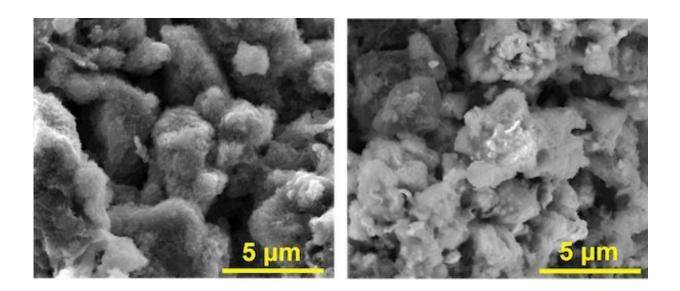


Asphalt helps lithium batteries charge faster

October 2 2017, by Mike Williams



Scanning electron microscope images show an anode of asphalt, graphene nanoribbons and lithium at left and the same material without lithium at right. The material was developed at Rice University and shows promise for high-capacity lithium batteries that charge 20 times faster than commercial lithium-ion batteries. Credit: Tour Group

A touch of asphalt may be the secret to high-capacity lithium metal batteries that charge 10 to 20 times faster than commercial lithium-ion batteries, according to Rice University scientists.

The Rice lab of chemist James Tour developed anodes comprising porous <u>carbon</u> made from <u>asphalt</u> that showed exceptional stability after more than 500 charge-discharge cycles. A high-current <u>density</u> of 20



milliamps per square centimeter demonstrated the material's promise for use in rapid charge and discharge devices that require high-power density. The finding is reported in the American Chemical Society journal *ACS Nano*.

"The capacity of these batteries is enormous, but what is equally remarkable is that we can bring them from zero charge to full charge in five minutes, rather than the typical two hours or more needed with other batteries," Tour said.

The Tour lab previously used a derivative of asphalt—specifically, untreated gilsonite, the same type used for the battery—to capture greenhouse gases from natural gas. This time, the researchers mixed asphalt with conductive graphene nanoribbons and coated the composite with <u>lithium metal</u> through electrochemical deposition.

The lab combined the anode with a sulfurized-carbon cathode to make full batteries for testing. The batteries showed a high-power density of 1,322 watts per kilogram and high-energy density of 943 watt-hours per kilogram.

Testing revealed another significant benefit: The carbon mitigated the formation of <u>lithium</u> dendrites. These mossy deposits invade a battery's electrolyte. If they extend far enough, they short-circuit the anode and cathode and can cause the battery to fail, catch fire or explode. But the asphalt-derived carbon prevents any dendrite formation.

An earlier project by the lab found that an <u>anode</u> of graphene and carbon nanotubes also prevented the formation of dendrites. Tour said the new composite is simpler.

"While the capacity between the former and this new battery is similar, approaching the theoretical limit of lithium metal, the new asphalt-



derived carbon can take up more lithium metal per unit area, and it is much simpler and cheaper to make," he said. "There is no <u>chemical</u> <u>vapor deposition</u> step, no e-beam deposition step and no need to grow nanotubes from graphene, so manufacturing is greatly simplified."

More information: Tuo Wang et al. Ultrafast Charging High Capacity Asphalt-Lithium Metal Batteries, *ACS Nano* (2017). DOI: <u>10.1021/acsnano.7b05874</u>

Provided by Rice University

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