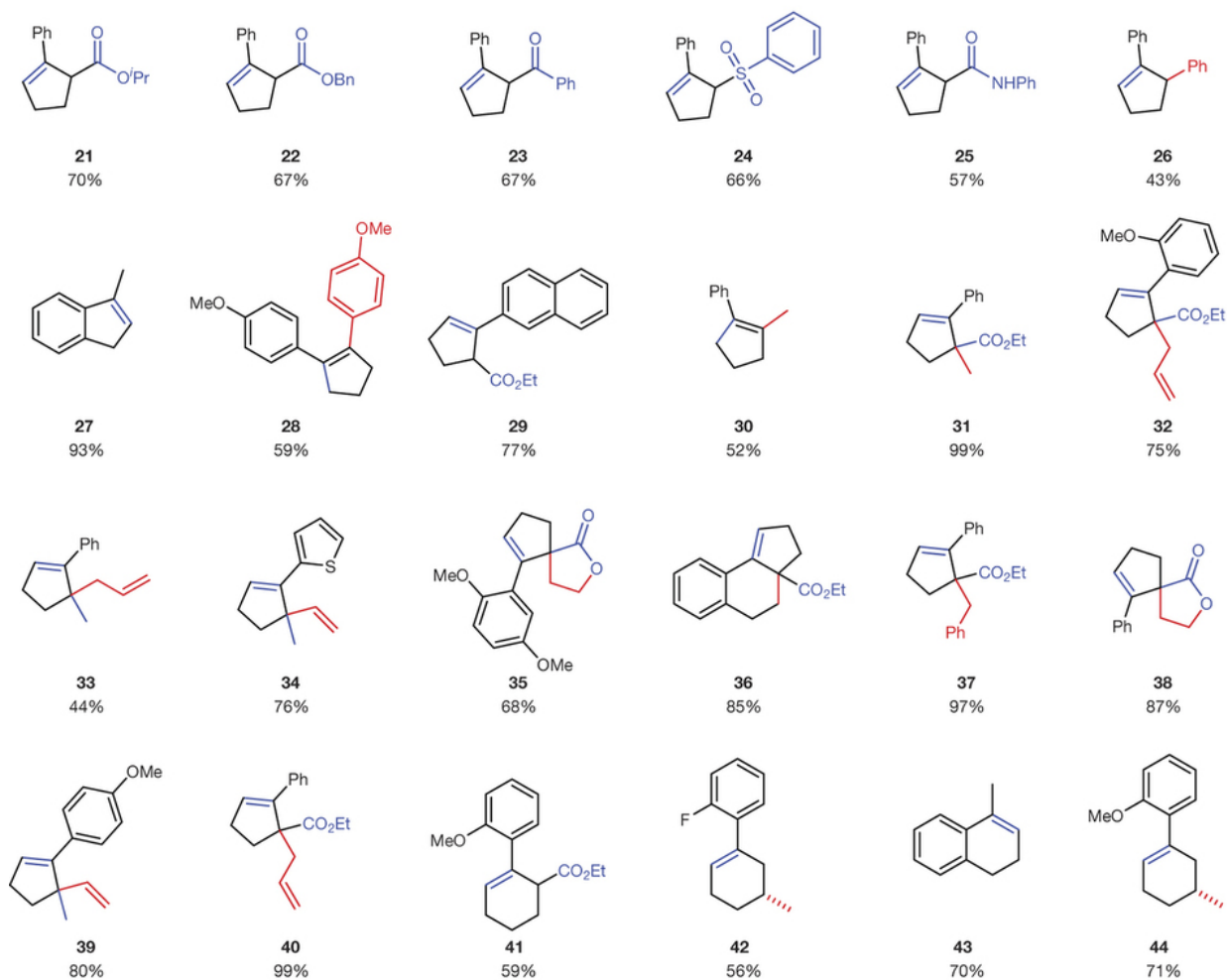
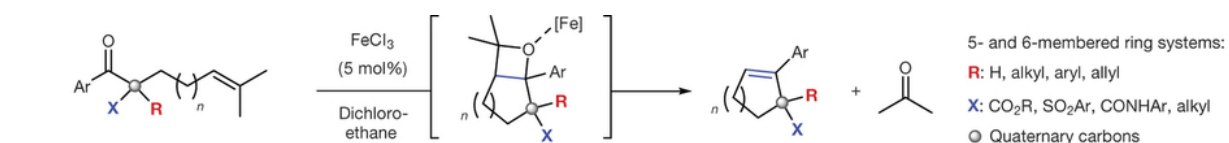


Safe and generalizable catalyst for carbonyl-olefin metathesis reaction

June 1 2016, by Heather Zeiger



Scope of the iron(III)-catalysed carbonyl–olefin metathesis reaction. Credit: (c) 2016 Jacob R. Ludwig, Paul M. Zimmerman, Joseph B. Gianino & Corinna S. Schindler, *Nature* 533, 374–379 (19 May 2016) doi:10.1038/nature17432

(Phys.org)—Olefin metathesis reactions where two allyls switch substituent groups, has proved to be a useful carbon-carbon bond forming reaction for drug discovery and other industrial processes. The carbonyl-olefin metathesis reaction was discovered around the same time as the olefin metathesis reaction, but has had limited applicability due to harsh reaction conditions and stringent reactant requirements. Until now, there have been no examples of catalysts for this reaction that are generalizable.

In a report in *Nature*, Jacob R. Ludwig, Paul M. Zimmerman, Joseph B. Gianino & Corinna S. Schindler of the University of Michigan have successfully found a Lewis acid catalyst that successfully effects a carbonyl-olefin ring-closing metathesis reaction for the direct synthesis of functionalized carbocycles. Furthermore, the catalyst uses iron, which is abundant and environmentally safe, and works under mild conditions for a variety of functional groups and at gram-level scales.

The olefin reaction mechanism involves a metal-catalyzed [2+2] cycloaddition of two allyls to form a four-member cyclic intermediate in which the metal is one of the members of the four-member ring. This is followed by a [2+2] cycloreversion resulting in products that have exchanged substituents. Carbonyl olefin metathesis reactions operate similarly. There are three carbonyl olefin metathesis reactions that have been observed, but each has its limitations. One involves a two-step photochemical cycloaddition followed by a thermolysis cycloreversion. Another involves a stoichiometric amount of transition-metal catalyst. Another reaction involves an organocatalyst that requires a cyclopropene

reactant. All of these are limited in scope and, in the case of the photochemical reaction, require prohibitively harsh conditions. The reaction involving a stoichiometric amount of transition-metal catalyst is the most versatile, but it produces a metal-oxo complex that consumes the catalyst.

Ludwing, et al. wanted to look for ways to form an *in situ* oxetane intermediate, a four-member cyclic intermediate with the oxygen as one of the atoms instead of a metal. This would avoid the formation of products that consume the catalyst. In general, they needed a catalyst that promotes [2+2] cycloaddition and [2+2] cycloreversion, does not promote side reactions such as polymerization or reforming reactants, and is chemoselective. They decided to investigate metal-derived Lewis acids.

To test various Lewis acid catalysts, they reacted a β -ketoester with a pendant isoprenyl group with MCl_x Lewis acids. They tested $AlCl_3$, $TiCl_4$, $SnCl_4$, $FeCl_2$, $ZnCl_2$, $SnCl_2$, and $FeCl_3$. They found that $FeCl_3$ formed the desired products and that optimal reaction conditions were 5 mol% $FeCl_3$ in dichloroethane at room temperature for one-to-twelve hours.

Once a catalyst was determined and reaction conditions were optimized, the next step was to test the generalizability of the reaction. Ludwig, et al. looked at various arene groups. Their reaction conditions worked with electron withdrawing and donating groups producing products in good yield. They then changed the β -substitution on the β -ketoesters and found the reaction worked with a variety of substituents. They also looked at the effects of incorporating a heteroatom, changing structural motifs, and changing the pendant isoprenyl group. Overall, the $FeCl_3$ catalyst showed remarkable versatility in the variety of reactions that were successfully completed.

These reactions provided insight into the carbonyl-olefin reaction

mechanism. The reaction either proceeds via a carbocation intermediate or in a concerted fashion that involves metal coordination to FeCl_3 . Subsequent studies using a mechanistic probe to trap the product that would be formed from a carbocation intermediate showed no results, thus demonstrating that a concerted and metal coordinated mechanism is likely. Computational studies indicate that the reaction mechanism is likely a concerted carbon-oxygen bond breakage and carbon-carbon bond formation mechanism that occurs asynchronously.

Carbonyl-[olefin metathesis](#) reactions had been limited in scope, but this study demonstrates a [catalyst](#) and optimal reaction conditions that are generalizable for a variety of reactants opening the door for the synthesis of complex compounds. The [reaction](#) is practically useful as iron is an earth-abundant metal and is environmentally safe, and the reactions are scalable.

More information: Jacob R. Ludwig et al. Iron(III)-catalysed carbonyl–olefin metathesis, *Nature* (2016). [DOI: 10.1038/nature17432](https://doi.org/10.1038/nature17432)

Abstract

The olefin metathesis reaction of two unsaturated substrates is one of the most powerful carbon–carbon-bond-forming reactions in organic chemistry. Specifically, the catalytic olefin metathesis reaction has led to profound developments in the synthesis of molecules relevant to the petroleum, materials, agricultural and pharmaceutical industries¹. These reactions are characterized by their use of discrete metal alkylidene catalysts that operate via a well-established mechanism². While the corresponding carbonyl–olefin metathesis reaction can also be used to construct carbon–carbon bonds, currently available methods are scarce and severely hampered by either harsh reaction conditions or the required use of stoichiometric transition metals as reagents. To date, no general protocol for catalytic carbonyl–olefin metathesis has been reported. Here we demonstrate a catalytic carbonyl–olefin ring-closing

metathesis reaction that uses iron, an Earth-abundant and environmentally benign transition metal, as a catalyst. This transformation accommodates a variety of substrates and is distinguished by its operational simplicity, mild reaction conditions, high functional-group tolerance, and amenability to gram-scale synthesis. We anticipate that these characteristics, coupled with the efficiency of this reaction, will allow for further advances in areas that have historically been enhanced by olefin metathesis.

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Citation: Safe and generalizable catalyst for carbonyl-olefin metathesis reaction (2016, June 1)
retrieved 12 September 2024 from

<https://phys.org/news/2016-06-safe-generalizable-catalyst-carbonyl-olefin-metathesis.html>

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