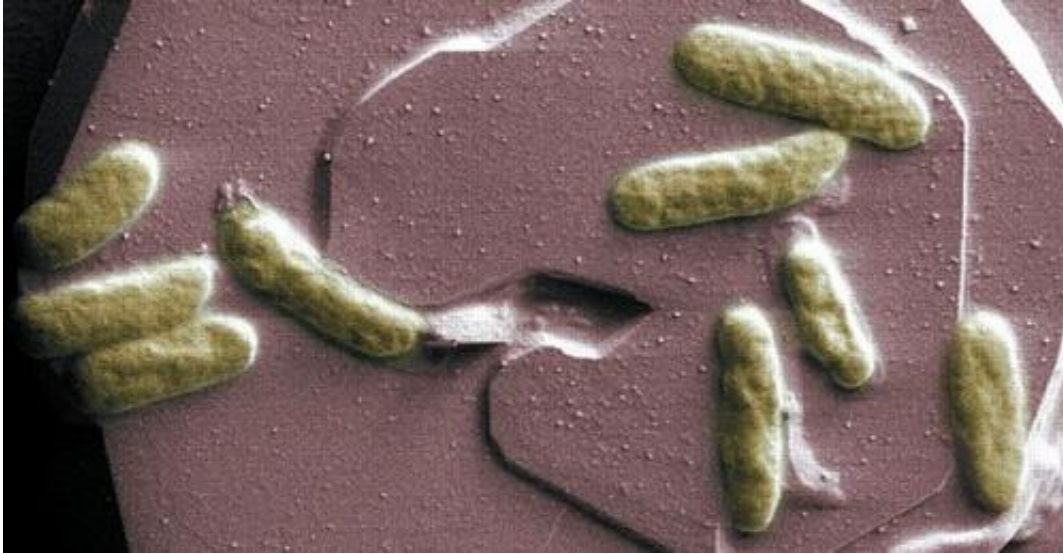


Breakthrough in race to create 'bio-batteries'

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Shewanella oneidensis bacteria. Credit: Alice Dohnalkova.

(Phys.org) —Scientists at the University of East Anglia have made an important breakthrough in the quest to generate clean electricity from bacteria.

Findings published today in the journal *Proceedings of the National Academy of Sciences (PNAS)* show that proteins on the surface of [bacteria](#) can produce an electric current by simply touching a [mineral surface](#).

The research shows that it is possible for bacteria to lie directly on the surface of a metal or mineral and transfer electrical charge through their

cell membranes. This means that it is possible to 'tether' bacteria directly to electrodes – bringing scientists a step closer to creating efficient [microbial fuel cells](#) or 'bio-batteries'.

The team collaborated with researchers at Pacific Northwest National Laboratory in Washington State in the US. The project was funded by the Biotechnology and Biological Sciences Research Council (BBSRC) and the US Department of Energy.

[Shewanella oneidensis](#) (pictured) is part of a family of [marine bacteria](#). The research team created a [synthetic version](#) of this bacteria using just the proteins thought to shuttle the electrons from the inside of the microbe to the rock.

They inserted these proteins into the lipid layers of vesicles, which are small capsules of [lipid membranes](#) such as the ones that make up a bacterial membrane. Then they tested how well electrons travelled between an [electron donor](#) on the inside and an iron-bearing mineral on the outside.

Lead researcher Dr Tom Clarke from UEA's school of Biological Sciences said: "We knew that bacteria can transfer electricity into metals and minerals, and that the interaction depends on special proteins on the surface of the bacteria. But it was not been clear whether these proteins do this directly or indirectly through an unknown mediator in the environment.

"Our research shows that these proteins can directly 'touch' the mineral surface and produce an electric current, meaning that is possible for the bacteria to lie on the surface of a metal or mineral and conduct electricity through their cell membranes.

"This is the first time that we have been able to actually look at how the

components of a bacterial cell membrane are able to interact with different substances, and understand how differences in metal and mineral interactions can occur on the surface of a cell.

"These bacteria show great potential as microbial fuel cells, where electricity can be generated from the breakdown of domestic or agricultural waste products.

"Another possibility is to use these bacteria as miniature factories on the surface of an electrode, where chemical reactions take place inside the cell using electrical power supplied by the electrode through these proteins."

Biochemist Liang Shi of Pacific Northwest National Laboratory said: "We developed a unique system so we could mimic electron transfer like it happens in cells. The electron transfer rate we measured was unbelievably fast - it was fast enough to support bacterial respiration."

The finding is also important for understanding how carbon works its way through the atmosphere, land and oceans.

"When organic matter is involved in reducing iron, it releases carbon dioxide and water. And when iron is used as an energy source, bacteria incorporate carbon dioxide into food. If we understand [electron transfer](#), we can learn how bacteria controls the carbon cycle," said Shi.

More information: 'Rapid electron exchange between surface-exposed bacterial cytochromes and Fe(III) minerals' by Thomas A Clarke, Gaye White, Julea N Butt, and David J Richardson (all UEA, UK), and Zhri Shi, Liang Shi, Zheming Wang, Alice C Dohnalkova, Matthew J Marshall, James K Fredrickson and John M Zachara (all PNNL, USA) is published by *PNAS* on Monday, March 25. (Will be available at www.pnas.org/lookup/doi/10.1073/pnas.1220074110)

Provided by University of East Anglia

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