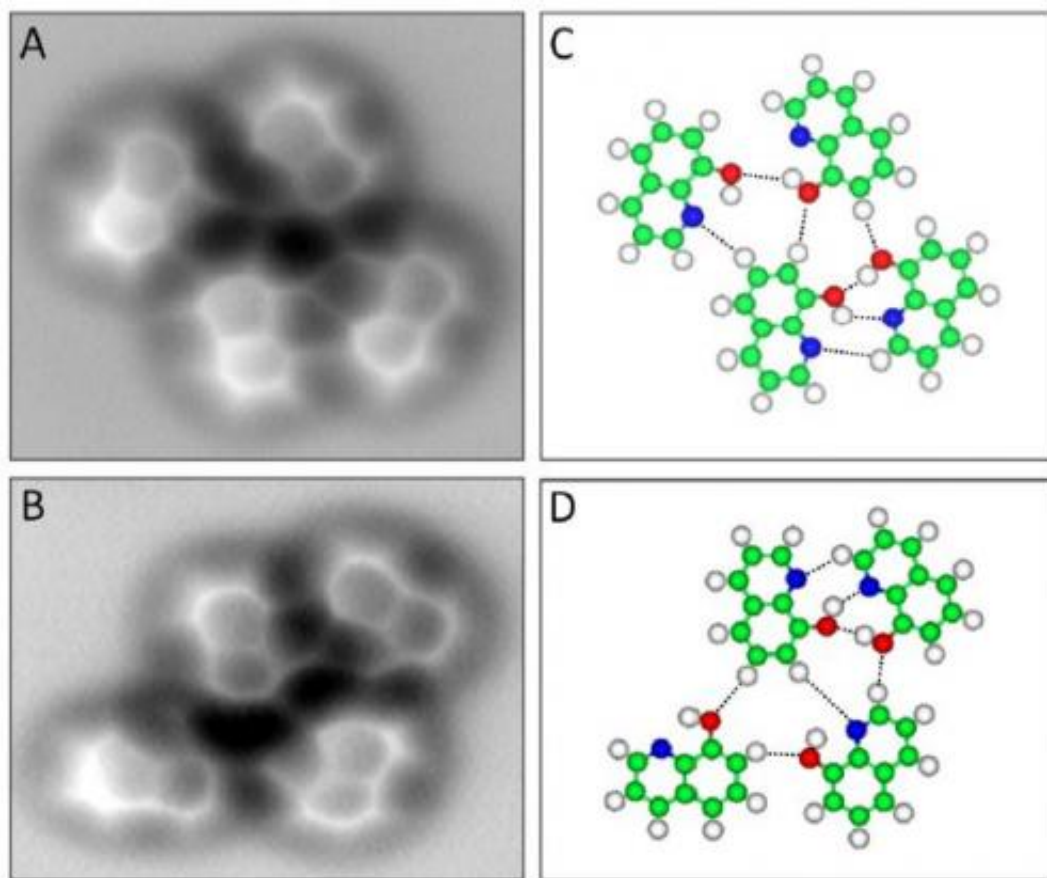


Researchers create image of weak hydrogen bond using AFM

October 1 2013, by Bob Yirka



AFM measurements of 8-hq assembled clusters on Cu(111). (A and B) Constant-height frequency shift images of typical molecule assembled clusters, and their corresponding structure models (C and D). Imaging parameters: $V = 0$ V, $A = 100$ p.m., $\Delta z = +10$ pm. Image size: (A) 2.3 nm \times 2.0 nm, (B) 2.5 nm \times 1.8 nm. The dashed lines in (C) and (D) indicate likely H bonds between 8-hq molecules. Credit: *Science* DOI:10.1126/science.1242603

(Phys.org) —Researchers at China's National Center for Nanoscience and Technology and Renmin University have used Atomic Force Microscopy (AFM) to create an image of the weak hydrogen bonds present in a molecule. In their paper published in the journal *Science*, the team describes how they used the non-contact form of AFM to capture an image of weak hydrogen bonds in a 8-hydroxyquinoline molecule (8hq).

Until recently, the most accurate images of molecules were obtained using scanning tunneling microscopy, recent advances with AFM, however (particularly the addition of a carbon monoxide molecule to the probe tip) have made it the method of choice for obtaining actual images of molecules and the bonds that hold them together. In this new effort, the research team has advanced the science of AFM by capturing an image of the weak [hydrogen bonds](#) present in a 8hq molecule.

Understanding how hydrogen bonds work is a very important part of science—they are responsible, for example, for holding together strands of DNA.

Currently, there are two ways to capture images using AFM: contact, and non-contact imaging. With contact imaging, the tip of the mechanical probe is made to touch the surface of the material being analyzed. The amount of deflection of the probe tip as its dragged is used to create an image. In non-contact mode, the tip is brought near to the surface, but doesn't touch it. An image is created by measuring changes to the oscillating tip probe from weak forces emanating from the source.

The team in China used the non-contact form of AFM and chose 8hq as a test subject because it's a relatively flat molecule. The resulting image is the first created using AFM to show weak hydrogen bonds, and is important because there is still debate about the nature of hydrogen bonds. For many years it was considered to be purely an electrostatic interaction—new evidence has cast doubt on that idea, suggesting that it

might be at least partially chemical. The new image doesn't clear up the debate, but does offer some intriguing possibilities for the future as it demonstrates that as AFM matures, it will offer more and more evidence of the true nature of [molecules](#) and how they interact.

More information: Real-Space Identification of Intermolecular Bonding with Atomic Force Microscopy, Published Online September 26 2013. *Science* [DOI: 10.1126/science.1242603](https://doi.org/10.1126/science.1242603)

ABSTRACT

We report a real-space visualization of the formation of hydrogen bonding in 8-hydroxyquinoline (8-hq) molecular assemblies on a Cu(111) substrate using noncontact atomic force microscopy (NC-AFM). The atomically resolved molecular structures enable a precise determination of the characteristics of hydrogen bonding networks, including the bonding sites, orientations, and lengths. The observation of bond contrast was interpreted by ab initio density functional calculations, which indicated the electron density contribution from the hybridized electronic state of the hydrogen bond. Intermolecular coordination between the dehydrogenated 8-hq and Cu adatoms was also revealed by the submolecular resolution AFM characterization. The direct identification of local bonding configurations by NC-AFM would facilitate detailed investigations of intermolecular interactions in complex molecules with multiple active sites.

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