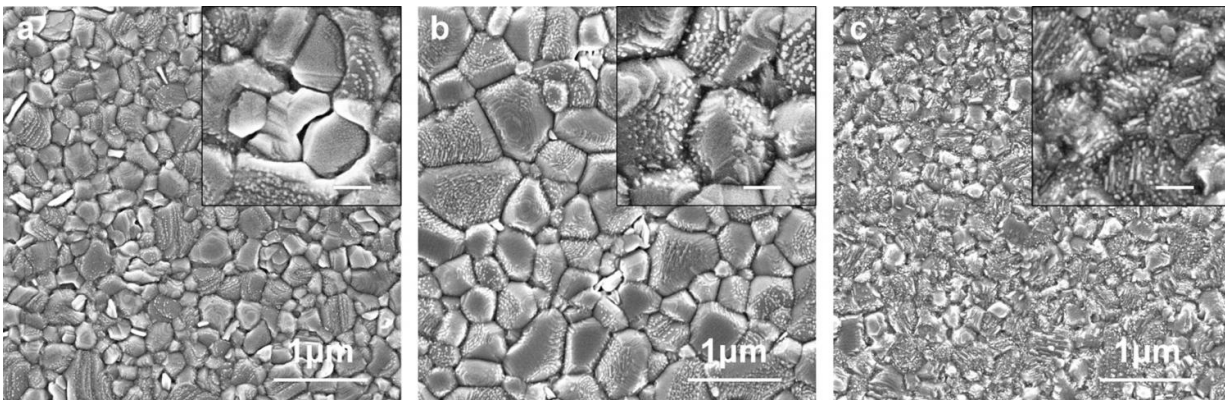


# New advances in solar cell technology

October 5 2016



A comparison of grain boundaries in MAPbI<sub>3</sub> perovskite films following thermal annealing (a), DMF solvent annealing (b), and methylamine post annealing treatment. The methylamine post annealing treatment shows the most improvement, as the grain boundaries become fused and less defined after application. Credit: Yan Jiang

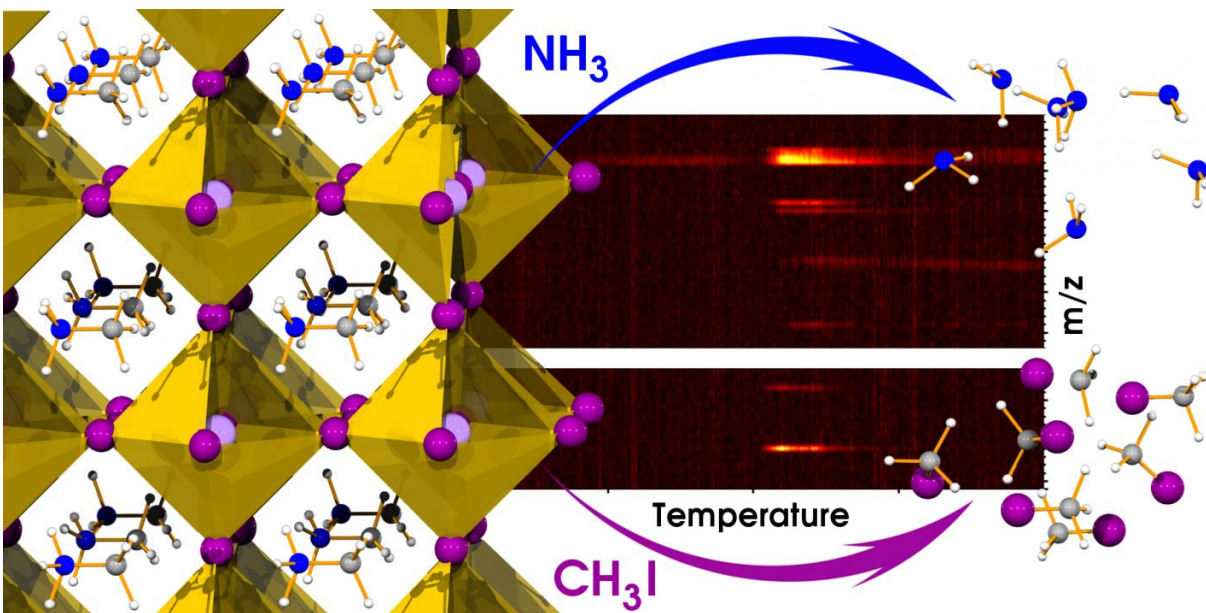
With the high environmental cost of conventional energy sources and the finite supply of fossil fuels, the importance of renewable energy sources has become much more apparent in recent years. However, efficiently harnessing solar energy for human use has been a difficult task. While silicon-based solar cells can be used to capture sunlight energy, they are costly to produce on an industrial scale. Research from the Energy Materials and Surface Sciences Unit at the Okinawa Institute of Science and Technology Graduate University (OIST), led by Prof. Yabing Qi, has focused on using organo-metal halide perovskite films in solar cells.

These perovskite films are highly crystalline materials that can be formed by a large number of different chemical combinations and can be deposited at low cost. Recent publications from Prof. Qi's lab cover three different areas of innovation in perovskite film research: a novel post annealing treatment to increase perovskite efficiency and stability, a discovery of the decomposition products of a specific perovskite, and a new means of producing perovskites that maintains solar efficiency when scaled up.

In order to be useful as solar cells, perovskite films must be able to harvest solar energy at a high efficiency that is cost-effective, be relatively easy to manufacture, and be able to withstand the outdoor environment over a long period of time. Dr. Yan Jiang in Prof. Qi's lab has recently published research in *Materials Horizons* that may help increase the solar efficiency of the organo-metal halide perovskite MAPbI<sub>3</sub>. He discovered that the use of a methylamine solution during post-annealing led to a decrease in problems associated with grain boundaries. Grain boundaries manifest as gaps between crystalline domains and can lead to unwanted charge recombination. This is a common occurrence in perovskite films and can reduce their efficiency, making the improvement of grain boundary issues essential to maintain high device performance. Dr. Jiang's novel post annealing treatment produced solar cells that had fused [grain boundaries](#), reduced charge recombination, and displayed an outstanding conversion efficiency of 18.4%. His treated perovskite films also exhibited exceptional stability and reproducibility, making his method useful for industrial production of solar cells.

One of the biggest disadvantages to the use of perovskites when compared to silicon in solar cells is their relatively short lifespan. In order to create a solar cell that can withstand the outdoor environment over a long period of time, it is crucial to determine the major products of perovskite decomposition. Previous research on MAPbI<sub>3</sub> perovskite

films led to the conclusion that the gas products of thermal degradation of this material were methylamine ( $\text{CH}_3\text{NH}_2$ ) and hydrogen iodide ( $\text{HI}$ ). However, exciting new research from Dr. Emilio J. Juarez-Perez, also in Prof. Qi's lab, published in *Energy & Environmental Science*, shows that major gas products of degradation are methyl iodide ( $\text{CH}_3\text{I}$ ) and ammonia ( $\text{NH}_3$ ) instead. Dr. Juarez-Perez used a combination of thermal gravimetric differential thermal analysis (TG-DTA) and mass spectrometry (MS) to correctly determine both the mass loss and chemical nature of these products. Because the products of decomposition have now been correctly identified, researchers can look for ways to prevent degradation of the material, leading to more stable materials for use in the future.



MAPbI<sub>3</sub> perovskite films decompose to form methyl iodide ( $\text{CH}_3\text{I}$ ) and ammonia ( $\text{NH}_3$ ), determined by thermal gravimetric differential thermal analysis and mass spectrometry. Credit: Emilio J. Juarez-Perez

A pervasive problem in academic research is often the inability to scale up experiments for use in industry. While perovskite films can be made with relative ease on a small scale in the laboratory, they can be difficult to replicate on the large scale needed for mass production. New research from Dr. Matthew Leyden in the *Journal of Materials Chemistry A* has the potential to make industrial production of perovskites much easier. His work uses chemical vapor deposition, a cost-effective process commonly used in industry, to create large solar cells and modules of FAPbI<sub>3</sub> perovskites. This is one of the first demonstrations of perovskite solar cells and modules fabricated by a method widely employed in industry, making the mass production of perovskite films more feasible. The solar cells and modules produced are significantly larger, e.g., 12 cm<sup>2</sup>, than those commonly studied in academia, typically

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