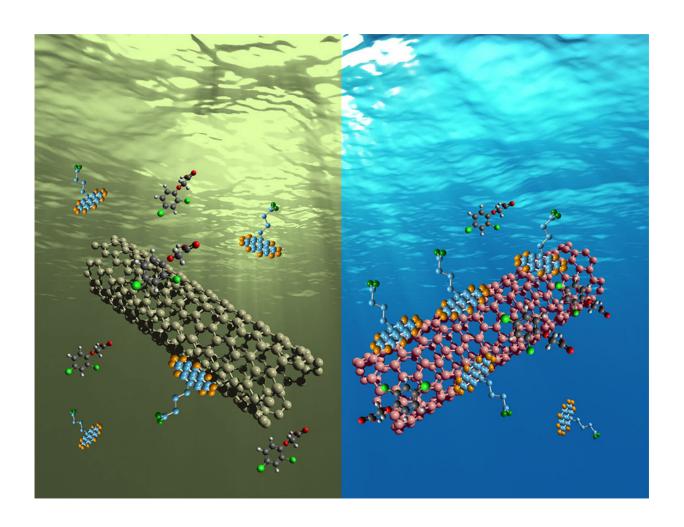


Reusable carbon nanotubes could be the water filter of the future

March 29 2017, by Susan Gawlowicz



Single-walled carbon nanotubes filter dirty water in experiments at RIT. Credit: John-David Rocha and Reginald Rogers



A new class of carbon nanotubes could be the next-generation clean-up crew for toxic sludge and contaminated water, say researchers at Rochester Institute of Technology.

Enhanced single-walled <u>carbon nanotubes</u> offer a more effective and sustainable approach to <u>water</u> treatment and remediation than the standard industry <u>materials</u>—silicon gels and activated carbon—according to a paper published in the March issue of *Environmental Science Water: Research and Technology*.

RIT researchers John-David Rocha and Reginald Rogers, authors of the study, demonstrate the potential of this emerging technology to clean polluted water. Their work applies carbon nanotubes to environmental problems in a specific new way that builds on a nearly two decades of nanomaterial research. Nanotubes are more commonly associated with fuel-cell research.

"This aspect is new—taking knowledge of carbon nanotubes and their properties and realizing, with new processing and characterization techniques, the advantages nanotubes can provide for removing contaminants for water," said Rocha, assistant professor in the School of Chemistry and Materials Science in RIT's College of Science.

Rocha and Rogers are advancing nanotube technology for environmental remediation and water filtration for home use.

"We have shown that we can regenerate these materials," said Rogers, assistant professor of chemical engineering in RIT's Kate Gleason College of Engineering. "In the future, when your water filter finally gets saturated, put it in the microwave for about five minutes and the impurities will get evaporated off."

Carbon nanotubes are storage units measuring about 50,000 times



smaller than the width of a human hair. Carbon reduced to the nanoscale defies the rules of physics and operates in a world of quantum mechanics in which small materials become mighty.

"We know carbon as graphite for our pencils, as diamonds, as soot," Rocha said. "We can transform that soot or graphite into a nanometertype material known as graphene."

A single-walled <u>carbon nanotube</u> is created when a sheet of graphene is rolled up. The physical change alters the material's chemical structure and determines how it behaves. The result is "one of the most heat conductive and electrically conductive materials in the world," Rocha said. "These are properties that only come into play because they are at the nanometer scale."

The RIT researchers created new techniques for manipulating the tiny materials. Rocha developed a method for isolating high-quality, single-walled carbon nanotubes and for sorting them according to their semiconductive or metallic properties. Rogers redistributed the pure carbon nanotubes into thin papers akin to carbon-copy paper.

"Once the papers are formed, now we have the adsorbent—what we use to pull the contaminants out of water," Rogers said.

The filtration process works because "carbon nanotubes dislike water," he added. Only the organic contaminants in the water stick to the nanotube, not the water molecules.

"This type of application has not been done before," Rogers said.
"Nanotubes used in this respect is new."

More information: John-David R. Rocha et al. Emerging investigators series: highly effective adsorption of organic aromatic molecules from



aqueous environments by electronically sorted single-walled carbon nanotubes, *Environ. Sci.: Water Res. Technol.* (2017). DOI: 10.1039/C6EW00284F

Provided by Rochester Institute of Technology

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