

# Energy-efficient green route to magnesium production

May 18 2017

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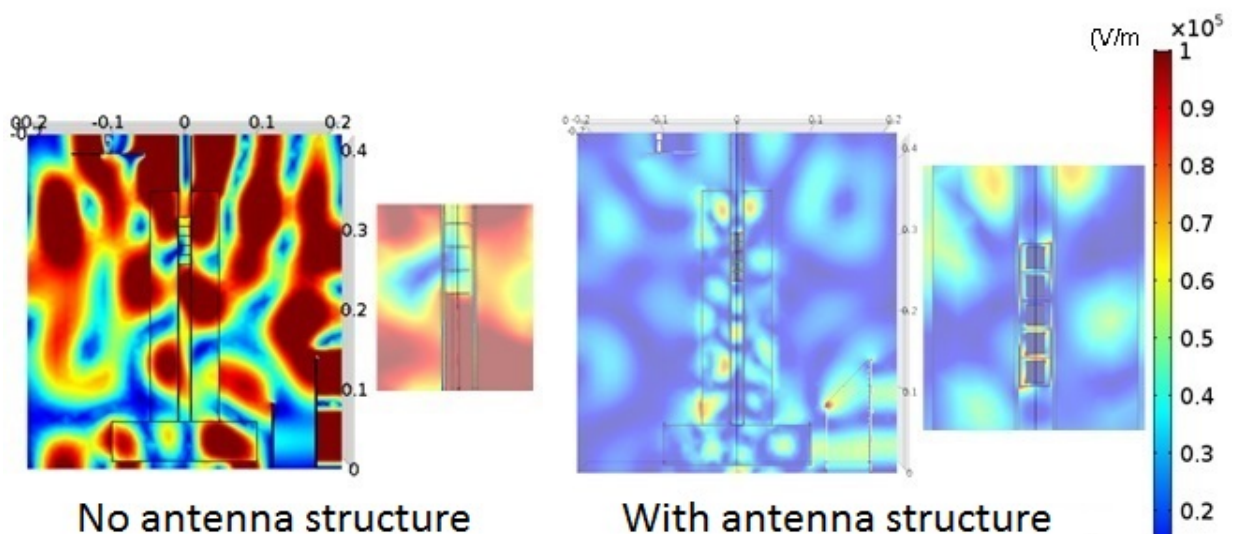
(Left) Actual pellet of dolomite and ferrosilicon. White portion is dolomite rich, and the black portion is ferrosilicon rich. Ferrosilicon is concentrated in the center portion. (Right) Five stacked to form the microwave wavelength (antenna structure). Credit: Tokyo Institute of Technology

A research group led by Professor Yuji Wada and Adjunct Professor Satoshi Fujii of the Tokyo Institute of Technology has devised a magnesium smelting method that uses nearly 70 percent less energy than conventional methods by using microwaves.

Dolomite ore (MgO, CaO), which is a raw material for [magnesium metal](#), does not absorb microwave [energy](#) well and does not generate heat. When electrically conductivity ferrosilicon (FeSi) used as a reducing agent was mixed with the raw dolomite material and made into an antenna structure, it more easily absorbed the microwave energy and reduced in temperature. Internal heating and contact point heating, which are microwave characteristics, were observed, and the average reaction temperature for this smelting was lowered from the conventional 1,200 - 1,400°C to 1,000°C.

This research result was published in the April 12th issue of *Scientific Reports*.

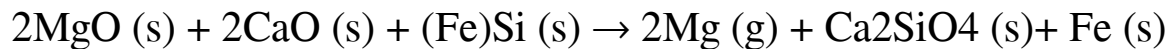
Currently, the smelting of magnesium metal is mainly performed using the Pidgeon method (a thermal reduction method) in which the material temperature is raised using a large amount of coal as the [heat source](#). About 80 percent of magnesium metal is produced in China. A large amount of coal is consumed for smelting, resulting in the generation of the air pollutant PM 2.5 (fine particulate matter) and the release of carbon dioxide to the atmosphere, which are major problems.



Difference in the distribution of the electric field in the applicator with and without the antenna structure through simulation. Credit: Tokyo Institute of Technology

The Pidgeon method is a technique for heating dolomite ore and silicon iron to high temperatures and then cooling the evaporated magnesium to obtain magnesium metal.

Where (s)=Solid and (g)=Gas:

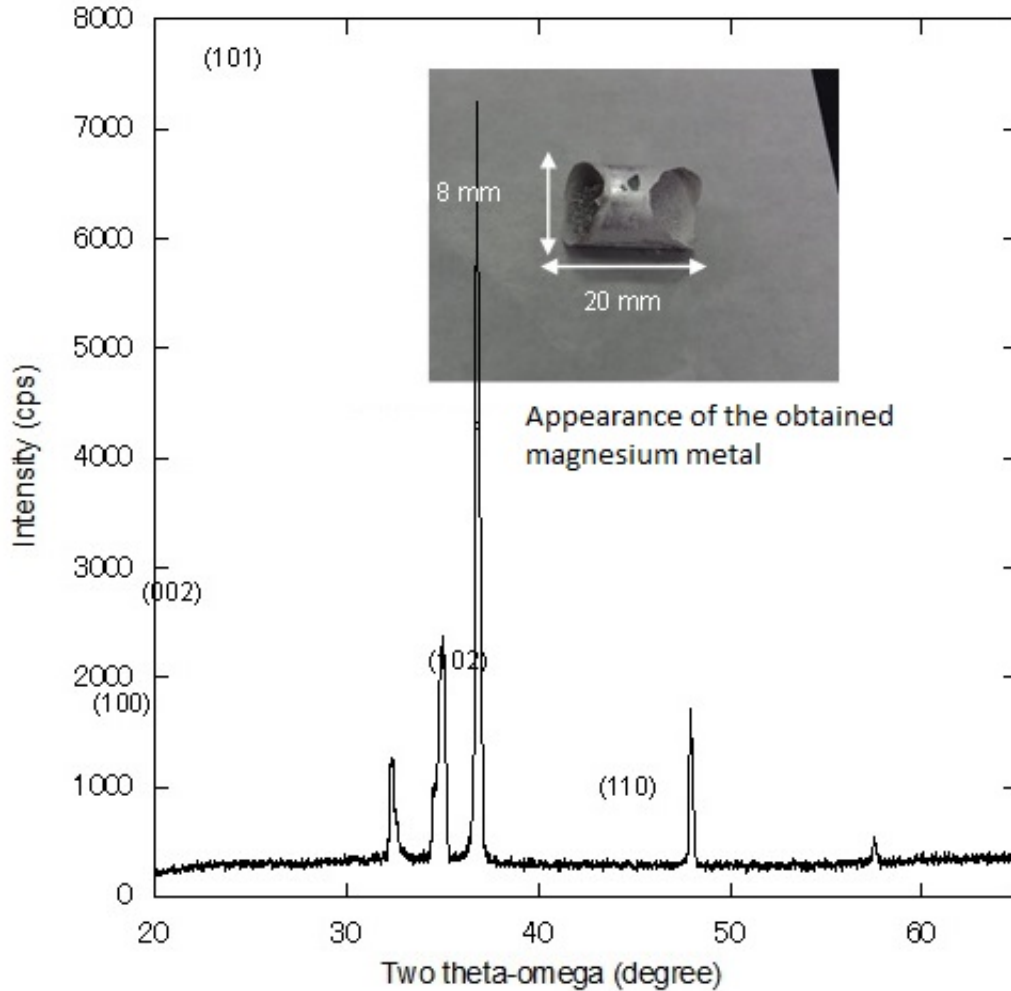


Dolomite mineral: MgO, CaO; Ferrosilicon: FeSi

Heat source: Coal

By using ferrosilicon as the reducing agent, devising the shape of the raw material pellet obtained by mixing dolomite and ferrosilicon and forming it as an antenna with a resonance structure of 2.45 GHz (the frequency for microwave ovens), it was possible to confine the [microwave energy](#) to the pellet.

In a small-scale experimental reactor, 1g of magnesium metal was smelted successfully. Also, in order to accurately estimate the energy, a demonstration furnace about five times larger than the experimental furnace was produced and experiments were conducted, resulting in the successful smelting of about seven grams of magnesium metal. This can reduce energy by 68.6 percent compared with the conventional method.



X-ray diffraction result of the obtained magnesium metal. Magnesium peak is observed. Credit: Tokyo Institute of Technology

## Future Developments

This technique is promising for the high-temperature reduction process of oxides. In the future, through further development of this research, it will be applied to the smelting of other metal [materials](#) to save energy with steel, metals, materials and chemistry, and reduce [carbon dioxide](#) pollution, which is one of the causes of global warming.

**More information:** Yuji Wada et al, Smelting Magnesium Metal using a Microwave Pidgeon Method, *Scientific Reports* (2017). [DOI: 10.1038/srep46512](https://doi.org/10.1038/srep46512)

Provided by Tokyo Institute of Technology

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