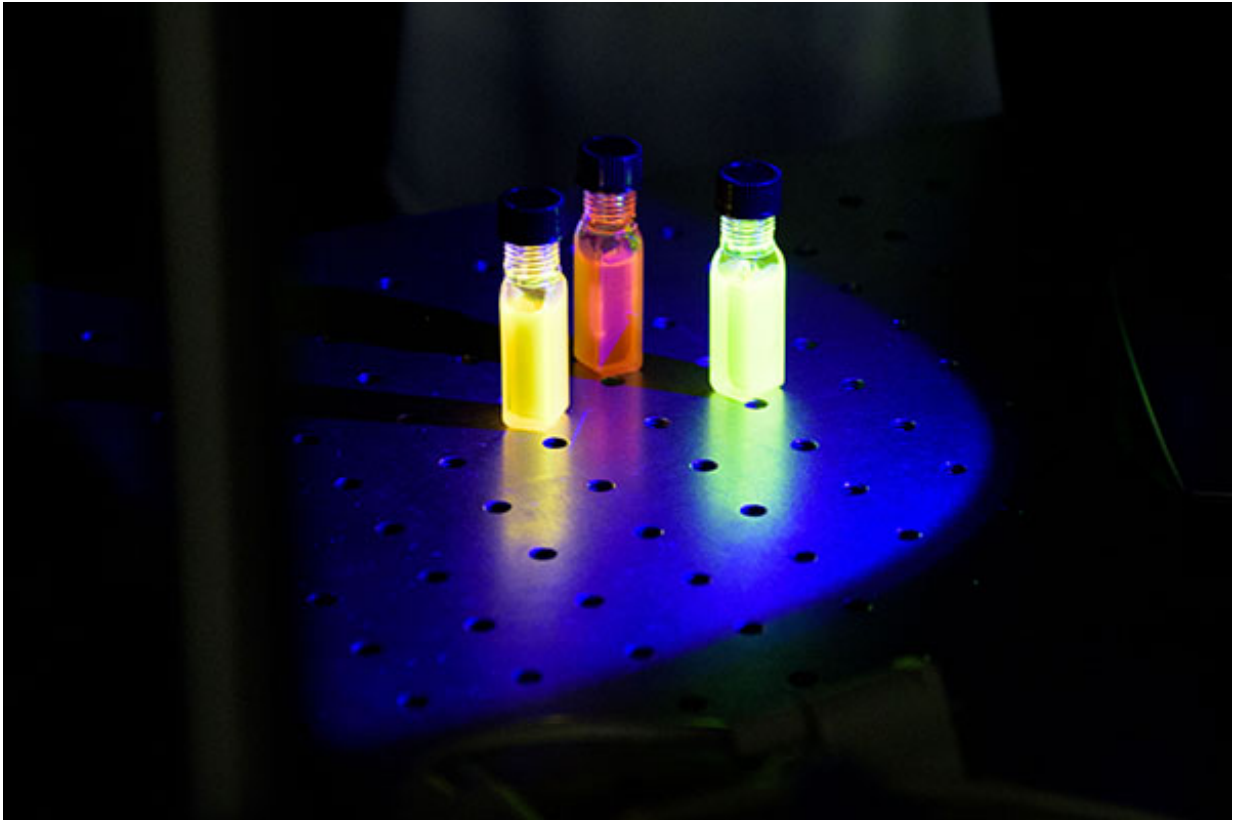


# Changing the color of light

July 23 2015, by Tracey Bryant

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A University of Delaware research team aims to to change the color of light, by developing new nanostructures that act like a ratchet -- combining the energy of two red photons of light into a single blue photon, which has higher energy. Such an advance could improve solar cell efficiency to chemotherapy treatments.

Credit: Evan Krape/University of Delaware

Researchers at the University of Delaware have received a \$1 million

grant from the W.M. Keck Foundation to explore a new idea that could improve solar cells, medical imaging and even cancer treatments. Simply put, they want to change the color of light.

They won't be tinkering with what you see out your window: no purple days or chartreuse nights, no edits to rainbows and blazing sunsets. Their goal is to turn low-energy colors of light, such as red, into higher-energy colors, like blue or green.

Changing the color of light would give solar technology a considerable boost. A traditional solar cell can only absorb light with energy above a certain threshold. Infrared light passes right through, its energy untapped.

However, if that low-energy light could be transformed into higher-energy light, a solar cell could absorb much more of the sun's clean, free, abundant energy. The team predicts that their novel approach could increase the efficiency of commercial solar cells by 25 to 30 percent.

The research team, based in UD's College of Engineering, is led by Matthew Doty, associate professor of materials science and engineering and associate director of UD's Nanofabrication Facility. Doty's co-investigators include Joshua Zide, Diane Sellers and Chris Kloxin, all in the Department of Materials Science and Engineering; and Emily Day and John Slater, both in the Department of Biomedical Engineering.

"This prestigious \$1 million grant from the Keck Foundation underscores the excellence and innovation of our University of Delaware faculty," says Nancy Targett, acting president of the University. "Clearly, the University of Delaware is pursuing big ideas in renewable energy and biomedicine with the potential to benefit the world."

"The University's Delaware Will Shine strategic plan challenges us to

think boldly as we seek solutions to problems facing society," Domenico Grasso, UD's provost, adds. "We congratulate the research team in the College of Engineering for this major award, and we look forward to their findings."

## **Changing the color of light**

"A ray of light contains millions and millions of individual units of light called photons," says project leader Matthew Doty. "The energy of each photon is directly related to the color of the light—a photon of red light has less energy than a photon of blue light. You can't simply turn a red photon into a blue one, but you can combine the energy from two or more red photons to make one blue photon."

This process, called "photon upconversion," isn't new, Doty says. However, the UD team's approach to it is.

They want to design a new kind of semiconductor nanostructure that will act like a ratchet. It will absorb two red photons, one after the other, to push an electron into an excited state when it can emit a single high-energy (blue) photon.

These nanostructures will be so teeny they can only be viewed when magnified a million times under a high-powered electron microscope.

"Think of the electrons in this structure as if they were at a water park," Doty says. "The first red photon has only enough energy to push an electron half-way up the ladder of the water slide. The second red photon pushes it the rest of the way up. Then the electron goes down the slide, releasing all of that energy in a single process, with the emission of the blue photon. The trick is to make sure the electron doesn't slip down the ladder before the second photon arrives. The semiconductor ratchet structure is how we trap the electron in the middle of the ladder until the

second photon arrives to push it the rest of the way up."

The UD team will develop new semiconductor structures containing multiple layers of different materials, such as aluminum arsenide and gallium bismuth arsenide, each only a few nanometers thick. This "tailored landscape" will control the flow of electrons into states with varying potential energy, turning once-wasted photons into useful energy.

The UD team has shown theoretically that their semiconductors could reach an upconversion efficiency of 86 percent, which would be a vast improvement over the 36 percent efficiency demonstrated by today's best materials. What's more, Doty says, the amount of light absorbed and energy emitted by the structures could be customized for a variety of applications, from lightbulbs to laser-guided surgery.

How do you even begin to make structures so tiny they can only be seen with an electron microscope? In one technique the UD team will use, called molecular beam epitaxy, nanostructures will be built by depositing layers of atoms one at a time. Each structure will be tested to see how well it absorbs and emits light, and the results will be used to tailor the structure to improve performance.

The researchers also will develop a milk-like solution filled with millions of identical individual nanoparticles, each one containing multiple layers of different materials. The [multiple layers](#) of this structure, like multiple candy shells in an M&M, will implement the photon ratchet idea.

Through such work, the team envisions a future upconversion "paint" that could be easily applied to [solar cells](#), windows and other commercial products.

## **Improving medical tests and treatments**

While the initial focus of the three-year project will be on improving solar energy harvesting, the team also will explore biomedical applications.

A number of diagnostic tests and medical treatments, ranging from CT and PET scans to chemotherapy, rely on the release of fluorescent dyes and pharmaceutical drugs. Ideally, such payloads are delivered both at specific disease sites and at specific times, but this is hard to control in practice.

The UD team aims to develop an upconversion nanoparticle that can be triggered by light to release its payload. The goal is to achieve the controlled release of drug therapies even deep within diseased human tissue while reducing the peripheral damage to normal tissue by minimizing the laser power required.

"This is high-risk, high-reward research," Doty says. "High-risk because we don't yet have proof-of-concept data. High-reward because it has such a huge potential impact in renewable energy to medicine. It's amazing to think that this same technology could be used to harvest more solar [energy](#) and to treat cancer. We're excited to get started!"

Provided by University of Delaware

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