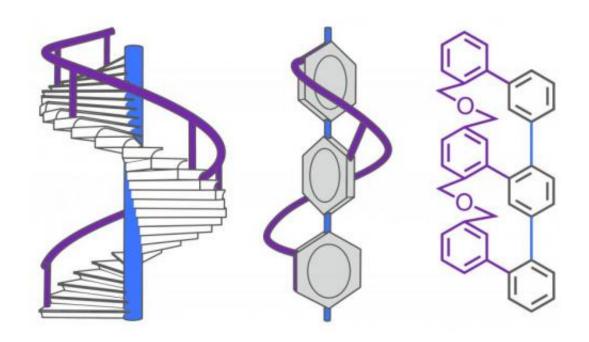


A twisted world: Chemists build a molecular banister

November 12 2014



Based on the strands of different lengths (blue and gray), the new helical molecule (right) adopts a spatial arrangement (schematic diagram in the center) that resembles the banister of a spiral staircase (left). Credit: University of Basel, Department of Chemistry

Chemists at the University of Basel in Switzerland have succeeded in twisting a molecule by combining molecular strands of differing lengths. The longer strand winds around a central axis like a staircase banister, creating a helical structure that exhibits special physical properties. The results were published in the renowned scientific journal *Angewandte*



Chemie International Edition.

The chemistry of all substances is to a large extent defined by their spatial arrangement. Many molecules can be present in two forms (enantiomers), which behave like a person's right and left hand. In particular, the organism makes a highly specific distinction between left-and right-handed molecules - a substance can, for example, be extremely active as a drug in one form, while its mirror image is entirely inert. The fundamental understanding of this "chirality", as it is called, has long been a central component of research in the field of chemistry.

Connecting strands of different sizes

The researchers headed by Professor Marcel Mayor in the Department of Chemistry at the University of Basel have developed a new approach to contort a small molecule into a form similar in appearance to the banister on a spiral staircase. At the molecular level, the interlinking of two oligomer strands with different lengths forces to the longer strand to wind around the shorter on its own to balance out the discrepancy in length. This creates a helix with a uniform twisting direction. As a consequence the entire molecule becomes chiral. The researchers were also able to demonstrate that it is possible to dynamically change the form of the helical molecule from left-handed to right and back again in just a few hours.

"It is not just the structural elegance of this molecule which makes it so unique," says Mayor. "Above all, it is a completely new way of constructing a continuous helix."

Efficient procedures for creating chiral compounds generate much interest in basic research and the industrial sector - they can, for example, be used in biological systems research, crop protection chemistry, and the pharmaceutical and fragrance industries. The project



was financially supported by the Swiss National Science Foundation.

More information: Rickhaus, M., Bannwart, L. M., Neuburger, M., Gsellinger, H., Zimmermann, K., Häussinger, D. and Mayor, M. Inducing Axial Chirality in a "Geländer" Oligomer by Length Mismatch of the Oligomer Strands, *Angewandte Chemie International Edition* (2014) | DOI: 10.1002/anie.201408424

Provided by University of Basel

Citation: A twisted world: Chemists build a molecular banister (2014, November 12) retrieved 18 April 2024 from https://phys.org/news/2014-11-world-chemists-molecular-banister.html

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