

Future looks bright for carbon nanotube solar cells

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Materials Science and Engineering Assistant Professor Michael Arnold. Credit: David Nevala.

(Phys.org) —In an approach that could challenge silicon as the predominant photovoltaic cell material, University of Wisconsin-Madison materials engineers have developed an inexpensive solar cell that exploits carbon nanotubes to absorb and convert energy from the sun.

The advance could lead to solar panels just as efficient, but much less expensive to manufacture, than current panels.

The proof-of-concept carbon nanotube solar cell can convert nearly 75 percent of the light it absorbs into electricity, says Michael Arnold, an assistant professor of [materials science and engineering](#) at UW-Madison and a pioneer in developing carbon nanotube-based materials for solar energy applications. "We've made a really fundamental key step in demonstrating that it will be possible to use these new carbon nanotube materials for solar cells one day," he says.

Arnold and PhD student Matthew Shea described the development in a paper published June 17, 2013, in the online edition of the journal *Applied Physics Letters*.

Silicon is abundant and an efficient solar energy gatherer, yet is expensive to process and manufacture into solar panels. As a result, researchers are studying alternative materials—among them, carbon nanotubes.

Recent advances have afforded researchers a greater level of control over the [chemical makeup](#) of carbon nanotubes, which in turn has opened the door to myriad applications. The thin spaghetti-like tubes are easy and inexpensive to manufacture, stable and durable, and are both good light absorbers and [electrical conductors](#).

Much of the current carbon nanotube solar cell research centers around proven solar cell materials that use mixed-in nanotubes to conduct the electrical charge. "That's only using half the capabilities that nanotubes offer," says Arnold, whose prior work with carbon nanotubes for transistors inspired him to explore applications in solar energy.

Building on a half-decade of research—including foundational studies

by PhD student Dominick Bindl—Arnold and Shea developed a solar cell that uses carbon nanotubes to collect light and convert it to electricity. "We're starting from the ground up and trying to get high efficiency out of the nanotubes," says Arnold. "We're trying to get as much power conversion as possible out of our material, and that's what's unique about our work."

Essentially, the proof-of concept solar cell is an ultrathin sheet, or film, of carbon nanotubes layered atop another thin sheet of a material called buckminsterfullerene, or C_{60} . The nanotubes absorb the bulk of the sunlight and retain the positive charge, while the C_{60} draws the negative charge.

Solar cell efficiency is the percentage of [solar energy](#) shining on a cell that the cell actually converts into electrical energy. When Arnold and his students began this research five years ago, their solar cells achieved power-conversion efficiencies of only about a millionth of a percent. Today - in contrast with the 15-percent average efficiency of conventional silicon solar cells—their proof of concept is 1 percent efficient.

While that number might seem low, Arnold is optimistic it can rise—in part because the sun-catching carbon nanotube layer of the [proof-of-concept](#) solar cell is just a few atoms thick. And, the cell converts approximately 75 percent of the sunlight it absorbs into electricity. "Of the light that is absorbed, we're converting most of it," says Arnold.

The next step in boosting that efficiency already is underway. The researchers now are focusing on augmenting the thickness of the [carbon nanotube](#) thin film from a mere 5 nanometers to at least 100—which, according to their theoretical models, ultimately could put the power-conversion efficiency of their [solar cells](#) in line with that of silicon cells. "What our work shows is that you will be able to get as high efficiency

as silicon eventually, and that's why we're excited," says Arnold.

More information:

apl.aip.org/resource/1/applab/v102/i24/p243101_s1

Provided by University of Wisconsin-Madison

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