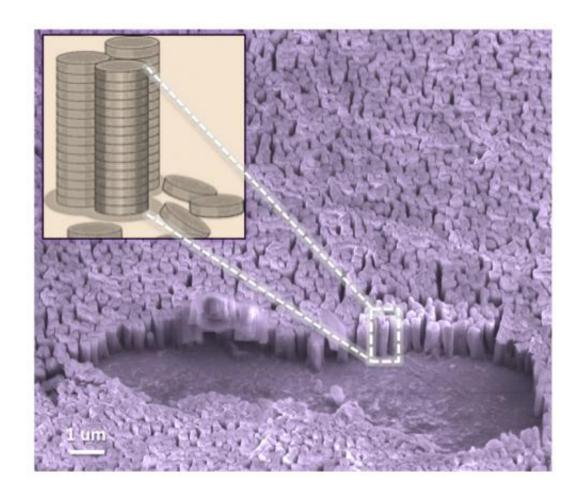


Blades of grass inspire advance in organic solar cells

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Vertical nanopillars are ideal geometries for getting around the challenges of producing polymer architecture to boost power-conversion efficiency of light to electricity to power electronic devices. Credit: UMass AMherst

Using a bio-mimicking analog of one of nature's most efficient light-



harvesting structures, blades of grass, an international research team led by Alejandro Briseno of the University of Massachusetts Amherst has taken a major step in developing long-sought polymer architecture to boost power-conversion efficiency of light to electricity for use in electronic devices.

Briseno, with colleagues and graduate students at UMass Amherst and others at Stanford University and Dresden University of Technology, Germany, report in the current issue of *Nano Letters* that by using single-crystalline organic nanopillars, or "nanograss," they found a way to get around dead ends, or discontinuous pathways, that pose a serious drawback when using blended systems known as bulk heterojunction donor-acceptor, or positive-negative (p-n), junctions for harvesting energy in organic solar cells.

Briseno's research group is one of very few in the world to design and grow organic single-crystal p-n junctions. He says, "This work is a major advancement in the field of organic solar cells because we have developed what the field considers the 'Holy Grail' architecture for harvesting light and converting it to electricity." The breakthrough in morphology control should have widespread use in solar cells, batteries and vertical transistors, he adds.

Briseno explains, "For decades scientists and engineers have placed great effort in trying to control the morphology of p-n junction interfaces in organic solar cells. We report here that we have at last developed the ideal architecture composed of organic single-crystal vertical nanopillars." Nanopillars are nanoscale, engineered surfaces with billions of organic posts that resemble blades of grass, and like grass blades they are particularly effective at converting light to energy.

The advance not only addresses the problem of dead ends or discontinuous pathways that make for inefficient energy transfer, but it



also solves some instability problems, where the materials in mixed blends of polymers tend to lose their phase-separated behavior over time, degrading energy transfer, the polymer chemist says. Also, materials in blended systems tend to be amorphous to semi-crystalline at best and "this is a disadvantage since charge transport is more efficient in highly crystalline systems."

Specifically, to control the molecular orientation and packing at electrode surfaces, the team combined knowledge about graphene and organic crystals. Though it was difficult, Briseno says, they managed to get the necessary compounds to stack like coins. Stacked compounds are ideal for charge transport since this configuration has the largest charge transport anisotropy. Charge transport anisotropy is a phenomenon where electrons flow faster along a particular crystallographic direction due to close molecule-molecule interactions. In this case, the anisotropy is along the nanopillar, perpendicular to the substrate.

Briseno says, "The biggest challenge in producing this architecture was finding the appropriate substrate that would enable the molecules to stack vertically. We had exploited essentially every substrate possible until we finally succeeded with graphene," he adds, which happened by accident when an undergraduate chose the wrong substrate to grow crystals on.

"For over a week the student was growing vertical crystals and we didn't even realize until we imaged the surface of the substrate with a scanning electron microscope. We were shocked to see little crystals standing upright! We ultimately optimized the conditions and determined the mechanism of crystallization," the polymer chemist adds.

Vertical nanopillars are ideal geometries for getting around these challenges, Briseno says, "because charge separation/collection is most efficient perpendicular to the plastic device. In this case, our nanopillars



highly resemble nanograss. Our systems share similar attributes of grass such as high density array system, vertical orientations and the ability to efficiently convert light into energy."

The technique is simple, inexpensive and applicable to a library of donor and acceptor compounds that are commercially available, he notes. "We envision that our nanopillar solar cells will appeal to low-end energy applications such as gadgets, toys, sensors and short lifetime disposable devices."

More information: *Nano Letters*, pubs.acs.org/doi/abs/10.1021/nl501933q

Provided by University of Massachusetts Amherst

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