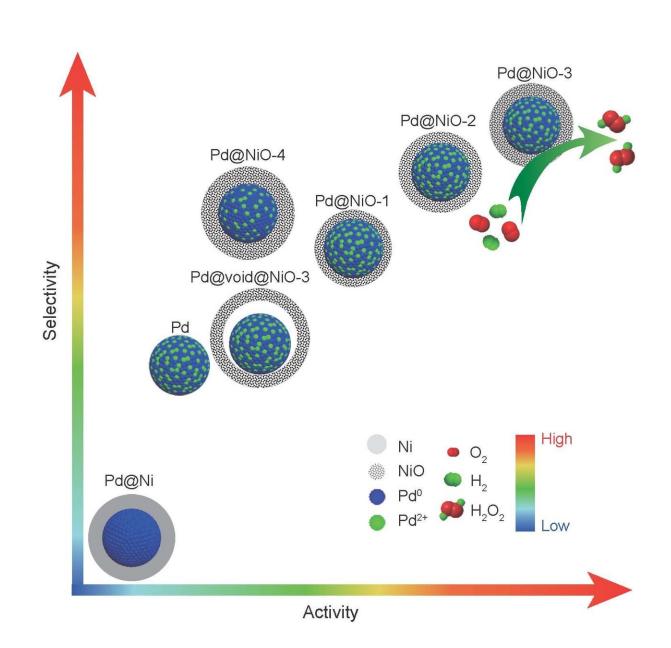


Interfacial engineering core@shell nanoparticles for active and selective direct H2O2 generation

September 20 2018





Schematic illustration showing the activity and selectivity toward H2O2 synthesis of 5 wt% Pd@Ni-3/TiO2, 5 wt% Pd@NiO-x/TiO2 (x = 1, 2, 3, 4), 5 wt% Pd@void@Ni-3/TiO2 and 5 wt% Pd/TiO2. Credit: Science China Press

A class of supported Pd@NiO-x core@shell catalysts have been constructed for direct H_2O_2 generation. The optimized Pd@NiO-3/TiO₂ exhibited high activity, superior selectivity, low degradation activity and excellent stability. The unique, cavity-contained interface structure can suppress the overbinding between Pd-core and (O-O)*, which is effective to prevent H_2O formation and guarantees high selectivity of H_2O_2 . The present work highlights the importance of interface engineering of Pd-based catalysts for direct H_2O_2 synthesis.

Hydrogen peroxide (H_2O_2) is a versatile chemical, widely applied in modern industry. To date, H_2O_2 is industrially manufactured by an indirect process that involves the sequential hydrogenation and oxidation of alkyl anthraquinone, an energy-intensive, multi-step process with high cost. By contrast, the direct synthesis of H_2O_2 from H2 and O2 is expected to be the most efficient way to produce H_2O_2 due to the remarkable advantages of atom economy, low energy consumption and H_2O as its only byproduct.

Currently, the direct synthetic route is mainly achieved by the supported Pd-based catalysts. The major problem associated with that is related to the low selectivity of H_2O_2 . Despite great efforts devoted to constructing Pd-based catalysts, understanding high-performance Pd-based catalysts for direct H_2O_2 generation from either deep characterization or theoretical investigation are still extremely limited.

In a new overview published in the Beijing-based National Science



Review, scientists at the Soochow University present the latest advances in direct H_2O_2 generation. Co-authors Yonggang Feng, Qi Shao, Bolong Huang, Junbo Zhang, and Xiaoqing Huang developed a class of Pd@NiOx nanoparticles with a unique core@shell interface structure, which achieves high activity, selectivity and stability for the direct H_2O_2 synthesis.

These scientists interpreted the mechanism from both electronic and energetic views. "Traditional Pd-based catalysts are very active for the side reactions, such as the decomposition and hydrogenation of H_2O_2 as well as the formation of H2O," they state in an article titled "Surface engineering in the interface of core/shell nanoparticles promotes hydrogen peroxide generation."

"It is considered that the intrinsic surface property of Pd-based catalysts is essential for the selectivity and activity of the direct H_2O_2 synthesis," they add. "This arises because the barrier for O-O bond scission is sensitive to Pd surface structure, the key parameter governing H_2O_2 synthesis and decomposition activity."

The creation of porous NiO shell is beneficial for exposing Pd active sites and thus enhancing the productivity of H_2O_2 . "By tuning the composition of Pd@NiO-x NPs and the reaction condition, the efficiency of H_2O_2 synthesis could be well optimized with 5 wt% Pd@NiO-3/TiO2 exhibiting the highest productivity (89 mol/(kgcath)) and selectivity (91%) to H_2O_2 as well as excellent stability," they state.

"The first principles simulations further revealed the mechanism from both electronic and energetic views," the scientists wrote. "The superiority in selectivity is achieved by a spontaneous bond scission of H-H and charge transfer from O20 to O22- within the cavity of NiO interfacing with Pd surface. (...) The high selectivity and activity make it one of the best catalysts for the direct H_2O_2 synthesis reported to date,"



they add. "The present work reported here highlights the importance of surface and interface engineering of Pd-based catalysts for the direct H_2O_2 synthesis with largely enhanced activity and selectivity."

More information: Yonggang Feng et al, Surface engineering at the interface of core/shell nanoparticles promotes hydrogen peroxide generation, *National Science Review* (2018). DOI: 10.1093/nsr/nwy065

Provided by Science China Press

Citation: Interfacial engineering core@shell nanoparticles for active and selective direct H2O2 generation (2018, September 20) retrieved 2 May 2024 from <u>https://phys.org/news/2018-09-interfacial-coreshell-nanoparticles-h2o2.html</u>

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