

Making space rocket fuel from water could drive a power revolution on Earth

September 28 2016, by Charles W. Dunnill And Robert Phillips



Credit: NASA/Northrop Grumman/William Furlong

Researchers led by NASA's former chief technologist are hoping to launch a satellite carrying water as the source of its fuel. The team <u>from</u> <u>Cornell University</u>, guided by Mason Peck, want their device to become the first shoebox-sized "CubeSat" to orbit the moon, while demonstrating the potential of water as a source of spacecraft fuel. It's a safe, stable substance that's <u>relatively common</u> even in space, but could also find greater use here on Earth as we search for alternatives to fossil



fuels.

Until we develop a warp drive or some other futuristic propulsion system, space travel is likely to rely largely on the kind of propellantbased rockets we use today. These work by firing gas out of the rear of the vehicle in a way that, thanks to the <u>laws of physics</u>, pushes it forward. Such propulsion systems for satellites need to be lightweight and carry a lot of energy in a small space (high energy density) in order to continuously pack a powerful punch over the many years, or even decades, that the craft are in orbit.

Safety too is of prime concern. Packing energy into a small volume and mass in the form of a fuel means even the slightest issue can have disastrous consequences, as we saw with the recent SpaceX rocket explosion. Putting satellites in orbit with any form of unstable fuel on board could spell disaster for expensive hardware or even worse, human life.

Water is a way around this issue because it is essentially an energy carrier rather than a fuel. <u>The Cornell team</u> isn't planning to use water itself as a propellant but to rather use electricity from <u>solar panels</u> to split the water into hydrogen and oxygen and use them as the fuel. The two gasses, when recombined and ignited will burn or explode, giving out the energy that they took in during the splitting process. This combustion of gasses can be used to drive the satellite forward, gaining speed or altering its position in orbit of whichever desired planet or moon is the target.

Solar panels, with high reliability and no moving parts, are ideally suited to operate in zero gravity and in the extreme environments of space, producing current from sunlight and allowing the satellite to actively engage on its mission. Traditionally this energy is stored in batteries. But the Cornell scientists want to use it to create their fuel source by splitting



the on-board water.



Toyota's hydrogen fuel cell car 'Mirai'. Credit: Toyota

Extra-terrestrial electrolysis

The proposed process – known as electrolysis – involves running a current through a water sample usually containing some soluble electrolyte. This breaks down the water into oxygen and hydrogen, which are released separately at the two electrodes. On Earth, gravity would then be used to separate the gasses so they can be harvested and used. In the free-flowing zero gravity of space, however, the satellite has to use



centrifugal forces from rotation to separate the gases from the solution.

Electrolysis has been use in space before to provide oxygen supplies for manned space missions without the need for high-pressure oxygen storage tanks, for example on the <u>International Space Station</u>. But instead of sending water into space in heavy loads on rockets, we could also one day extract it from the moon or asteroids. If the novel approach of using both the hydrogen and oxygen for satellite fuel proves successful, we could have a ready source of it waiting for us in space. This means it could shape how we power at least some of the spacecraft of the future.

From satellites to cars

As is often the case, developments in space technology are pushing concepts that have the potential to help overcome significant energy problems here on Earth. Electricity is really difficult to store and, as we increase our renewable energy supplies, we need to buffer the supply and demand. Wind and solar farms are really inefficient forms of renewable energy, not because of problems with the generating technology but because we often cannot do anything useful with the energy that they produce. The electricity grid struggles at times of high production and low energy need.

The answer, as in outer space propulsion, could involve using surplus electricity to split <u>water</u> into hydrogen and oxygen. This produces a storable, transportable commodity in the form of hydrogen fuel. When the energy is needed, it can be released by recombining it with oxygen from the atmosphere. This can either be done in a fuel cell to produce electricity again, or by burning it in a combustion engine or a hydrogen gas burner.

Welsh start-up firm Riversimple – along with major car manufacturers



Toyota and Volkswagen – is already producing hydrogen fuel-cell cars. So if the hydrogen is produced from solar energy in the same manner as Cornell's satellite, this <u>space</u> technology could become part of your everyday life sooner than you think.

This article was originally published on <u>The Conversation</u>. *Read the* <u>original article</u>.

Source: The Conversation

Citation: Making space rocket fuel from water could drive a power revolution on Earth (2016, September 28) retrieved 12 May 2024 from <u>https://phys.org/news/2016-09-space-rocket-fuel-power-revolution.html</u>

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