

What's behind the durian fruit's notorious stench

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Credit: American Chemical Society

Most people who have tried durian either love it or hate it. The fruit's yellowish flesh is sweet and custard-like, but it comes with an overpowering stench of garbage. Scientists studying the unique fruit have now analyzed a set of 20 stinky and fruity chemical ingredients and found that a mere two compounds can re-create the overall smell. Their findings appear in ACS' *Journal of Agricultural and Food Chemistry*.

Native to southeast Asia - where the fruit is considered a delicacy, but also banned from some public spaces due to its odor—its appeal has spread to Westerners who relish trying distinct foods from around the



world. Scientists interested in the fruit have identified several compounds that contribute to its smell, which has been said to reek of gym socks, garbage and rotting meat. Curious to better understand the complex scent, Martin Steinhaus and colleagues parsed the odor compounds further.

The researchers calculated the "odor activity values" of 19 of the durian's smelly compounds to see which ones were the most potent. Among the strongest were compounds that smelled of fruit, rotten onion and roasted onion. These were followed by chemicals with strong notes of cabbage and sulfur. Further experimentation found that putting just two specific compounds together—fruity ethyl (2S)-2-methylbutanoate and oniony 1-(ethylsulfanyl)ethanethiol—effectively resembled the <u>fruit</u> 's entire set of odoriferous and fragrant <u>compounds</u>.

More information: Jia-Xiao Li et al. Insights into the Key Compounds of Durian (L. 'Monthong') Pulp Odor by Odorant Quantitation and Aroma Simulation Experiments, *Journal of Agricultural and Food Chemistry* (2017). DOI: 10.1021/acs.jafc.6b05299

Abstract

Sixteen compounds, previously identified as potent odorants by application of an aroma extract dilution analysis and the gas chromatography—olfactometry analysis of static headspace samples, were quantitated in the pulp of durians, variety Monthong, and odor activity values (OAVs) were calculated by dividing the concentrations obtained by the odor thresholds of the compounds in water. In combination with data recently reported for hydrogen sulfide and short-chain alkanethiols, OAVs > 1 were obtained for 19 compounds, among which ethyl (2S)-2-methylbutanoate (fruity; OAV 1700000), ethanethiol (rotten onion; OAV 480000), and 1-(ethylsulfanyl)ethane-1-thiol (roasted onion; OAV 250000) were the most potent, followed by methanethiol (rotten, cabbage; OAV 45000), ethane-1,1-dithiol (sulfury, durian; OAV 23000),



and ethyl 2-methylpropanoate (fruity; OAV 22000). Aroma simulation and omission experiments revealed that the overall odor of durian pulp could be mimicked by only two compounds, namely, ethyl (2S)-2-methylbutanoate and 1-(ethylsulfanyl)ethane-1-thiol, when combined in their natural concentrations.

Provided by American Chemical Society

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