

Exploring the conversion of heat to electricity in single molecules

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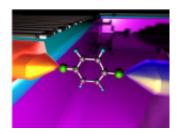


Figure 1: A schematic model of a single-molecule device. A molecule is interconnected to hot and cold electrodes via chemical bonds. Credit: Osaka University

The direct conversion of a temperature difference into electricity, known as the thermoelectric effect, is an environmentally friendly approach to directly harvesting electricity from heat. The ability of a material to convert heat to electricity is measured by its thermoelectric figure of merit. Materials with a high thermoelectric figure of merit are thus widely desired for use in energy harvesting. Quantum confinement effects in nanomaterials arising from their discrete electronic states may increase their thermoelectric figure of merit. In particular, a single molecule bridging two electrodes displays quantum confinement. Optimization of the electronic states of a single molecule bridging electrodes could yield a large thermoelectric effect. The contact between the molecule and electrodes will also influence its thermoelectric behavior. However, this relationship has seldom been considered



because of technical difficulties.

Researchers at Osaka University have recently investigated the influence that the geometry of single molecule—<u>electrode</u> contacts has on the thermoelectric behavior of the molecule. As reported in a recent edition of *Scientific Reports*, they simultaneously measured the <u>electrical</u> <u>conductance</u> and thermovoltage of <u>molecules</u> with different groups anchoring the molecules to the electrodes at room temperature in vacuum.

The team first fabricated structures consisting of gold electrodes bridged by various single molecules. The distance between the electrodes, which were held under a temperature gradient, was repeatedly increased and decreased while the electrical conductance and thermovoltage of each structure was measured.

"We investigated the thermoelectric characteristics of various single benzene-based molecules with an emphasis on influence of their junction structures," says corresponding author Makusu Tsutsui. "The molecules displayed different behavior depending on their electrodeanchoring groups, and all molecule types displayed multiple thermovoltage states."

The multiple thermovoltage states of the molecules were investigated by thermoelectric measurements and theoretical analysis. The largest thermoelectric effect was observed for structures containing a stretched thiol linkage with the gold electrode. The increased thermovoltage of the structures with a stretched gold—thiol bond was attributed to this configuration shifting the energy level of the molecule involved in electron transport to a more favorable position.

"The observed dependence of thermovoltage on the anchoring group in the junction structures reveals a way to modulate the <u>thermoelectric</u>



performance of single-molecule devices," explains Tsutsui.

The group's results expand our understanding of how the geometry of a single-molecule device can influence its thermoelectric figure of merit. These findings should contribute to the development of single-molecule thermoelectric devices that can efficiently derive electricity from heat.

More information: Makusu Tsutsui et al. Roles of vacuum tunnelling and contact mechanics in single-molecule thermopower, *Scientific Reports* (2017). DOI: 10.1038/srep44276

Provided by Osaka University

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