

Four ways the ozone hole is linked to climate, and one way it isn't

November 11 2016, by Rebecca Lindsey



A screenshot from an animation of ~200 three-second exposures displaying the awesome movement of Auroras. Patrick Cullis took pictures of the Milky Way over Atmospheric Research Observatory at the South Pole, Antarctica, and then switched the camera 90 degrees toward the Amundsen-Scott South Pole Station, where an amazing aurora built on June 23, 2009. Credit: NOAA Headquarters



To mark the peak of ozone hole season, we're highlighting four ways that climate and the ozone hole are related—and one important way they aren't. But before you read any further, please take a few moments to enjoy a gorgeous view of the night sky (and an aurora) over the South Pole captured by Patrick Cullis, a NOAA/CIRES scientist who was stationed there from 2008-2009. Antarctica's long months of polar darkness are a key ingredient to the making of the yearly ozone hole, but that doesn't diminish their unique beauty.

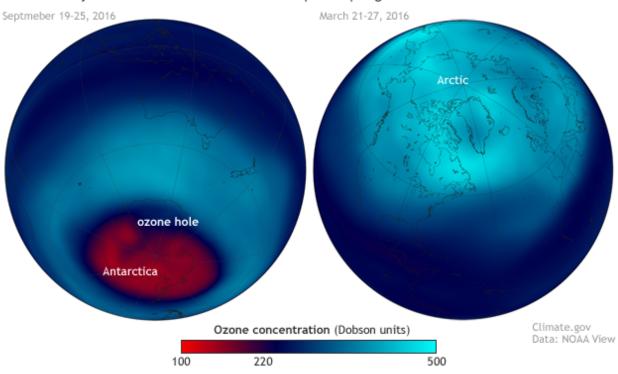
Antarctica is the only place on Earth where it's reliably cold enough for an ozone hole to form

The <u>ozone hole</u> is an annual thin spot that forms in the <u>stratospheric</u> <u>ozone layer</u> over Antarctica in mid-September and October. When it comes to the ozone hole, chlorine is the enemy. The chlorine comes from chlorofluorocarbons (CFCs, for short), which were widely used in early refrigeration and cooling systems. Under most atmospheric conditions, when CFCs begin to degrade, the chlorine they contain first gets incorporated into variety of smaller molecules that do not directly harm the ozone layer.

The large-scale transformation of relatively harmless forms of chlorine into an army of ozone-destroying assassins only takes place in one environment: on the surface of droplets and crystals in an unusual kind of cloud. Made of a mixture of water and nitric or sulfuric acid, these polar stratospheric clouds only form where temperatures drop to at least -78 °C (-108 °F).

Such conditions are periodically met over the Arctic and other high latitude locations, but the only place where such deep cold persists over a wide area is in the stratosphere over Antarctica in the winter and early spring.





The ozone layer in Southern and Northern Hemisphere spring

Stratospheric ozone concentrations in spring 2016 in the Southern (left, September 19–26) and Northern (right, March 21-27) Hemispheres. Each year, ozone amounts drop below 220 Dobson Units—the threshold that marks the start of an ozone hole—across a large area of Antarctica. Northern Hemisphere values are rarely that low. NOAA Climate.gov image based on satellite ozone data. Credit: NOAA Headquarters

Southern Hemisphere spring is ozone hole season

The ozone hole doesn't exist year round: it's seasonal. During the Antarctic winter, when polar stratospheric clouds become widespread, chemical reactions convert less reactive forms of chlorine into large amounts of highly reactive forms. These highly reactive gases have such a weak hold on their chlorine atoms that they are only stable in the dark.



When the Sun rises at the end of Southern Hemisphere winter, sunlight degrades these reactive gases, releasing "free radicals" of chlorine. A single chlorine-containing free radical can catalyze the destruction of thousands and thousands of ozone molecules. Ozone destruction usually peaks in mid-October. Ozone loss tapers off in late spring as the polar vortex weakens. Temperatures rise, and fewer clouds form. Ozone-rich air from lower latitudes mixes back into the polar stratosphere, and the ozone hole disappears until the next spring.



As colorful as any aurora, iridescent polar stratospheric clouds glitter in the low light of the spring sunrise at McMurdo Station in Antarctica in late August or early September 2003. Credit: Seth White, UNAVCO

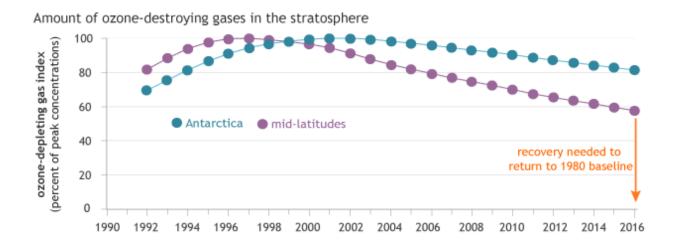


The largest ozone holes are often associated with the coldest winters

Thanks to the Montreal Protocol, the production and use of ozonedamaging chemicals has declined. But CFCs are extremely long-lived under most atmospheric conditions, so it will be decades before the global concentration returns to pre-1980 levels.

In the meantime, the size of the ozone hole in any given year is influenced by natural variability in global atmospheric circulation. These variations influence the spread of ozone-depleting chemicals into the polar stratosphere, and, most importantly, they determine the extent and severity of cold temperatures.

The final conversion of chlorine from degraded CFCs into its ozonedestroying, free radical form takes place only within polar stratospheric clouds. In general, the colder the winter, the more clouds. The more clouds, the larger the pool of highly reactive chlorine-containing gases, and the worse the ozone hole in spring.



Ozone-depleting gases reached their peak concentrations in the mid-latitudes (purple) in the mid-1990s and over Antarctica (blue) in the early 2000s. Because it takes decades for chlorofluorocarbons to degrade, it will be many decades



before the ozone hole recovers. Credit: NOAA Climate.gov graph, based on data from NOAA Earth System Resources Lab

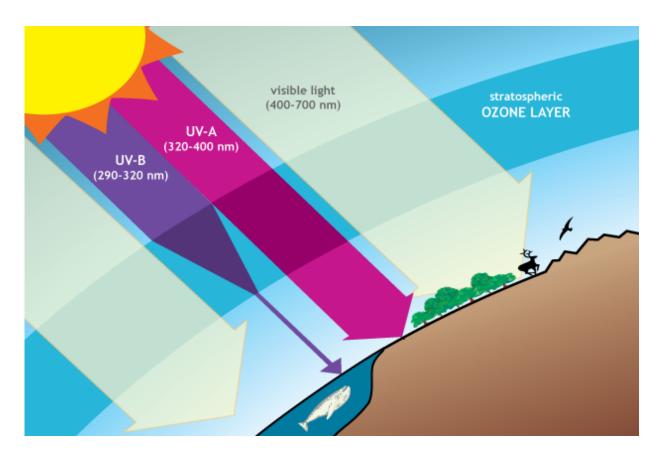
The ozone hole did not cause global warming

Because the ozone layer normally blocks ultraviolet (UV) light, an ozone hole allows more UV light than usual to reach the surface. However, the additional energy added to the Earth system from the ozone hole is so small that it couldn't be responsible for the warming trend that's been occurring.

How small? Well, the vast majority of sunlight is the light we can see—visible light with wavelengths of 400-700 nanometers. UV light is only about 8 percent of all sunlight to begin with, and the ozone layer and oxygen (both of which absorb UV) only permit a fraction of that to reach the surface. The additional amount of UV that the Antarctic ozone hole allows to reach the surface for a month or so each year is a small fraction of an already small amount of sunlight—too small to explain global warming.

That doesn't mean the ozone hole isn't important, however. UV light can cause sunburns, cataracts, genetic mutations, and cancer. It can damage land and ocean-based marine life, including the tiny <u>phytoplankton</u> that form the base of the ocean food web.





Ozone in the stratosphere provides protection from harmful ultraviolet (UV) radiation. The ozone layer is mostly transparent to incoming visible light and longer-wavelength UV, but it absorbs a large amount of damaging UV-B. Credit: NOAA Headquarters

The ozone hole has left a mark on Southern Hemisphere summer climate

When it's intact, the <u>ozone layer</u> absorbs UV light, which warms the stratosphere. When the ozone hole forms each spring, total ozone concentrations in the lower stratosphere dip. The annual dip has caused a strong springtime cooling trend in the lower polar stratosphere over the past few decades.



This springtime cooling amplifies the temperature contrast between the poles and the equator. Strong temperature contrasts often intensify atmospheric circulation, and models indicate that springtime cooling of the polar stratosphere due to <u>ozone loss</u> is largely responsible for changes in atmospheric circulation across the Southern Hemisphere.

These circulation changes have affected climate at the surface, primarily in the summer. The prevailing mid-latitude westerly winds have strengthened and shifted closer to the pole, drawing mid-latitude storm tracks—and associated rainfall—southward as well. More intense surface winds appear to have intensified the subtropical ocean gyres and increased the overturning circulation in the Southern Ocean. In subtropical latitudes, summertime precipitation has increased.

And even though the ozone hole can't explain global-scale temperature trends over the past few decades, models do suggest that changes in atmospheric circulation due to the ozone hole have contributed to seasonal surface temperature trends in a few Southern Hemisphere locations, including warming of high latitudes of the Southern Ocean in late winter, and summertime warming of the Antarctic Peninsula and cooling of the continental interior.

Provided by NOAA Headquarters

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