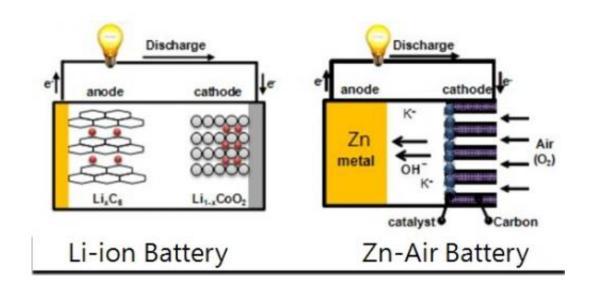


Affordable, scalable process of carbon nanotube-based catalyst outperforming platinum for electric-automobiles

July 3 2013



This shows the structure of Li-ion Battery and Zn-air Battery. Credit: Ulsan National Institute of Science and Technology

Korean researchers from Ulsan National Institute of Science and Technology (UNIST), S. Korea, developed a novel bio-inspired composite electrocatalyst outperforming platinum.

This research work was published on June 25, in the journal *Nature Communications*.



The research team from UNIST, S. Korea, developed an inexpensive and scalable bio-inspired composite electrocatalyst, iron phthalocyanine with an axial ligand anchored on single-walled carbon nanotubes, demonstrating a higher electrocatalytic activity for oxygen reduction than the state-of-the-art Pt/C catalysts as well as an exceptional durability during cycling in an alkaline media.

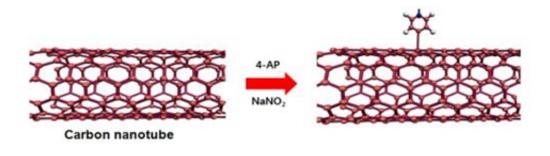
Electrocatalysts for oxygen reduction are critical components that may dramatically enhance the performance of fuel cells and metal-air batteries, which are perceived to be the power for future <u>electric</u> vehicles.

Currently Pt and its alloy are known as the most efficient catalysts for activation of the oxygen reduction reaction. However, the application of them is limited due to their high costs and scarce reserves.

Scientists worldwide are speculating to find better catalysts which outperform <u>platinum catalysts</u> with low cost and simple production processes.

The UNIST research team led by Prof. Jaephil Cho, dean of the Interdisciplinary School of Green Energy, UNIST, demonstrated a new strategy to rationally design inexpensive and durable electrochemical oxygen reduction catalysts for metal-air batteries and fuel cells.





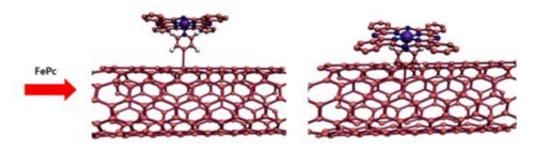


Diagram of synthesis of new carbon nanotube-based catalyst

This is a diagram of synthesis of new carbon nanotube-based catalyst. Credit: UNIST

The research team designed a new class of ORR catalysts using pyridinefunctionalized carbon nanotubes (CNTs) to anchor FePc molecules and provide the axial ligand for the iron centre. At the same time, the CNTs provide an easy pathway for fast <u>electron transfer</u> from the current collector to the ORR active sites.

The resulting material, bio-inspired FePc-Py-CNTs catalyst has shown outstanding durability and electrocatalytic activity for ORR in an alkaline media, offering better performance than a commercial Pt/C catalyst. Compared to other unpyrilysed metal macrocycles catalysts, this bio inspired FePc-Py-CNTs catalyst has achieved a much longer cycle life, reaching more than 1,000 cycles in a durability test.



"I believe the FePc-Py-CNTs catalysts is a technologically promising candidate for practical applications in metal-air batteries and alkaline fuel cells," said Prof. Cho. "The origin of the enhanced performance for this bio-inspired catalysts in aromatic macrocycle, provides important insight into rational design of metal macrocycles catalysts for other applications such as solar harvesting and catalysts for other redox reactions."

More information: Promotion of Oxygen reduction by a bio-inspired tethered iron phthalocyanine carbon nanotube-based catalyst, *Nature Communications*, 2013.

Provided by Ulsan National Institute of Science and Technology

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