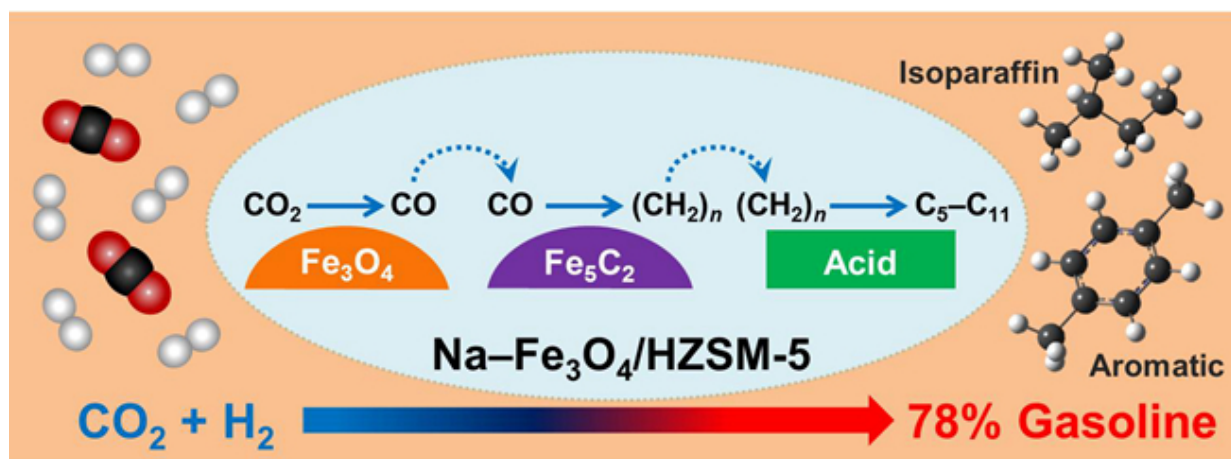


Scientists develop efficient multifunctional catalyst for CO₂ hydrogenation to gasoline

May 2 2017



CO₂ hydrogenation to gasoline-range hydrocarbons over Na-Fe₃O₄/Zeolite multifunctional catalyst. Credit: WEI Jian

Converting CO₂ from a detrimental greenhouse gas into value-added liquid fuels not only contributes to mitigating CO₂ emissions, but also reduces dependence on petrochemicals. However, since CO₂ is a fully oxidized, thermodynamically stable and chemically inert molecule, the activation of CO₂ and its hydrogenation to hydrocarbons or other alcohols comprises challenging tasks. Most research to date, unsurprisingly, is focusing on selective hydrogenation of CO₂ to short-chain products, while few studies on long-chain hydrocarbons, such as gasoline-range (C₅-C₁₁) hydrocarbons. The key to this process is to

search for a highly efficient catalyst.

The research team led by Dr. SUN Jian and Prof. GE Qingjie in Dalian Institute of Chemical Physics has succeeded in preparing a highly efficient, stable, and multifunctional Na-Fe₃O₄/HZSM-5 catalyst for the direct production of gasoline from CO₂ hydrogenation. This catalyst exhibited 78 percent selectivity to C₅-C₁₁ as well as low CH₄ and CO selectivity under industrial relevant conditions. And gasoline fractions are mainly isoparaffins and aromatics, thus favoring the octane number. Moreover, the multifunctional catalyst exhibited a remarkable stability for 1,000 h on stream, which definitely has the potential as a promising industrial catalyst for CO₂ utilization to liquid fuels.

In-depth characterizations indicate that this [catalyst](#) enables RWGS over Fe₃O₄ sites, olefin synthesis over Fe₅C₂ sites and oligomerization/aromatization/isomerization over zeolite acid sites. The concerted action of the active sites calls for precise control of their structures and proximity. This study paves a new path for the synthesis of liquid fuels by utilizing CO₂ and H₂. Furthermore, it provides an important approach for dealing with the intermittency of renewable sources (sun, wind and so on) by storing energy in [liquid fuels](#).

This work was published in *Nature Communications*.

More information: Jian Wei et al, Directly converting CO₂ into a gasoline fuel, *Nature Communications* (2017). [DOI: 10.1038/ncomms15174](https://doi.org/10.1038/ncomms15174)

Provided by Chinese Academy of Sciences

Citation: Scientists develop efficient multifunctional catalyst for CO₂ hydrogenation to gasoline

(2017, May 2) retrieved 26 April 2024 from <https://phys.org/news/2017-05-scientists-efficient-multifunctional-catalyst-co2.html>

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