

# CO<sub>2</sub> shortage—why can't we just pull carbon dioxide out of the air?

July 4 2018, by Peter Styring And Katy Armstrong

---



Credit: Marcin Jozwiak from Pexels

More people than ever are acutely aware that rising levels of carbon dioxide (CO<sub>2</sub>) in the atmosphere are accelerating climate change and global warming. And yet food manufacturers have been issuing stark

warnings that they've nearly run out of the gas, which is used in many products from [beer to crumpets](#). The obvious question is: why we can't just capture the excess CO<sub>2</sub> from the atmosphere and use that?

It is actually possible to take CO<sub>2</sub> from the atmosphere using a process known as direct air capture. Indeed, there are a number of companies across the world, including [one in Switzerland](#) and [another in Canada](#), that can [already carry out this process](#). In theory, it could turn a problem into a valuable resource, particularly in developing countries with little other natural wealth.

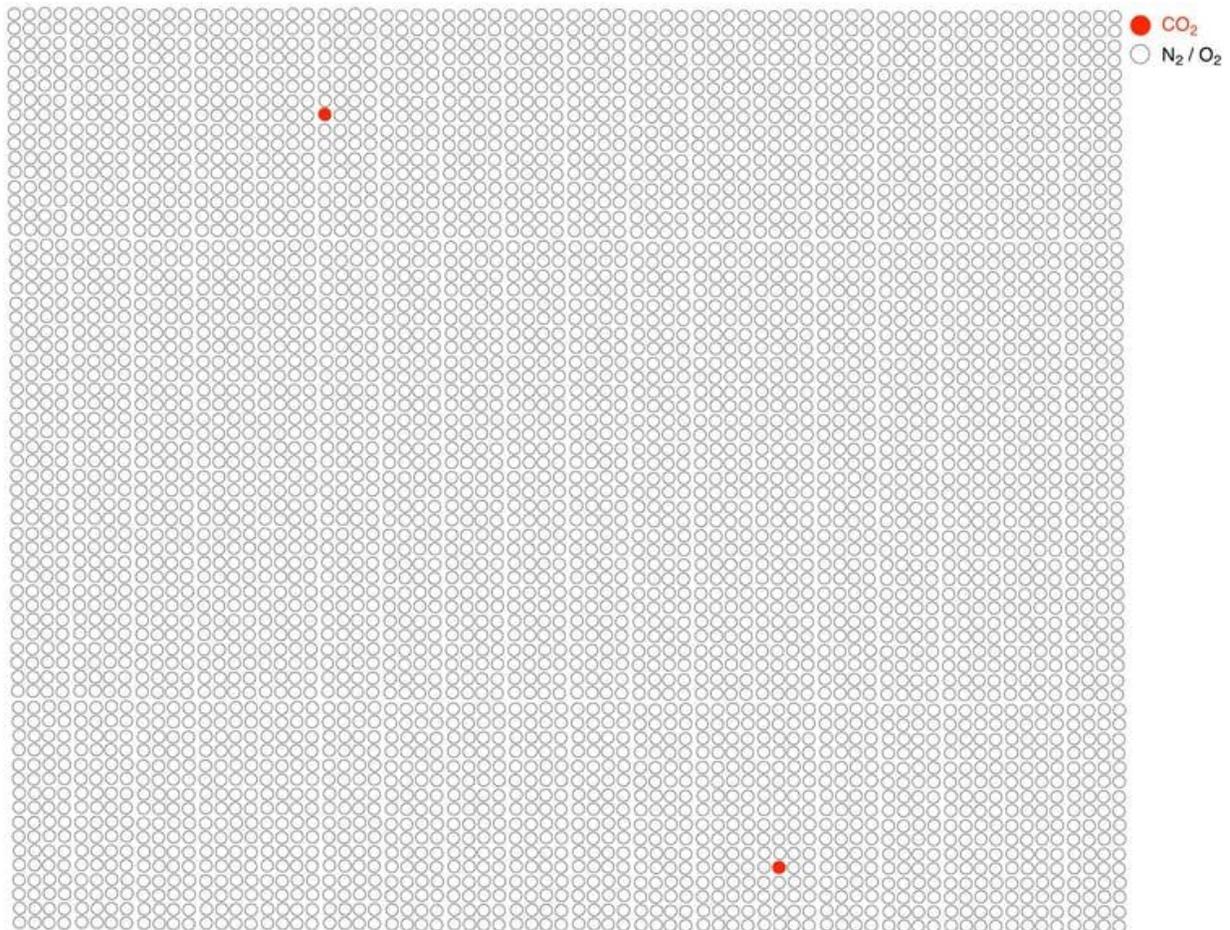
The problem is the cost. While the amount of CO<sub>2</sub> in the air is damaging the climate, relatively speaking there are so few CO<sub>2</sub> molecules in the air that sucking them out is very expensive. But there may be other solutions that could help reduce [carbon emissions](#) and provide a new source of CO<sub>2</sub> for industry.

It's all a matter of concentration and energy consumption. The amount of CO<sub>2</sub> in the air (which is mostly made up of nitrogen and oxygen) is around [400 parts per million](#) or 0.04%. If we were to represent a sample of molecules from the air as a bag of 5,000 balls, just two of them would be CO<sub>2</sub>. Pulling them out of the bag would be very difficult.

We can capture CO<sub>2</sub> using what's known as a sorbent material that either physically interacts or bonds with the gas at a molecular level. To capture a viable amount of CO<sub>2</sub> from the air, we would need to compress huge amounts in order to pass it through the sorbent, something that would require a lot of energy.

The exhaust of power stations is a more concentrated source of CO<sub>2</sub> (and one responsible for so much of our total carbon emissions). The Carbon XPRIZE, a competition to encourage the development of carbon capture and utilisation technology, has identified [ten finalists](#) that focus on

capturing CO<sub>2</sub> from power plants rather than the atmosphere.



Like finding a ball in a bag. Credit: Peter Styring, Author provided

Yet while the typical CO<sub>2</sub> concentration of [around 10%](#) (600 balls out of the 5,000) in power station exhaust is much greater than that of air, capturing the CO<sub>2</sub> would still be a costly way of purifying the gas using current technologies. You also need to remove the water vapour in the exhaust, which would require [more energy](#).

## Better sources

As it becomes more important to reduce the concentration of CO<sub>2</sub> in the atmosphere, or if you needed to produce the gas in remote locations with large renewable energy sources, direct air capture could become a viable technology. But at the moment there are CO<sub>2</sub> sources that are more concentrated and so [cheaper to harness](#).

For example, distilleries and breweries produce the gas as a waste product with high purity (over 99.5%) once any water has been removed. Cement works, steel works and other process industries also have relatively high [CO<sub>2</sub> concentrations](#). Building smaller facilities that just capture the CO<sub>2</sub> from individual factories and plants would be a cheaper way to create a new source of the gas. They may also prove a good investment at plants that need their own supply of CO<sub>2</sub> to carry out their processes.

The current CO<sub>2</sub> shortage is mainly affecting the food and drink industry. But we're also starting to see a greater push to use CO<sub>2</sub> in other industries as a way of creating a market for a substance that is otherwise a waste product contributing to dangerous [climate change](#). You can now buy chemicals and building materials that started life as CO<sub>2</sub> molecules instead of fossil fuels, for example, including mineral aggregates that actually capture more carbon than is [used to produce them](#).

As more of these CO<sub>2</sub> utilisation technologies emerge, demand for the gas will increase and so will the need for more localised production. The future is about turning a waste into a commodity.

This article was originally published on [The Conversation](#). Read the [original article](#).

## Provided by The Conversation

Citation: CO2 shortage—why can't we just pull carbon dioxide out of the air? (2018, July 4)  
retrieved 25 April 2024 from

<https://phys.org/news/2018-07-co2-shortagewhy-carbon-dioxide-air.html>

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.