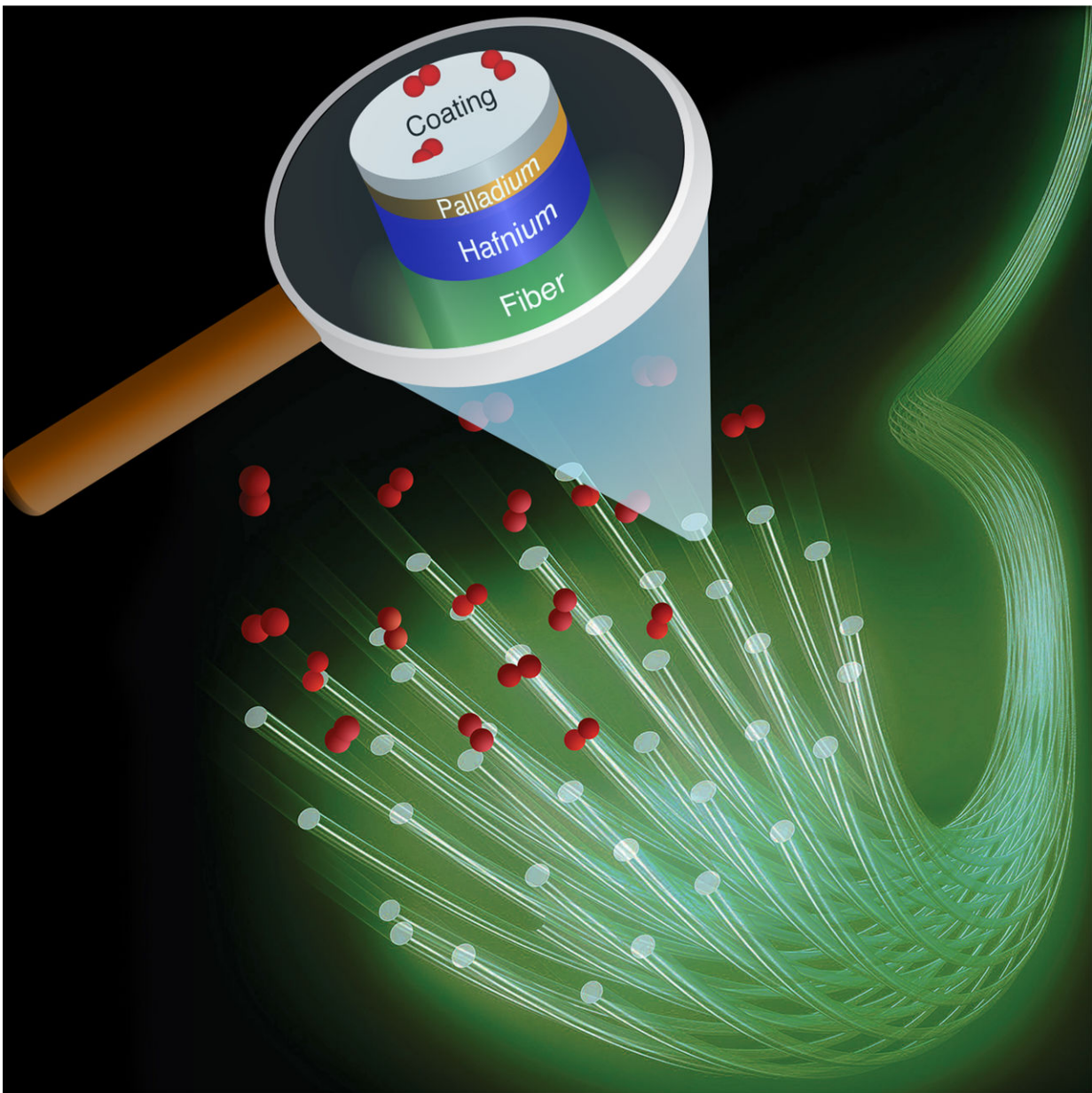


# Researchers develop extremely sensitive hydrogen sensor

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Credit: TU Delft, Bernard Dam

Hydrogen is a highly promising energy carrier. But it can also be dangerous, as it is combustible and difficult to detect. Using hydrogen safely requires sensors that can detect even the smallest of leaks. Researchers from Delft University of Technology (TU Delft, The Netherlands), KU Leuven (Belgium) and the Rutherford Appleton Laboratory (UK) have discovered that the metal hafnium is perfect for the job.

Hydrogen is a viable replacement for fossil fuels as an energy carrier in the near future. It is clean, as water vapour and heat are the only byproducts of combustion, and you can make as much of it as you want using just water and another energy source (for example, solar energy). But there are still a few hurdles to be cleared to organize infrastructure around [hydrogen](#). One of these hurdles is finding a reliable way of detecting [hydrogen gas](#). Hydrogen reacts with oxygen, which can potentially cause an explosion. It is also important to ensure that as little hydrogen as possible leaks into the atmosphere.

## Highly sensitive

Although it is difficult to detect hydrogen, it is not impossible. Optical sensors exist. These are materials that absorb [hydrogen atoms](#), a process that alters their reflectivity. This change in reflectivity can be measured, thus providing information about the amount of hydrogen in a particular location.

"Until now, pure palladium was mainly used as an optical hydrogen sensor," says Prof. Bernard Dam from TU Delft. "But over the last few years, we at Delft have shown that a gold-palladium alloy is a much

better sensor. Fellow researchers around the world are also studying this." Palladium-gold, which is also used to make jewellery, has the advantage that it works at room temperature. Unfortunately, it is unable to detect low hydrogen pressures.

## Easy to calibrate

Research led by PhD candidate Christiaan Boelsma from TU Delft, which has now been published in *Nature Communications*, shows that hafnium does have this sensitivity. Bernard Dam says, "The unique property of this material is that it can optically measure a minimum of six orders of magnitude in pressure. The lowest pressure measured is  $10^{-7}$  atmospheres, but this pressure is determined by the measurement setup. It looks as if a pressure of three orders of magnitude lower could be measured with hafnium, but we need to do more research to confirm this." Another benefit is that the optical properties of hafnium change linearly with the [pressure](#) and temperature of the material. "This makes hafnium sensors very easy to calibrate," says Dam.

So is hafnium a better hydrogen sensor than palladium-gold in every respect? No, since the material works best at a temperature of around 120 degrees Celsius. Dam thinks that this problem can be solved by placing a thin layer of [hafnium](#) on top of an optical fibre, before heating the fibre with a warm-up LED (see picture).

## Hydrogen economy

Delft is currently conducting extensive research into hydrogen. Late last year, Prof. Fokko Mulder unveiled the "Battolyser," a device that combines electricity storage and [hydrogen production](#) in one system. The Battolyser is a cheap way of creating and storing hydrogen, and brings the so-called '[hydrogen economy](#)' a step closer. This research into

sensitive hydrogen [sensors](#) represents another step in the same direction.

**More information:** C. Boelsma et al. Hafnium—an optical hydrogen sensor spanning six orders in pressure, *Nature Communications* (2017).  
[DOI: 10.1038/ncomms15718](https://doi.org/10.1038/ncomms15718)

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