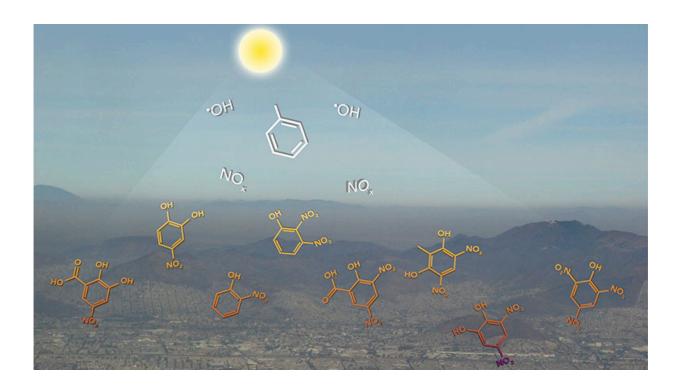


Light-absorbing particles identified in environmental chamber experiments

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The brown haze hanging over Mexico City is largely composed of brown carbon. Formed through a complex mix of pollution-born and natural-source chemicals in the atmosphere, brown carbon is highly light absorbent. Looking at the picture, imagine standing on the street under this haze: it is a blanket of warmth with an unpleasant odor. This blanket not only impacts the well-being of residents living under its spell, it affects the balance of energy between the Earth and sun. Understanding its impact will help scientists plug brown carbon into the models helping project climate impacts. Credit: Pacific Northwest National Laboratory



Ubiquitous yet mysterious. Light-absorbing, carbon-containing particles, also known as brown carbon, are prevalent in the atmosphere yet highly variable. Scientists are working to fill in knowledge gaps in how they form, their chemical properties, and how much light they absorb.

A research team led by Pacific Northwest National Laboratory conducted controlled experiments in PNNL's environmental chamber to mimic the <u>chemical</u> reactions taking place in the atmosphere. They looked at the impact of different particle ingredients (precursors), the reaction conditions (temperature and <u>light</u>), and the amount of atmospheric moisture (<u>relative humidity</u>) on brown carbon formation and aging. The results suggest that brown carbon formed from common, human-caused pollution could have a significant impact on the energy balance of the Earth. Further, their work indicates the need to carefully revisit how brown carbon is represented in climate models.

There are two types of carbon-containing (organic) atmospheric <u>particles</u> and both are very good at absorbing sunlight—thus, important to consider as atmospheric warmers. Black carbon particles are fine particulates emitted by high-temperature burning primarily of fossil fuels, such as in diesel engines. Brown carbon particles are emitted by combustion of organic materials, or biomass, such as natural or humancaused fires, crop residue, and land clearing. However, recent research has shown that brown carbon can also be formed when mixtures of natural and human-made chemicals react together in the atmosphere in the presence of sunlight to produce "secondary" organic aerosols (SOA). The research suggests that this secondary brown carbon could have significant impacts on local or regional climate, providing new insights on how to better evaluate SOA impacts in climate models.





Dr. John Shilling is shown standing in PNNL's cutting-edge environmental chamber, used by scientists to simulate, in controlled conditions, the chemical reactions and microphysical processes that occur in the natural atmosphere. The data generated in this laboratory is used to reduce the uncertainty associated with representing the organic aerosol lifecycle in climate models. The Atmospheric Research Chamber is located in the Atmospheric Measurements Laboratory in Richland, Washington. Credit: Pacific Northwest National Laboratory

The PNNL research team, including collaborators from Concordia University, conducted a series of experiments in PNNL's environmental chamber to investigate the effects of several well-controlled parameters



on brown carbon absorption, including volatile organic <u>carbon</u> precursor types, nitric oxide concentrations, relative humidity levels and light exposure (photolysis aging time). In addition to using the solution-based spectrometer to measure light absorption, they also applied massspectrometry techniques to dig into the chemical compositions of SOA products. These measurements combined provide information on both refractive indices, or light absorption coefficients of <u>brown carbon</u> and their chemical properties.

Researchers are planning a more detailed analysis of chemical compositions responding to <u>light absorption</u>. The inherent SOA formation mechanisms, which are substantially altered by environmental conditions, are an additional avenue of inquiry.

More information: Jiumeng Liu et al. Optical properties and aging of light-absorbing secondary organic aerosol, *Atmospheric Chemistry and Physics* (2016). DOI: 10.5194/acp-16-12815-2016

Provided by Pacific Northwest National Laboratory

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