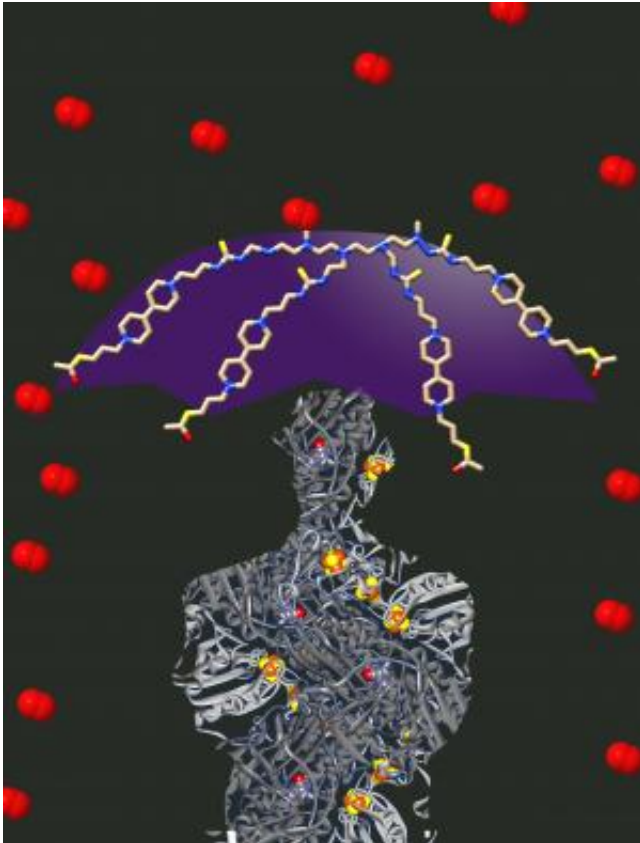


A protecting umbrella against oxygen

August 4 2014



Redox hydrogel works like a protective umbrella against oxygen. Credit: RESOLV

In the development of fuel cells the effort of generations of scientist and engineers have led to efficient and stable catalysts based on noble metals. These catalysts have reached the required threshold in terms of performance for applications such as electric cars. However, the high

costs of the scarce noble metals render their widespread application economically less viable.

In a paper published this week in the journal *Nature Chemistry*, researchers from the Center for Electrochemical Sciences – CES at the Ruhr-University Bochum and from the Max-Planck-Institute for Chemical Energy Conversion in Mülheim an der Ruhr report a novel concept to work with efficient and possibly cheaper catalysts. A kind of buffer protects the catalysts against the hostile conditions encountered in fuel cells, which have been to date dismissed utilization.

Hydrogenases, an alternative to platinum?

Bio-catalysts for hydrogen oxidation – so called hydrogenases – are also part of nature. They developed just by using elements available to living organisms, that means without any [noble metals](#). Complex hydrogenases evolved based solely on abundant elements such as nickel and iron. The most efficient hydrogenases reach the turnover rate of platinum and the supply of the elements they consist of is virtually unlimited. "For this reason Hydrogenases are potentially interesting alternatives to noble metals", says Prof. Dr. Wolfgang Schuhmann (Professorship for analytical chemistry at the RUB). Though hydrogenases are not able to work while being under the constraints encountered in a fuel cell. Traces of oxygen and extreme electrical potentials cause deactivation processes.

Redox hydrogel, a shield for efficient but sensitive catalysts

The team in Bochum and Mülheim focused nevertheless on a new strategy to accommodate sensitive catalyst to the working conditions of standard fuel cells. The key idea is to shield the catalyst in a protective matrix with properties specifically designed to prevent the deactivation

processes. Instead of contacting the hydrogenase directly to the electrode, an immobilization in a redox hydrogel shall protect the construct. It is designed to serve both as a redox buffer and an oxygen scavenger. Hence, within the hydrogel film, neither high potential, nor oxygen affects the bio-catalyst. Under working conditions the hydrogel-modified fuel cell is able to convert chemical energy from hydrogen into electrical energy over several weeks, while in absence of the hydrogel, the hydrogenase is deactivated within seconds.

A major step toward a novel fuel cell redesign

"The hydrogel concept opens the possibility to integrate a variety of other sensitive biological or artificial [catalyst](#) for which the intrinsic stability cannot be further increased", says Prof. Wolfgang Lubitz, Director at the MPI CEC. "This is a major step toward a novel [fuel cell](#) redesign, which may reposition them at the forefront in the race toward global sustainable energy industry."

More information: "A redox hydrogel protects hydrogenase from high-potential deactivation and oxygen damage." Nicolas Plumeré et al. In: *Nature Chemistry* (2014) [DOI: 10.1038/nchem.2022](https://doi.org/10.1038/nchem.2022)

Provided by Ruhr-Universitaet-Bochum

Citation: A protecting umbrella against oxygen (2014, August 4) retrieved 25 April 2024 from <https://phys.org/news/2014-08-umbrella-oxygen.html>

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.
