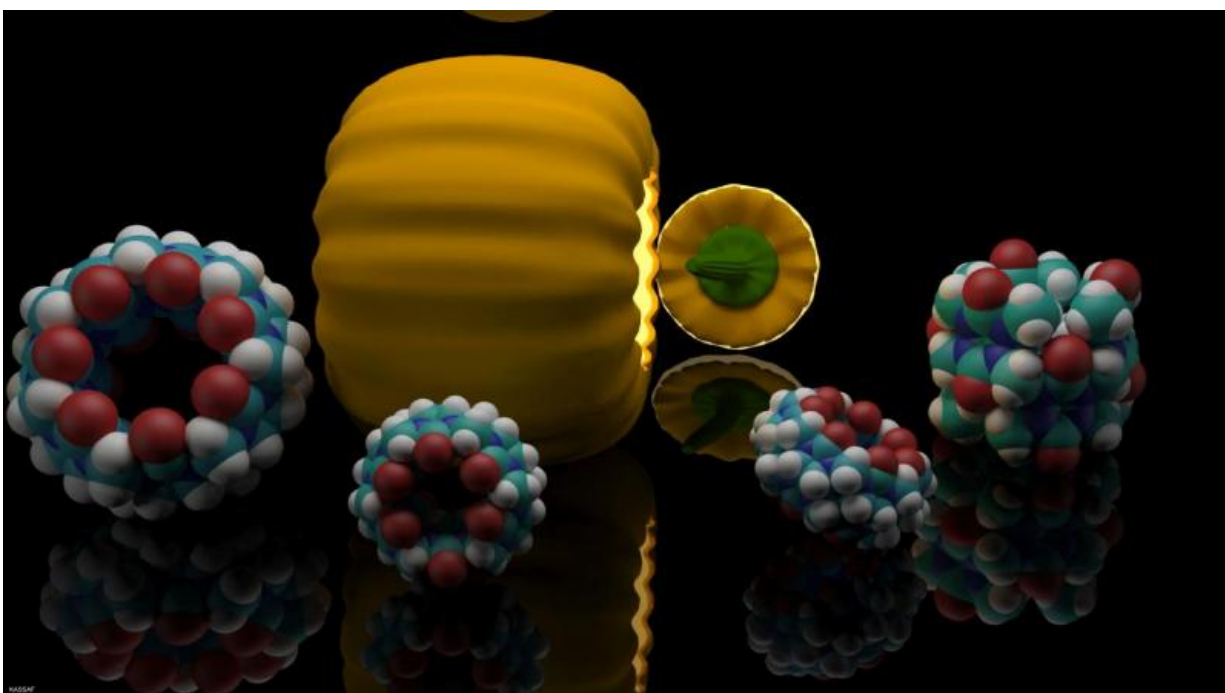


Scientists report a promising new transporter for active substances

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Molecules of the class of cucurbiturils assume the shape of a hollowed-out pumpkin and, thus, have space for transporting active substances. Credit: Jacobs University/Khaleel Assaf

Synthetic hosts transport medical substances and hormones into the body and release them at specific points. They enclose the active substances in a cavity. In the case of steroids, this task has been accomplished mainly by ring-shaped glucose molecules. Now, scientists of KIT and Jacobs

University Bremen have discovered a new class of host molecules, barrel-shaped cucurbiturils. They can make mostly insoluble steroids, such as cortisone or estradiol, act more gently and more efficiently.

"We found that the host class of cucurbiturils has a higher affinity to [steroids](#) used for medical purposes than cyclodextrines," Frank Biedermann, scientist of the KIT Institute of Nanotechnology, explains. The ring-shaped glucose of cyclodextrine is a relatively large molecule that can be adapted easily thanks to its flexible shape. On the other hand, it also collapses more easily. To produce the required water solubility, an increased dose of the active substance and transporter medium is required. This increases undesired side effects of the respective medicine. In addition, cyclodextrines preferably bind to thinner molecule chains, such as cholesterol that are irrelevant as active substances.

Based on experiments with the hormones testosterone and estradiol, the inflammation inhibitor cortisol, and the muscle relaxants pancuronium and vecuronium, the experts proved that steroid cucurbiturils are far more stable and more strongly increase water solubility of their host molecule. In addition, they can act as deposit of the active substance, because they also remain stable in blood serum and gastric acid and release steroids more slowly in the body. The new host group is biocompatible and can be applied at a reduced dose and more selectively. As a consequence, medicine based on steroids can act much better, side effects can be reduced, and production costs decrease.

"With the help of cucurbiturils, new and more efficient formulations of steroid substances might be developed in the future," Werner Nau, expert for supramolecular chemistry at Jacobs University Bremen, says. But not only pharmacology, also biological fundamental research will profit from the new substance transporters in the opinion of both scientists. In combination with an indicator dye, cucurbiturils allow for

observing in real time the interaction between steroids and enzymes on their way through the body.

A new research project just started by Biedermann at KIT is aimed at showing the variety of ways in which these molecules can be used. The scientist wants to demonstrate that cucurbiturils do not only mobilize steroids, but can also immobilize them again e.g. when they enter the groundwater with body excretions.

More information: Alexandra I. Lazar et al. Nanomolar Binding of Steroids to Cucurbiturils: Selectivity and Applications, *Journal of the American Chemical Society* (2016). [DOI: 10.1021/jacs.6b07655](https://doi.org/10.1021/jacs.6b07655)

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