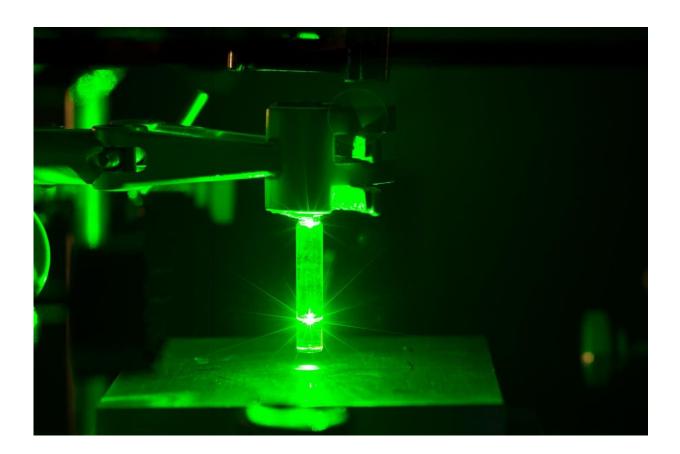


Lighting the way to switch chemical reaction pathways

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Researchers used green laser to modulated the reactivity of triazolinediones (TADs), powerful chemical coupling agents. Credit: QUT

Could the manufacture of the integrated circuits and chips for our everyday electronic devices be made simpler, safer and cheaper simply



by being able to switch coloured light on and off?

Researchers from Queensland University of Technology (QUT), Karlsruhe Institute of Technology (KIT) and Ghent University have stepped towards this by pioneering a system that modulates visible, coloured <u>light</u> to change the reactions of a powerful chemical coupling agent.

Their findings have been published in Nature Communications.

- Researchers used green laser light to control the reactivity of triazolinediones (TADs), coupling agents which swiftly create bonds with other chemicals, necessary to make materials
- Under green light the TADs stopped reacting; when the light was switched off, the TADs became highly reactive again
- The light-switching process could be repeated multiple times
- The experiment showed two different products can be created from the same set-up simply by switching coloured light on and off

Professor Barner-Kowollik, from QUT's Science and Engineering Faculty, and Ghent University's Professor Filip Du Prez co-supervised the collaborative international research project. Lead author Hannes Houck is conducting PhD studies across the three partner institutions, supported by the Research Foundation-Flanders (FWO).

Professor Barner-Kowollik said the ability to use visible light as a remotecontrolled on/off chemical reaction switch opened up possibilities for future industry applications in chemical and advanced manufacturing, including computer chip fabrication.

"At the moment, ultraviolet (UV) light, which has shorter wavelengths than light in the visible spectrum, is used in industry to drive chemical



processes," he said.

"Industrial processes making use of less harmful visible light are scarce - a sharp contrast to what occurs in nature.

"For plants, visible light plays a critical role in chemical processes. Trees harvest light during the day and use this as an energy source to grow, releasing oxygen in the process. At night, however, when light is no longer available, the chemical process is altered and plants release carbon dioxide.

"We've been inspired by such natural processes and designed a completely light-switchable chemical reaction system for the first time."

The research team said their system may be applied to create lightsensitive materials for 3D <u>laser lithography</u>, allowing printing of very small structures that could be used in such things as computer chip fabrication. 3D laser lithography is a type of 3D printing with direct laser light, and used for creating very precise structures in the micro range, such as scaffolds for cells.

QUT is exploring 3D laser lithography applications with leading KIT physicist Professor Martin Wegener.

"Contemporary <u>chip fabrication</u> is a complex, and expensive, system of chemical processes," Professor Barner-Kowollik said. "Here, due to the so-called light diffraction limit, radiation with short wavelengths - which is very harsh UV light - is used.





Researchers QUT Professor Christopher Barner-Kowollik, right, and Hannes Houck. Credit: QUT

"But what if we could use visible light to reversibly switch certain <u>chemical processes</u> and get around the light diffraction limit and print very, very small structures, for example five nanometres wide?

"Being able to switch chemical reactivity within 3D laser lithography could revolutionise chip printing, and make it cheaper, simpler and safer.

"There are substantial barriers to overcome, but potentially the system we have devised with <u>visible light</u> as a <u>chemical</u> deactivation mechanism could provide an avenue to achieve that."



More information: Hannes A. Houck et al, Controlling thermal reactivity with different colors of light, *Nature Communications* (2017). DOI: 10.1038/s41467-017-02022-0

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