

'Forbidden' substances on super-Earths

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Using mathematical models, scientists have simulated the interiors of super-Earths and discovered that they may contain compounds that are forbidden by the classical rules of chemistry—these substances may increase the heat transfer rate and strengthen the magnetic fields on these planets. The findings have been presented in a paper published in the journal *Scientific Reports*.

In a previous study, the authors of the paper used an algorithm called USPEX to identify new compounds of sodium and chlorine, as well as other exotic substances.

In their latest paper, the researchers attempted to find out which compounds may be formed by silicon, oxygen, and <u>magnesium</u> at high pressures. These particular elements were not chosen randomly.

"Earth-like <u>planets</u> consist of a thin silicate crust, a silicate-oxide mantle that makes up approximately seven-eighths of the Earth's volume and consists of more than 90 percent of silicates and magnesium oxide—and an iron core. We can say that magnesium, oxygen, and silicon form the basis of chemistry on Earth and on Earth-like planets," says Oganov.

Using the USPEX algorithm, the researchers investigated various structural compositions of Mg-Si-O that may occur at pressures ranging from 5 to 30 million atmospheres. Such pressures may exist in the interiors of super-Earths—planets with a solid surface mass several times greater than the mass of the Earth. There are no planets like this in the solar system, but astronomers know of planets orbiting other stars



that are not as heavy as the gas giants, but are considerably heavier than the Earth. They are called super-Earths. These planets include the recently discovered Gliese 832c, which is five times heavier than the Earth, or the mega-Earth Kepler-10c, which is 17 times heavier than the Earth.

The results of the computer modelling show that the interiors of these planets may contain the exotic compounds $MgSi_3O_{12}$ and $MgSiO_6$. They have many more oxygen atoms than the $MgSiO_3$ on Earth.

In addition, $MgSi_3O_{12}$ is a metal oxide and a conductor, whereas other substances consisting of Mg-Si-O atoms are dielectrics or semiconductors."Their properties are very different to normal compounds of magnesium, oxygen, and silicon—many of them are metals or semiconductors. This is important for generating magnetic fields on these planets. As magnetic fields produce electrical currents in the interiors of a planet, high conductivity could mean a significantly more powerful magnetic field," explains Oganov.

A more powerful <u>magnetic field</u> means more powerful protection from cosmic radiation, and consequently more favourable conditions for living organisms. The researchers also predicted new magnesium and silicon oxides that do not fit in with the rules of classical chemistry—SiO, SiO₃, and MgO₃, in addition to the oxides MgO₂ and Mg3O₂ previously predicted by Oganov at lower pressures.

The computer model also enabled the researchers to determine the decomposition reactions that MgSiO₃ undergoes at the ultra-high pressures on super-Earths—post-perovskite.

"This affects the boundaries of the layers in the mantle and their dynamics. For example, an exothermic phase change speeds up the convection of the mantle and the <u>heat transfer</u> within the planet, and an



endothermic phase change slows them down. This means that the speed of motion of lithospheric plates on the planet may be higher," says Oganov.

Convection, which determines plate tectonics and the mixing of the mantle, can either be faster (speeding up the mixing of the mantle and heat transfer) or slower. In endothermic change, a possible scenario could be the disintegration of a planet into several independently convecting layers, he noted.

The fact that the Earth's continents are in constant motion, floating on the surface of the mantle, is what gives volcanism and a breathable atmosphere. If continental drift were to stop, it could have disastrous consequences for the climate.

More information: Haiyang Niu et al. Prediction of novel stable compounds in the Mg-Si-O system under exoplanet pressures, *Scientific Reports* (2015). DOI: 10.1038/srep18347

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