

# Chirality switching in biomineral structures

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Jeremy the snail (top) has a counter-clockwise rotating shell, contrary to that of most of his fellow snails (bottom). Credit : Dr Angus Davison, University of Nottingham

Researchers at McGill University have discovered a mechanism by which helical biomineral structures can be synthesized to spiral clockwise or counterclockwise using only either the left-handed or right-handed version of a single acidic amino acid.

Marc McKee, a professor in McGill's Faculty of Dentistry and the Department of Anatomy and Cell Biology, and his team, along with collaborators from Johns Hopkins University, shed light on how certain biological structures—like terrestrial and marine snail shells—can have both clockwise and counterclockwise spirals, not only within the same species, but also within the [shell](#) of an individual organism.

In an article published online in *Science Advances*, Professor McKee and his team constructed chiral helicoidal structures of the biomineral calcium carbonate (a mineral typically found in shells) by the simple addition of specific [amino acids](#).

Remarkably, they could also make the helicoids turn clockwise or counterclockwise by simply using only either the left-handed or right-handed version of the amino acids aspartic acid and glutamic acid—these being abundant in proteins that guide biomineralization processes in many organisms.

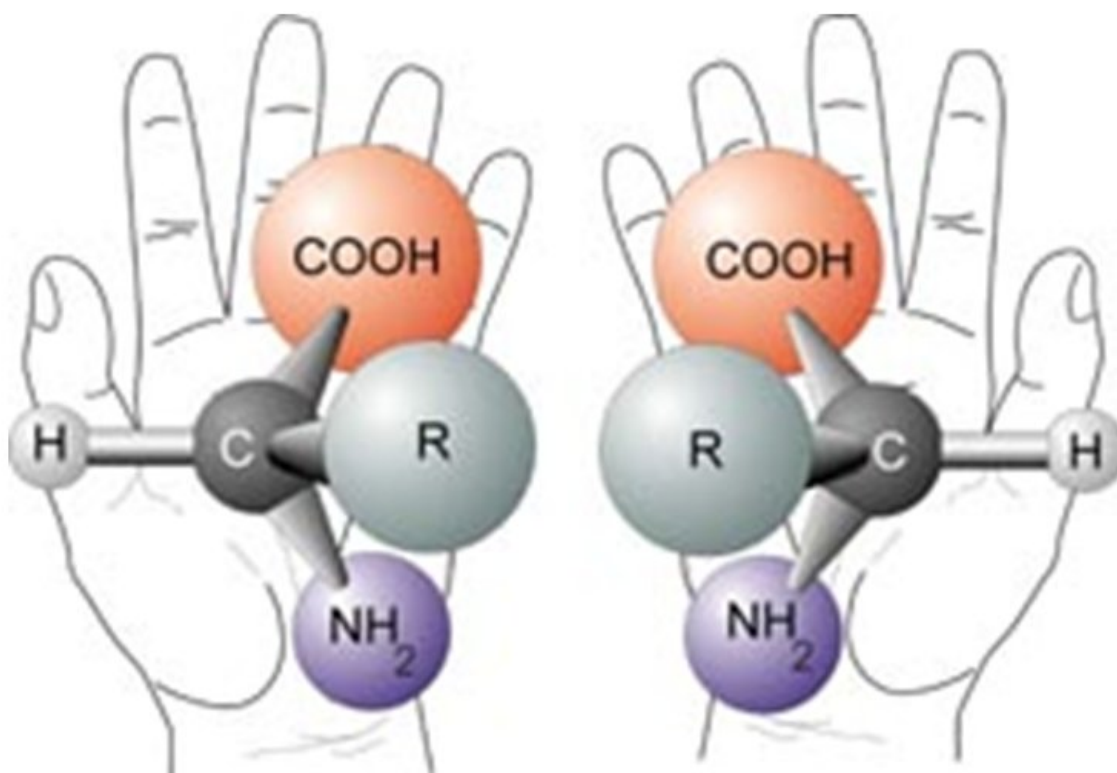
"This work provides key information towards understanding how organisms can have biomineralized structures rotating in opposite directions," says McKee, senior author of the study. "Our findings also give insight into pathologic chiral malformations that might arise in human otoconia, structures in the inner ear whose normal functioning maintains balance by gravity sensing and by detecting linear acceleration. Our work predicts how pathologic chiral malformations might arise in human otoconia and could one day be used to develop therapies for vertigo (loss of balance) based on this understanding".

## What is chirality?

Have you ever heard of Jeremy the snail? He became famous in 2016 because unlike most of his fellow brown garden snails, his shell spirals counter-clockwise. This anomaly made it hard for British researchers

studying Jeremy to find him a mate.

Why is the spiral of Jeremy's shell so important? Because it's one example of a widespread phenomenon found throughout nature: chirality. Objects that are mirror images of each other are considered to be chiral if it's impossible to superimpose them on each other. That is, whichever way you turn them, they will never be the same. Jeremy's shell, with its counterclockwise rotation, cannot be superimposed onto the clockwise-rotating shells of the vast majority of his fellow snails.



Much like our hands are non-superimposable images of themselves, many molecules are also chiral. Amino acids, the building blocks of protein, have chiral forms, that is, where one form is left-handed and the other form is right-handed and will never be the same, whichever way you turn them. Credit: Public domain, via Wikimedia Commons

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Scientists originally hypothesized that biological helical structures—such as snail shells or the twisting tooth of the narwhal—could be explained by the actions of chiral molecules that caused them to [spiral](#) clockwise (left-handed) or counterclockwise (right-handed). In nature, however, where both rotations can sometimes be found, this can't be the case because only left-handed amino acids and biomolecules are generally involved in biological processes.

**More information:** Wenge Jiang et al. Chiral switching in biomineral suprastructures induced by homochiral l-amino acid, *Science Advances* (2018). [DOI: 10.1126/sciadv.aas9819](https://doi.org/10.1126/sciadv.aas9819)

Provided by McGill University

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