

A deeper understanding of a surface phenomenon

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A Deeper Understanding of a Surface Phenomenon. Credit: Okinawa Institute of Science and Technology Graduate University - OIST

Phenomena involving surface tension are extremely complex and have applications in our everyday lives, and OIST researchers are tackling the complicated mathematics behind the physics.

The effects of <u>surface</u> tension are of central importance in many everyday phenomena: it causes small droplets of rain to stick to your windows, creates bubbles when you add detergent in your sink, and propels <u>water</u>-striding insects on the surface of ponds. It even induces the "tears of wine", a ring of clear liquid near the top of the inside rim of a glass of wine from which droplets continuously form and drop back



into the wine below. However, despite its omnipresence and long history of scientific observations, surface tension – and the interaction of liquids with different surface tensions – is not yet fully understood. Facing the complexity of the problem, simplified models have been used for decades. But now, OIST researchers have taken a new step toward a more complete understanding, reported in the *Journal of Fluid Mechanics*, by addressing an intricate property of surface tension. The outcome reveals that a commonly used approximation provides surprisingly accurate results, despite the complexity of the issue.

The work is a continuation of previous studies in which OIST scientists studied the movement of acetone droplets propelled by surface tension, gliding on the surface of water.

"Because the phenomenon is so complicated, I thought about the simplest system which has not yet been studied, trying to guess the forces applied on the droplet that would make it move," explained Dr. Stoffel Janssens. "This is a small step towards completely understanding how a droplet of liquid moves on water due to surface tension."

The researchers in the Mathematics, Mechanics, and Materials Unit headed by Prof. Eliot Fried designed a simple device in the shape of a rectangular vessel filled with water, in which a stationary cylinder is partly submerged. A surfactant – an organic molecule that decreases surface tension - is applied onto the surface of the water on one side of the cylinder, which allows the scientists to measure the profiles of the surfactant-free and the surfactant-laden portions of the water surface. These measurements helped the scientists to build a theoretical model that is used to determine the forces that act on the cylinder.

The essence of the complexity lies with the geometry: the water surface curves towards the cylinder and the curvature directly depends on the surface <u>tension</u>. With a surfactant on one side of the cylinder, the water



surface profiles on each side of the cylinder are asymmetric due to the different curvatures, which makes calculating the forces on the cylinder mathematically more complex. Until now, a simplified model ignored these curves under the assumption that the surface of the water remains completely flat on both sides of the cylinder.

The scientists described the problem with three independent approaches: they calculated the <u>mechanical forces</u> applied to the cylinder first numerically, requiring the help of a computer to provide solutions for the complex equations. Inspired by the numerical results, they moved to an analytical method, also referred to as the "pen and paper" method, to confirm the model.

"With the numerical method, you need to provide the algorithm with input values, so in a sense you are solving the equations for very specific cases," commented Prof. Fried. "You can get a feel for what is going on but you can't provide a general proof. If you do it analytically with pen and paper, without having chosen particular numerical values, then you have something that applies in general."





The complex schematic used to describe the physics behind the interaction of a partly submerged cylinder with on the left-hand side a surfactant-laden and on the right-hand side a surfactant-free water surface. With a surfactant on the left-hand side water surface, we obtain asymmetric interfaces that are described by different curves, resulting in different forces applied to the cylinder. The horizontal force component of the force that acts on the cylinder is exactly equal to the difference between the surface tensions, when measured per unit length of the cylinder. Credit: Okinawa Institute of Science and Technology Graduate University - OIST

The researchers finally analyzed the problem using a third independent approach based on energy rather than mechanical forces.

All three methods lead to the same unexpected outcome: we can safely ignore the curvature of the water-air interface since the calculated forces



are equal to those obtained by the simplified model, where the <u>water</u> <u>surface</u> is assumed to be flat.

"We could not rule out the possibility that ignoring the curvature of the water-air interface would lead to unacceptable errors," commented Dr. Janssens. "Surprisingly, however, it turns out that the simplified model used for decades is highly accurate!"

On top of solving an intricate problem, the outcome of this research provides more confidence in measurements like the Langmuir balance that has been used for the past century. Moreover, this is the first step towards understanding the complex phenomenon of an acetone droplet gliding on water because of <u>surface tension</u>.

"This work could be viewed as a reductionist approach, where we have taken a step back from that very complicated phenomenon and try to solve it piecewise starting from the most basic features," concluded Prof. Fried. "Now we can move onto a somewhat more challenging problem, for example where the cylinder is allowed to glide on top the fluid."

More information: Stoffel D. Janssens et al. Effect of a surface tension imbalance on a partly submerged cylinder, *Journal of Fluid Mechanics* (2017). DOI: 10.1017/jfm.2017.537

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