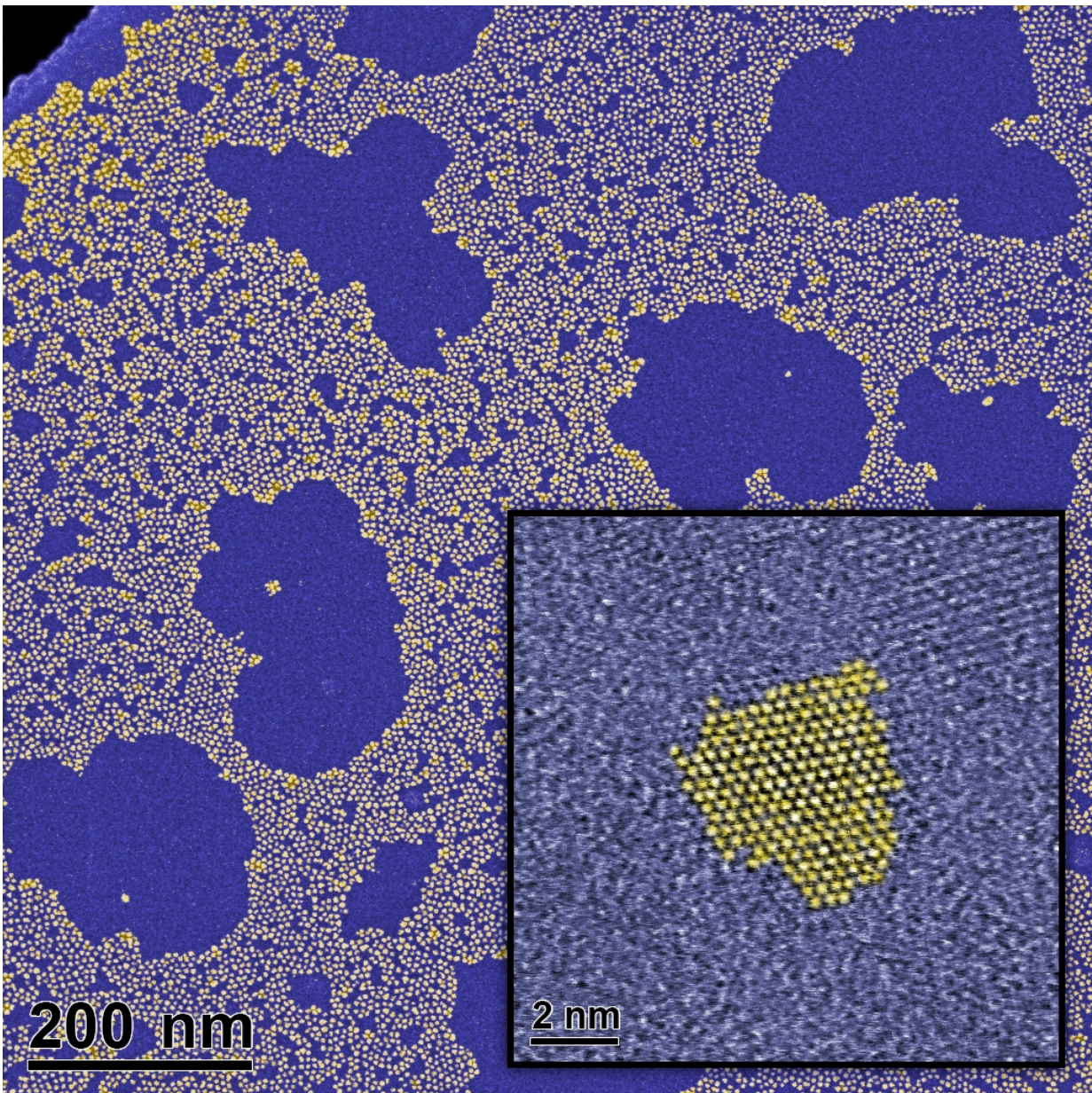


Quantum dots with impermeable shell: A powerful tool for nanoengineering

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Images of ZnO quantum dots prepared by the Institute of Physical Chemistry of the Polish Academy of Sciences in Warsaw, taken by transmission electron microscopy. False colors. Credit: IPC PAS

Unique optical features of quantum dots make them an attractive tool for many applications, from cutting-edge displays to medical imaging. Physical, chemical or biological properties of quantum dots must, however, be adapted for specific applications. Unfortunately, quantum dots prepared by chemical methods using copper-based click reactions destroy the ability of quantum dots to emit light. Russian scientists have shown, however, that zinc oxide (ZnO) quantum dots prepared by a new method, after modification by the click reaction via copper ions, fully retain their ability to emit light.

"Click reactions catalyzed by copper cations have long attracted the attention of chemists dealing with [quantum dots](#). The experimental results, however, were disappointing: After modification, the luminescence was so poor that they were just not fit for use. We were the first to demonstrate that it is possible to produce quantum dots from organometallic precursors while preserving their valuable optical properties after being subjected to copper-catalysed click reactions," says Prof. Janusz Lewinski (IPC PAS, FC WUT).

Quantum dots are crystalline structures with a few nanometers in size. As semiconductor materials, they exhibit a variety of interesting features typical of quantum objects, including absorbing and emitting radiation at a strictly defined energy. Since atoms interact with light in a similar way, quantum dots are often called artificial atoms. In some respects, however, quantum dots are more versatile than atoms. Optical properties of each dot actually depend on the size and the type of material from which they are formed. This means that quantum dots may be precisely

designed for specific applications.

To adapt them for specific applications, quantum dots must be tailored in terms of physico-chemical properties. For this purpose, chemical molecules with suitable characteristics are attached to their surface. Due to the simplicity, efficacy, and speed of the process, an exceptionally convenient method is the click reaction. Unfortunately, copper ion click reactions result in the almost complete quenching of the luminescence of the quantum dots.

"Failure is usually a result of the inadequate quality of quantum dots, which is determined by the synthesis method. Currently, ZnO dots are mainly produced by the sol-gel method from inorganic precursors. Quantum dots generated in this manner are coated with a heterogeneous and probably leaky protective shell, made of various sorts of chemical molecules. During a click reaction, the copper ions are in direct contact with the surface of quantum dots and quench the luminescence of the dot, which becomes completely useless," explains Dr. Agnieszka Grala (IPC PAS), the first author of the article in the *Chemical Communications* journal.



ZnO quantum dots prepared by the Institute of Physical Chemistry of the Polish Academy of Sciences in Warsaw are stabilized by impermeable protective „umbrellas” -- coatings built of organic compounds. Starring as shining quantum dots: Dr. Agnieszka Grala and Ph.D. student Malgorzata Wolska-Pietkiewicz. Credit: IPC PAS, Grzegorz Krzyzewski

For several years, Prof. Lewinski's team has been developing alternative methods for the preparation of high quality ZnO quantum dots. The method presented in this paper affords the quantum dots derived from organozinc precursors. Composition of the nanoparticles can be programmed at the stage of precursors preparation, which makes it possible to precisely control the character of their organic-inorganic interface.

"Nanoparticles produced by our method are crystalline and all have almost the same size. They are spherical and have characteristics of typical quantum dots. Every nanoparticle is stabilized by an impermeable protective jacket, built of organic compounds, strongly anchored on the surface of the semiconductor core. As a result, our quantum dots remain stable for a long time and do not aggregate—that is, clump together in solutions," says Malgorzata Wolska-Pietkiewicz, a PhD student at FC WUT.

"The key to success is producing a uniform stabilizing shell. Such coatings are characteristic of the ZnO quantum dots obtained by our method. The organic layer behaves as a tight protective umbrella protecting dots from direct influence of the [copper ions](#)," says Dr. Grala and clarifies: "We carried out [click reaction](#) known as alkyne-azide cycloaddition, in which we used a copper(I) compound as catalysts. After functionalization, our quantum dots shone as brightly as at the beginning."

Quantum dots have multiple applications in various industrial processes and as nanomarkers in biology and medicine, where they are combined with biologically active molecules. Nanoobjects functionalized in this manner are used to label both individual cells as well as whole tissues. The unique properties of quantum dots also enable long-term monitoring of the labeled item. Commonly used quantum dots, however, contain toxic heavy metals, including cadmium. In addition, they clump together

in solutions, which supports the thesis of the lack of tightness of their shells. Meanwhile, the ZnO dots produced by Prof. Lewinski's group are non-toxic, they do not aggregate, and can be bound to many chemical compounds, so they are more suitable for medical diagnosis and for imaging cells and tissues.

More information: Agnieszka Grala et al, 'Clickable' ZnO nanocrystals: the superiority of a novel organometallic approach over the inorganic sol–gel procedure, *Chem. Commun.* (2016). [DOI: 10.1039/C6CC01430E](https://doi.org/10.1039/C6CC01430E)

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