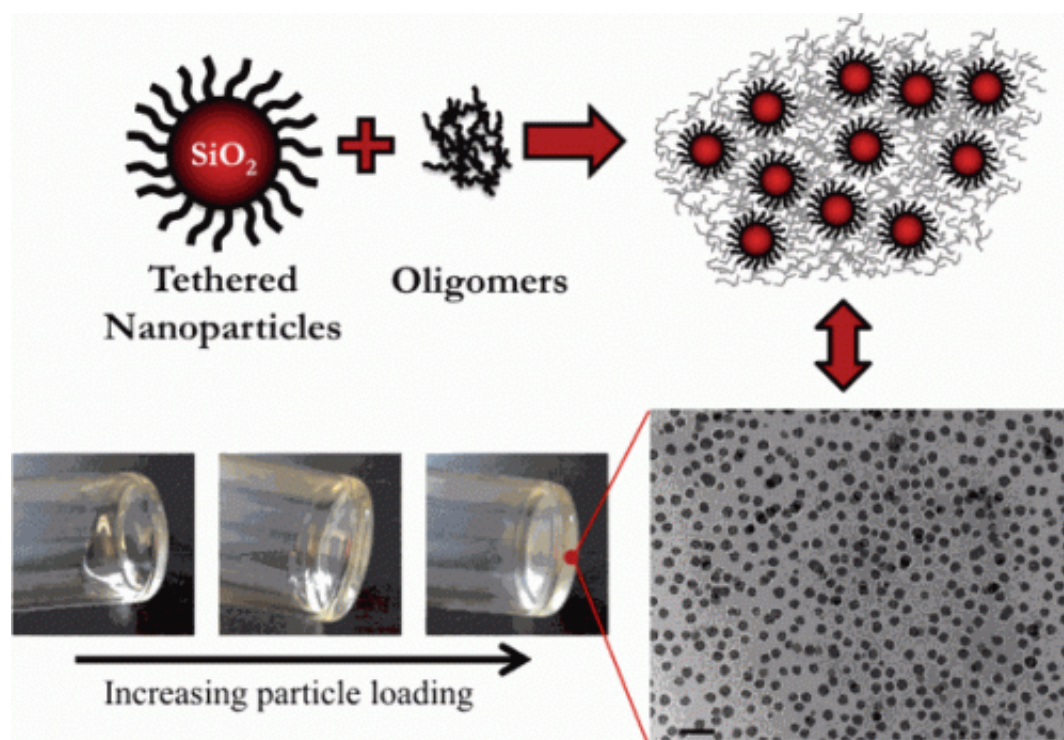


Arrangements and mobility of soft nanoparticles in dense suspensions

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A schematic of soft nanocolloidal suspensions comprising of soft polyethylene glycol (PEG) tethered silica nanoparticles suspended in PEG oligomers. The pictures show the variation in physical characteristics with increasing particle loading and the electron micrograph shows the well-dispersed particles in these suspensions.

(Phys.org) —The unusual properties of water, including its anomalous thermal expansion and density anomaly, have intrigued researchers for

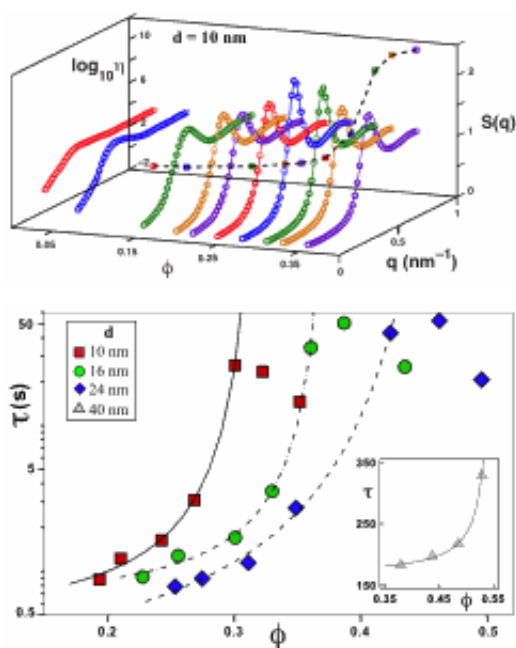
decades. These properties are notoriously hard to investigate experimentally owing to the inherently small length scales and complex interactions that appear to govern the physics of these materials. Studies of small particles (colloids) dispersed in solvents, known as colloidal suspensions, used as models for atomic and molecular liquids have shown that some of these anomalies can be engineered in colloidal suspensions of soft particles.

A report published in *Physical Review Letters* describes research carried out at the U.S. Department of Energy Office of Science's Advanced Photon Source (APS) at Argonne National Laboratory that elucidates the arrangements and mobility of soft [nanoparticles](#) in dense suspensions that mirror the anomalies observed in complex liquids like water. This discovery, which is the first instance of experimental observation of such behavior in a colloidal suspension, allows for an extension of the toolbox of the experimental physicist interested in employing suspensions to mimic molecular liquids, with the added advantage of readily accessible length and time scales.

The research team, with members from Cornell University and Argonne, synthesized soft nanoparticles by densely tethering small polymers onto the surface of silica nanoparticles. Small-angle x-ray scattering (SAXS) and x-ray photon correlation spectroscopy (XPCS) measurements were carried out at X-ray Science Division beamlines 12-ID-B and 8-ID-I at the APS to reveal the equilibrium structure and the characteristics of particle motion, respectively.

It was found that the particle arrangements become more disordered and move faster when more particles are added into the suspension beyond a critical particle volume fraction, coinciding with a sharp increase in the resistance by the system of grafted nanoparticles to physical deformations.

This is in contrast to the usual situation where increasing the concentration of particles in a dilute suspension decreases the space available for placing new particles, thus increasing particle ordering and slowing them down.



Evolution of structure factor ($S(q)$) and relaxation times (τ) with particle loading (ϕ) for soft nanoparticle suspensions. The appearance of maxima in the height of the first peak of $S(q)$ clearly indicates the structural anomaly, while the maxima in relaxation times indicates the transport anomaly in the soft nanocolloidal suspensions. These results were obtained from SAXS and XPCS measurements, respectively.

"It becomes easier for any particle in these suspensions to diffuse when they are more surrounding by their neighbors; the counterintuitive nature of this situation can be illustrated with the following analogy - it is easier to make a run and score a touch down when the opposing team has fifteen people in defense," said Samanvaya Srivastava, a senior graduate

student at Cornell University and lead author of the Physical Review Letters article.

The anomalous behavior of [particles](#) suspended in water and other complex liquids has been long argued to exist for systems with soft repulsion, which is characterized by a potential energy that exhibits a finite width over which particle interaction occurs. These empirical findings lend support to an emerging consensus from simulation studies and provide for a model system for studying systems with soft repulsive interactions.

More information: Samanvaya Srivastava, Lynden A. Archer, Suresh Narayanan, "Structure and Transport Anomalies in Soft Colloids", *Phys. Rev. Lett.* 110, 148302 (2013). [DOI: 10.1103/PhysRevLett.110.148302](https://doi.org/10.1103/PhysRevLett.110.148302)

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