

New explosives provide enhanced safety, high energy

October 23 2015, by Kevin Roark



A small amount of explosive material is subjected to shock wave with a striker plate to induce detonation. An explosive material's insensitivity — its resistance to accidental or unintended detonation — is one set of data measured using this type of test. Credit: LANL photo by Daniel Preston



Los Alamos National Laboratory explosives chemist David Chavez has synthesized a pair of novel molecules, one possessing a unique fused three-ring structure. These materials could usher in a new class of explosives that provide high-energy output with enhanced safety.

"There is a general trend that the higher the performance of an energetic material, the more sensitive the material is to insults such as impact, spark and friction," said Chavez. "The Laboratory is interested in creating and understanding the science of 'outlier' <u>materials</u> which do not follow this trend."

The research is published as the cover article in today's issue of the prestigious German applied chemistry journal *Angewandte Chemie International Edition*.

In the article Chavez, of the High Explosives Science and Technology group, details two materials never before synthesized, the first, a dinitro-diamino compound, displays performance approaching that of conventional high performance materials, but is insensitive to impact spark and friction, making it very safe to handle. The second molecule, a new heterocyclic ring system called a tricyclic 1,2,3,4-tetrazine, is more like a conventional energetic in performance and safety, and forms the foundation of the synthesis of future materials of similar structure.

"The materials could have applications where almost any kind of energetic is needed," said Chavez. "These could replace explosives and propellants that can't meet today's safety requirements. Over time, explosives safety is becoming more and more important, particularly in propulsion, some military applications and in the mining industry. It's also highly desirable to find a energetic materials that require less quantity, maintains high performance, are insensitive and aren't adversely affected by large temperature changes."





Explosives chemist David Chavez weighs a small amount of tetrazine, an explosives precursor. Chavez has synthesized two new explosives molecules that promise high-energy with enhanced safety.

Chavez worked for more that two years on his idea of a unique molecular arrangement of the Carbon, Hydrogen, Nitrogen and Oxygen elements that are the basic constituents of all explosives. The insensitive nature of the dintro-diamino compound comes in part from extensive hydrogen bonding among the molecules. This effect is similar to the glue of a "post-it" note. The glue (hydrogen-bonds) is strong enough to keep the molecules intact, but weak enough that the molecules don't break down – initiating detonation – when the molecules are pulled apart during destructive stimuli, enhancing safety.



"There were a lot of different ideas about how to isolate and confirm the right structure," said Chavez, "learning along the way from every experiment that doesn't work, then eventually having success. One of the great things about Los Alamos is having the ability to push the frontiers, to do the new thing that no one's been able to do before."

More information: David E. Chavez et al. Synthesis and Thermal Behavior of a Fused, Tricyclic 1,2,3,4-Tetrazine Ring System, *Angewandte Chemie International Edition* (2015). DOI: 10.1002/anie.201506744

Provided by Los Alamos National Laboratory

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