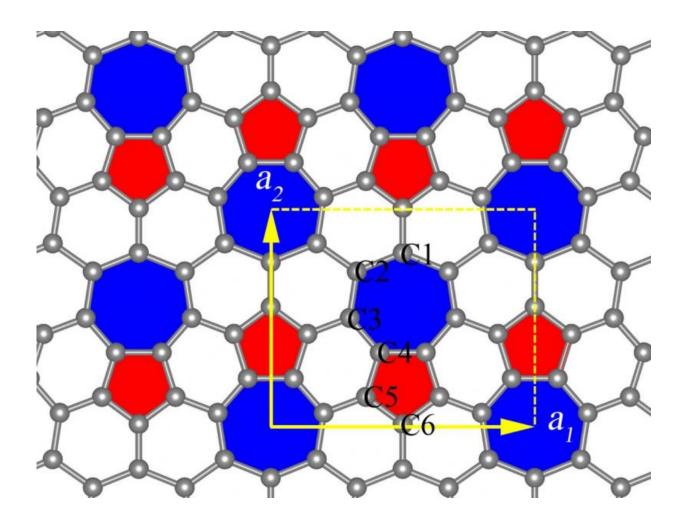


## Phagraphene, a relative of graphene, discovered

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A phagraphene structure. Credit: Artyom Oganov

A group of scientists from Russia, the USA and China have predicted



the existence of a new two-dimensional carbon material via computer generated simulation, a "patchwork" analogue of graphene called phagraphene. The results of their investigation were recently published in the journal *Nano Letters*.

"Unlike graphene, a hexagonal honeycomb structure with atoms of carbon at its junctions, phagraphene consists of penta-, hexa- and heptagonal carbon rings. Its name comes from a contraction of Penta-Hexa-heptA-graphene," says Oganov, head of the MIPT Laboratory of Computer Design.

Two-dimensional materials, composed of a one-atom-thick layer, have attracted great attention from scientists in the last few decades. The first of these materials, graphene, was discovered in 2004 by two MIPT graduates, Andre Geim and Konstantin Novoselov. In 2010, Geim and Novoselov were awarded the Nobel Prize in physics for that achievement.

Due to its two-dimensional structure, graphene has absolutely unique properties. Most materials can transmit electric current when unbound <u>electrons</u> have an energy that corresponds to the conduction band of the material. When there is a gap between the range of possible electron energies, the valence band, and the range of conductivity (the so-called forbidden zone), the material acts as an insulator. When the <u>valence band</u> and <u>conduction band</u> overlap, it acts a conductor, and electrons can move under the influence of electric field.

In graphene, each carbon atom has three electrons that are bound to electrons in neighboring atoms, forming chemical bonds. The fourth electron of each atom is "delocalized" throughout the whole <u>graphene</u> <u>sheet</u>, which allows it to conduct electrical current. At the same time, the forbidden zone in the graphene has zero width. If you plot the electron energy and their location in graph form, you get a figure resembling an



hour glass, i.e. two cones connected by vertices. These are known as Dirac cones.

Due to this unique condition, electrons in graphene behave very strangely: all of them have one and the same <u>velocity</u> (which is comparable to the velocity of light), and they possess no inertia. They appear to have no mass. And, according to the theory of relativity, particles traveling at the velocity of light must behave in this manner. The velocity of electrons in graphene is about 10 thousand kilometers a second (electron velocities in a typical conductor vary from centimeters up to hundreds of meters per second).

Phagraphene, discovered by Oganov and his colleagues through the use of the USPEX algorithm, as well as graphene, is a material in which Dirac cones appear, and electrons behave similarly to particles without mass.

"In phagraphene, due to the different number of atoms in the rings, the Dirac cones are 'inclined.' That is why the velocity of electrons in it depends on the direction. This is not the case in graphene. It would be very interesting for future practical use to see where it will be useful to vary the electron velocity," Artyom Oganov explains.

Phagraphene possesses all the other properties of <u>graphene</u> that allows it to be considered an advanced material for flexible electronic devices, transistors, solar batteries, display units and many other things.

**More information:** "Phagraphene: A Low-Energy Graphene Allotrope Composed of 5–6–7 Carbon Rings with Distorted Dirac Cones," *Nano Lett.*, Article ASAP. <u>DOI: 10.1021/acs.nanolett.5b02512</u>



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