

Turbocharge for lithium batteries

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Prof. Dina Fattakhova-Rohlfing. Credit: Forschungszentrum Jülich / Sascha Kreklau

A team of material researchers from Juelich, Munich, and Prague has succeeded in producing a composite material that is particularly suited for electrodes in lithium batteries. The nanocomposite material might help to significantly increase the storage capacity and lifetime of

batteries as well as their charging speed. The researchers have published their findings in the journal *Advanced Functional Materials*.

Lithium-ion batteries are the ultimate benchmark when it comes to mobile phones, tablet devices, and electric cars. Their [storage capacity](#) and power density are far superior to other rechargeable battery systems. Despite recent progress, however, smartphone batteries only last a day and [electric cars](#) need hours to be recharged. Scientists are therefore working on ways to improve the power densities and charging rates of all-round batteries. "An important factor is the anode material," explains Dina Fattakhova-Rohlfing from the Institute of Energy and Climate Research (IEK-1).

"In principle, anodes based on tin dioxide can achieve much higher specific capacities, and therefore store more energy, than the carbon anodes currently being used. They have the ability to absorb more lithium ions," says Fattakhova-Rohlfing. "Pure [tin oxide](#), however, exhibits very weak cycle stability—the storage capability of the batteries steadily decreases and they can only be recharged a few times. The volume of the anode changes with each charging and discharging cycle, which leads to it crumbling."

One way of addressing this problem is hybrid [materials](#) or nanocomposites—composite materials that contain nanoparticles. The scientists developed a material comprising [tin oxide nanoparticles](#) enriched with antimony, on a base layer of graphene. The graphene basis aids the structural stability and conductivity of the material. The tin oxide particles are less than three nanometres in size—in other words less than three millionths of a millimetre—and are directly "grown" on the graphene. The small size of the particle and its good contact with the graphene layer also improves its tolerance to volume changes—the lithium cell becomes more stable and lasts longer.

Three times more energy in one hour

"Enriching the nanoparticles with antimony ensures the material is extremely conductive," explains Fattakhova-Rohlfing. "This makes the anode much quicker, meaning that it can store one-and-a-half times more energy in just one minute than would be possible with conventional graphite anodes. It can even store three times more energy for the usual charging time of one hour."

"Such high energy densities were only previously achieved with low charging rates," says Fattakhova-Rohlfing. "Faster charging cycles always led to a quick reduction in capacity." The antimony-doped anodes developed by the scientists, however, retain 77 % of their original capacity even after 1,000 cycles.

"The nanocomposite anodes can be produced in an easy and cost-effective way. And the applied concepts can also be used for the design of other anode materials for lithium-ion batteries," explains Fattakhova-Rohlfing. "We hope that our development will pave the way for [lithium-ion batteries](#) with a significantly increased energy density and very short charging time."

More information: Florian Zoller et al, Making Ultrafast High-Capacity Anodes for Lithium-Ion Batteries via Antimony Doping of Nanosized Tin Oxide/Graphene Composites, *Advanced Functional Materials* (2018). [DOI: 10.1002/adfm.201706529](https://doi.org/10.1002/adfm.201706529)

Provided by Forschungszentrum Juelich

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