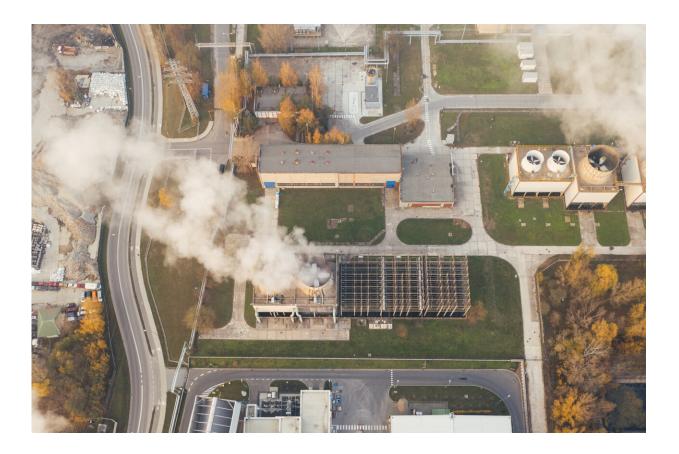


## **Opinion: Worthless mining waste could suck CO2 out of the atmosphere and reverse emissions**

April 21 2017, by Simon Redfern



Credit: Marcin Jozwiak from Pexels

The Paris Agreement commits nations to limiting global warming to less than 2°C by the end of the century. However, it is becoming increasingly



apparent that, to meet such a massive challenge, societies will need to do more than simply reduce and limit carbon emissions. It seems likely that large scale removal of greenhouse gases from the atmosphere may be called for: so-called "negative emissions".

One possibility is to use waste material from mining to trap  $CO_2$  into new minerals, locking it out of the atmosphere. The idea is to exploit and accelerate the same geological processes that have regulated Earth's climate and surface environment over the 4.5 billion years of its existence.

Across the world, deep and open-pit mining operations have left behind huge piles of worthless rubble – the "overburden" of rock or soil that once lay above the useful coal or metal ore. Often, this rubble is stored in dumps alongside tiny fragments of mining waste – the "tailings" or "fines" left over after processing the ore. The fine-grained waste is particularly reactive, chemically, since more surface is exposed.

A lot of energy is spent on extracting and crushing all this waste. However, breaking rocks into smaller pieces exposes more fresh surfaces, which can react with  $CO_2$ . In this sense, energy used in mining could itself be harvested and used to reduce atmospheric carbon.

This is one of the four themes of a <u>new £8.6m research programme</u> launched by the UK's <u>Natural Environment Research Council</u>, which will investigate new ways to reverse emissions and remove <u>greenhouse gases</u> from the atmosphere.





Spoil tips from current and historic mining operations, such as this gold mine in Kazakhstan, could provide new ways to draw  $CO_2$  from the atmosphere. Credit: Ainur Seitkan, Earth Sciences, University of Cambridge

The process we want to speed up is the "carbonate-silicate cycle", also known as the <u>slow carbon cycle</u>. Natural silicate rocks like granite and basalt, common at Earth's surface, play a key part in regulating carbon in the atmosphere and oceans by removing  $CO_2$  from the atmosphere and turning it into carbonate rocks like chalk and limestone.

Atmospheric  $CO_2$  and water can react with the <u>silicate rocks</u> to dissolve elements they contain like calcium and magnesium into the water, which



also soaks up the  $CO_2$  as bicarbonate. This weak solution is the natural river water that flows to the oceans, which hold more than <u>60 times more carbon than the atmosphere</u>. It is here, in the oceans, that the calcium and bicarbonate can recombine, over millions of years, and crystallise as calcite or chalk, often instigated by marine organisms as they build their shells.

Today, rivers deliver <u>hundreds of millions</u> of tonnes of carbon each year into the oceans, but this is still around <u>30 times less</u> than the rate of <u>carbon</u> emission into the <u>atmosphere</u> due to fossil fuel burning. Given immense geological time scales, these processes would return atmospheric  $CO_2$  to its normal steady state. But we don't have time: the blip in  $CO_2$  emissions from industrialisation easily unbalances nature's best efforts.

The natural process takes millions of years – but can we do it in decades? Scientists looking at accelerated mine waste dissolution will attempt to answer a number of pressing questions. The group at <u>Cambridge</u> which I lead will be investigating whether we can speed up the process of silicate minerals from pre-existing mine waste being dissolved into water. We may even be able to harness friendly microbes to enhance the reaction rates.

Another part of the same project, conducted by colleagues in <u>Oxford</u>, <u>Southampton</u> and <u>Cardiff</u>, will study how the calcium and magnesium released from the silicate mine <u>waste</u> can react back into minerals like calcite, to lock  $CO_2$  back into solid minerals into the geological future.

Whether this can be done effectively without requiring further fossil fuel energy, and at a scale that is viable and effective, remains to be seen. But accelerating the reaction rates in mining wastes should help us move at least some way towards reaching our climate targets.



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