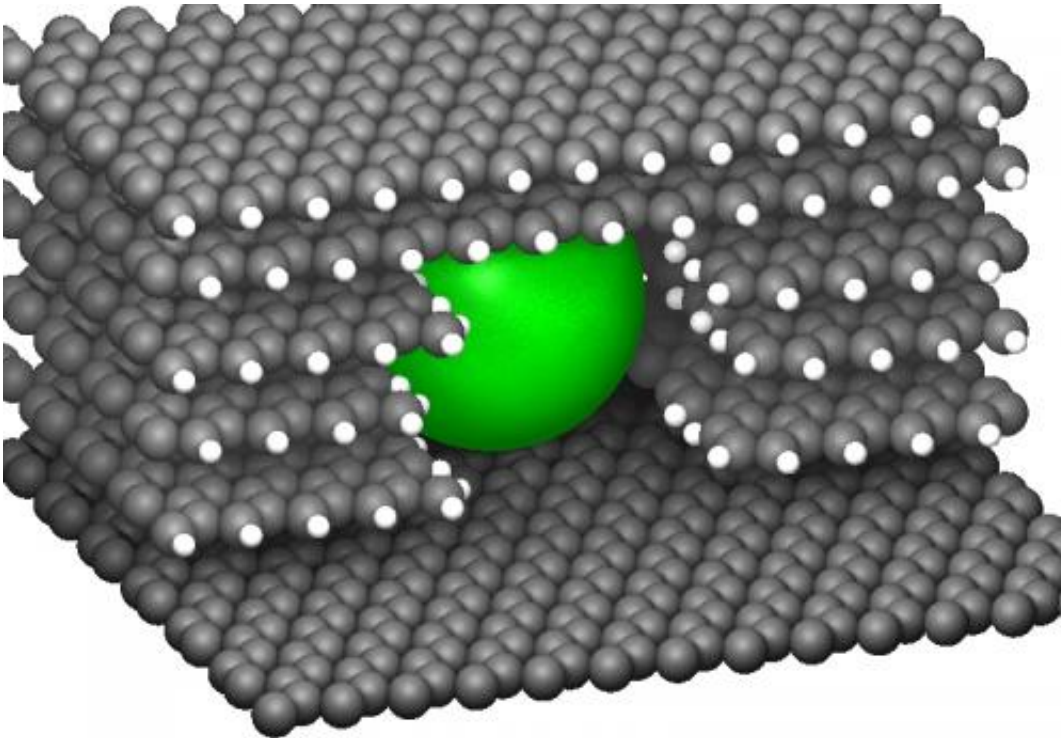


Nanoparticles digging the world's smallest tunnels

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Graphite consists of layered carbon atoms. A metal particle bores into the graphite sample from the edges of these layers. Credit: KIT

The world's smallest tunnels have a width of a few nanometers only. Researchers from Karlsruhe Institute of Technology (KIT) and Rice University, USA, have dug such tunnels into graphite samples. This will allow structuring of the interior of materials through self-organization in

the nanometer range and tailoring of nanoporous graphite for applications in medicine and battery technology. Results are now presented in the scientific journal *Nature Communications*.

The tunnels are manufactured applying nickel nanoparticles to graphite which then is heated in the presence of [hydrogen gas](#). The surface of the [metal particles](#), that measure a few nanometers only, serves as a catalyst removing the carbon atoms of the graphite and converting them by means of hydrogen into the gas methane. Through [capillary forces](#), the nickel particle is drawn into the "hole" that forms and bores through the material. The size of the tunnels obtained in the experiments was in the range of 1 to 50 nanometers, which about corresponds to one thousandth of the diameter of a human hair.

To furnish proof of the real existence of these graphite tunnels, the researchers have made use of scanning electron and [scanning tunneling microscopy](#). "Microscopes, in fact, image only the upper layers of the sample," the principal authors of the study, Maya Lukas and Velimir Meded from KIT's Institute of Nanotechnology, explain. "The tunnels below these upper layers, however, leave atomic structures on the surface whose courses can be traced and which can be assigned to the nanotunnels by means of the very detailed scanning tunneling microscopy images and based on computerized simulations." In addition, the depth of the tunnels was determined precisely by means of a series of images taken by a [scanning electron microscope](#) from different perspectives.

Porous graphite is used, for example, in the electrodes of lithium ion batteries. The charge time could be reduced using materials with appropriate pore sizes. In medicine, porous graphite could serve as a carrier of drugs to be released over longer periods of time. Replacing graphite by nonconductive materials, e.g. boron nitride, with atomic structures similar to that of graphite, the tunnels could serve as basic

structures for nanoelectronic components such as novel sensors or solar cells.

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