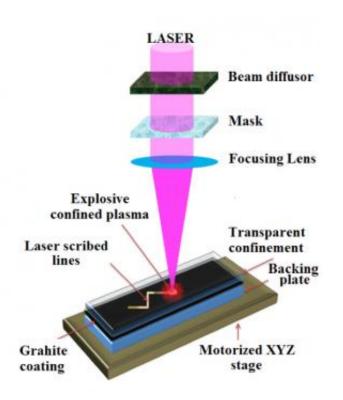


## **'Direct writing' of diamond patterns from graphite a potential technological leap**

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This illustration depicts a new technique that uses a pulsing laser to create synthetic nanodiamond films and patterns from graphite, with potential applications from biosensors to computer chips. Credit: Purdue University image/Gary Cheng

(Phys.org) —What began as research into a method to strengthen metals has led to the discovery of a new technique that uses a pulsing laser to create synthetic nanodiamond films and patterns from graphite, with



potential applications from biosensors to computer chips.

"The biggest advantage is that you can selectively deposit nanodiamond on rigid surfaces without the high temperatures and pressures normally needed to produce <u>synthetic diamond</u>," said Gary Cheng, an associate professor of industrial engineering at Purdue University. "We do this at room temperature and without a high temperature and pressure chamber, so this process could significantly lower the cost of making diamond. In addition, we realize a direct writing technique that could selectively write nanodiamond in designed patterns."

The ability to selectively "write" lines of diamond on surfaces could be practical for various potential applications including biosensors, quantum computing, fuel cells and next-generation computer chips.

The technique works by using a multilayered film that includes a layer of graphite topped with a glass cover sheet. Exposing this layered structure to an ultrafast-pulsing laser instantly converts the graphite to an ionized plasma and creates a downward pressure. Then the graphite plasma quickly solidifies into diamond. The glass sheet confines the plasma to keep it from escaping, allowing it to form a nanodiamond coating.

"These are super-small <u>diamonds</u> and the coating is super-strong, so it could be used for high-temperature sensors," Cheng said.

Research findings are detailed in a paper that appeared online in the Nature journal *Scientific Reports*. The paper was authored by former Purdue doctoral students Yuefeng Wang, Yingling Yang, Ji Li and Martin Y. Zhang; postdoctoral research associate Jiayi Shao; doctoral students Qiong Nian and Liang Tang; and Cheng.

The researchers made the discovery while studying how to strengthen metals using a thin layer of graphite and a nanosecond-pulsing laser. A



doctoral student noticed that the laser was either causing the graphite to disappear or turn semi-transparent.

"The black coating of graphite was gone, but where did it go?" Cheng said.

Subsequent research proved the <u>graphite</u> had turned into diamond. The Purdue researchers have named the process confined pulse laser deposition (CPLD).

The research team confirmed that the structures are diamond using a variety of techniques including transmission electron microscopy, X-ray diffraction and the measurement of electrical resistance.

A U.S. patent application has been filed on the concept through the Purdue Office of Technology Commercialization. More research is needed to commercialize the technique, Cheng said.

**More information:** Direct Laser Writing of Nanodiamond Films from Graphite under Ambient Conditions, Qiong Nian, Yuefeng Wang, Yingling Yang, Ji Li, Martin Y. Zhang, Jiayi Shao, Liang Tang & Gary J. Cheng, *Scientific Reports*, 2014.

## ABSTRACT

Synthesis of diamond, a multi-functional material, has been a challenge due to very high activation energy for transforming graphite to diamond, and therefore, has been hindering it from being potentially exploited for novel applications. In this study, we explore a new approach, namely confined pulse laser deposition (CPLD), in which nanosecond laser ablation of graphite within a confinement layer simultaneously activates plasma and effectively confine it to create a favorable condition for nanodiamond formation from graphite. It is noteworthy that due to the local high dense confined plasma created by transparent confinement



layer, nanodiamond has been formed at laser intensity as low as 3.7 GW/cm2, which corresponds to pressure of 4.4 GPa, much lower than the pressure needed to transform graphite to diamond traditionally. By manipulating the laser conditions, semi-transparent carbon films with good conductivity (several  $k\Omega/Sq$ ) were also obtained by this method. This technique provides a new channel, from confined plasma to solid, to deposit materials that normally need high temperature and high pressure. This technique has several important advantages to allow scalable processing, such as high speed, direct writing without catalyst, selective and flexible processing, low cost without expensive pico/femtosecond laser systems, high temperature/vacuum chambers.

## Provided by Purdue University

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