CO<sub>2</sub> during the production of hydrogen from natural gas," he says.

Provided by SINTEF

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