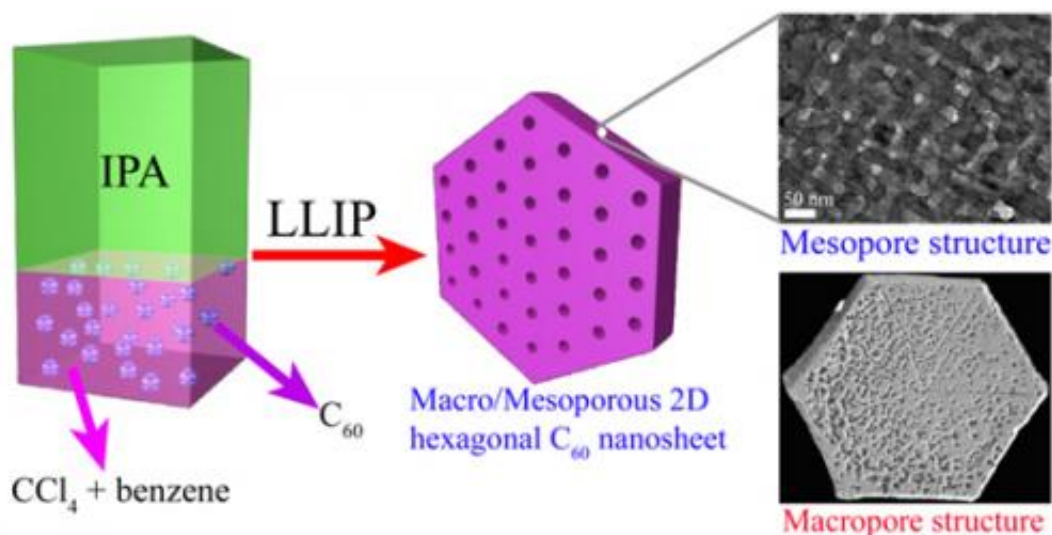


# Fullerene crystals with bimodal pore architectures

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Synthetic route of producing mesoporous crystalline fullerene using a liquid-liquid interface

A research group headed by MANA Scientist Dr. Lok Kumar Shrestha of the Supermolecules Unit, for the first time demonstrated template-free novel mesoporous carbon material: fullerene (C<sub>60</sub>) crystals with bimodal pore architectures and having highly crystallized framework. Experiments have proven that this novel meso- and macroporous material show better electrochemical performance compared to pristine C<sub>60</sub> due to higher electrochemically active surface areas.

In this research, novel fullerene (C<sub>60</sub>) crystals with bimodal mesoporous

and macroporous structures composed of a highly crystallized framework has prepared by using a liquid-liquid interfacial precipitation (LLIP) method involving the interface between isopropyl alcohol (IPA) and a [saturated solution](#) of  $C_{60}$  in a mixture of benzene and [carbon tetrachloride](#) ( $CCl_4$ ). The resulting mesoporous  $C_{60}$  exhibits two-dimensional (2D) hexagonal plate morphology.

Porosity and electrochemically active surface area could be flexibly controlled by increasing the mixing fraction of  $CCl_4$  and benzene. The synergistic effect of mixing solvents ( $CCl_4$  and benzene) is mainly responsible for the formation of such [porous structure](#). Otherwise, in an individual IPA/ $CCl_4$  and IPA/benzene system, 2D plate like and 1D nanowiskers morphology without pores are observed. In solution-based crystallization (LLIP method), solvent molecule gets entrapped during crystallization, which upon slow release/or evaporation creates a porous structure. It is expected that this methodological innovation will be a milestone for the production of highly crystalline carbon-based materials offering better performance in catalytic, electrochemical, and sensing properties.

These research results are recently published in *Journal of American Chemical Society*, 2013, 135, 586-589.

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