

Scientists invent new approach in quest for organic solar panels and flexible electronics

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University of Vermont scientists have invented a new way to create what they are calling an electron superhighway in an organic semiconductor that promises to allow electrons to flow faster and farther -- aiding the hunt for flexible electronics, organic solar cells, and other low-cost alternatives to silicon. To explore these organic materials, UVM graduate students (from left) Naveen Rawat and Lane Manning, and professors Randy Headrick and Madalina Furis, deployed this table-top scanning laser microscope. Their latest finding is reported in the journal *Nature Communications* -- and may, someday not too far



off, let you roll up your computer like a piece of paper. Credit: Joshua Brown, UVM

TV screens that roll up. Roofing tiles that double as solar panels. Sunpowered cell phone chargers woven into the fabric of backpacks. A new generation of organic semiconductors may allow these kinds of flexible electronics to be manufactured at low cost, says University of Vermont physicist and materials scientist Madalina Furis.

But the basic science of how to get electrons to move quickly and easily in these organic materials remains murky.

To help, Furis and a team of UVM materials scientists have invented a new way to create what they are calling "an electron superhighway" in one of these materials—a low-cost blue dye called phthalocyanine—that promises to allow electrons to flow faster and farther in organic semiconductors.

Their discovery, reported Sept. 14 in the journal *Nature Communications*, will aid in the hunt for alternatives to traditional siliconbased electronics.

Hills and potholes

Many of these types of flexible electronic devices will rely on thin films of organic materials that catch sunlight and convert the light into electric current using excited states in the material called "excitons." Roughly speaking, an exciton is a displaced electron bound together with the hole it left behind. Increasing the distance these excitons can diffuse—before they reach a juncture where they're broken apart to produce electrical current—is essential to improving the efficiency of organic



semiconductors.

Using a new imaging technique, the UVM team was able to observe nanoscale defects and boundaries in the crystal grains in the thin films of phthalocyanine—roadblocks in the electron highway. "We have discovered that we have hills that electrons have to go over and potholes that they need to avoid," Furis explains.

To find these defects, the UVM team—with support from the National Science Foundation—built a scanning laser microscope, "as big as a table" Furis says. The instrument combines a specialized form of linearly polarized light and photoluminescence to optically probe the molecular structure of the phthalocyanine crystals.

"Marrying these two techniques together is new; it's never been reported anywhere," says Lane Manning '08 a doctoral student in Furis' lab and coauthor on the new study.

The new technique allows the scientists a deeper understanding of how the arrangement of molecules and the boundaries in the crystals influence the movement of excitons. It's these boundaries that form a "barrier for exciton diffusion," the team writes.

And then, with this enhanced view, "this energy barrier can be entirely eliminated," the team writes. The trick: very carefully controlling how the thin films are deposited. Using a novel "pen-writing" technique with a hollow capillary, the team worked in the lab of UVM physics and materials science professor Randy Headrick to successfully form films with jumbo-sized crystal grains and "small angle boundaries." Think of these as easy-on ramps onto a highway—instead of an awkward stop sign at the top of a hill—that allow excitons to move far and fast.

Better solar cells



Though the *Nature Communications* study focused on just one <u>organic</u> <u>material</u>, phthalocyanine, the new research provides a powerful way to explore many other types of organic materials, too—with particular promise for improved solar cells. A recent U.S. Department of Energy report identified one of the fundamental bottlenecks to improved solar power technologies as "determining the mechanisms by which the absorbed energy (exciton) migrates through the system prior to splitting into charges that are converted to electricity."

The new UVM study—led by two of Furis' students, Zhenwen Pan G'12, and Naveen Rawat G'15—opens a window to view how increasing "long-range order" in the organic semiconductor films is a key mechanism that allows excitons to migrate farther. "The molecules are stacked like dishes in a dish rack," Furis explains, "these stacked molecules—this dish rack—is the electron superhighway."

Though excitons are neutrally charged—and can't be pushed by voltage like the electrons flowing in a light bulb—they can, in a sense, bounce from one of these tightly stacked molecules to the next. This allows organic <u>thin films</u> to carry energy along this molecular highway with relative ease, though no net electrical charge is transported.

"One of today's big challenges is how to make better photovoltaics and solar technologies," says Furis, who directs UVM's program in materials science, "and to do that we need a deeper understanding of exciton diffusion. That's what this research is about."

More information: Polarization-resolved spectroscopy imaging of grain boundaries and optical excitations in crystalline organic thin films, *Nature Communications*, 6, Article number: 8201 DOI: 10.1038/ncomms9201



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