

Nano-velcro clasps heavy metal molecules in its grips

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(Phys.org)—Researchers have devised a simple, system based on nanoparticles, to detect mercury as well as others pollutants. This technology makes it possible to easily and inexpensively test for these substances in water and, more importantly, in the fish that we eat. This will be published in *Nature Materials*.

Mercury, when dumped in lakes and rivers, accumulates in fish, and often ends up on our plates. A Swiss-American team of researchers led by Francesco Stellacci at the Ecole Polytechnique Fédérale de Lausanne (EPFL) and Bartosz Grzybowski at Northwestern University has devised a simple, inexpensive system based on nanoparticles, a kind of nanovelcro, to detect and trap this toxic pollutant as well as others. The particles are covered with <u>tiny hairs</u> that can grab onto toxic <u>heavy</u> <u>metals</u> such as mercury and cadmium. This technology makes it possible



to easily and inexpensively test for these substances in water and, more importantly, in the fish that we eat. Their new method can measure methyl mercury, the most common form of <u>mercury pollution</u>, at unprecedentedly small attomolar concentrations. The system is outlined in an article appearing September 9, 2012 in the journal <u>Nature Materials</u>.

Methyl mercury, toxic and difficult to monitor

Researchers are particularly interested in detecting mercury. Its most common form, methyl mercury, accumulates as one goes up the food chain, reaching its highest levels in large <u>predatory fish</u> such as tuna and swordfish. In the US, France and Canada, public health authorities advise pregnant women to limit <u>fish consumption</u> because mercury can compromise nervous system development in the developing fetus.

"The problem is that current monitoring techniques are too expensive and complex," explains Constellium Chair holder at EPFL and co-author Francesco Stellacci. "We periodically test levels of mercury in drinking water, and if those results are good, we make the assumption that levels are acceptable in between those testing periods." But industrial discharge fluctuates.

A simple, inexpensive new technology

The technology developed by the Swiss-American team is simple to use. A strip of glass covered with a film of "hairy" nanoparticles is dipped into the water. When an ion – a positively charged particle, such as a methyl mercury or cadmium ion – gets in between two hairs, the hairs close up, trapping the pollutant.

A voltage-measuring device reveals the result; the more ions there are



trapped in the nano-velcro, the more electricity it will conduct. So to calculate the number of trapped particles, all one needs to do is measure the voltage across the nanostructure.

By varying the length of the nano-hairs, the scientists can target a particular kind of pollutant. "The procedure is empirical," explains Stellacci. Methyl mercury, fortunately, has properties that make it extremely easy to trap without accidentally trapping other substances at the same time; thus the results are very reliable.



Commercial strip of glass covered with a film of 'hairy' nanoparticles, a kind of a 'nano-velcro,' that can be dipped into water to measure mercury levels. Credit: The Grzybowski Group, Northwestern University

The interesting aspect of this approach is that the 'reading' glass strip could costs less than 10 dollars, while the measurement device will cost



only a few hundreds of dollars. The analysis can be done in the field, so the results are immediately available. "With a conventional method, you have to send samples to the laboratory, and the analysis equipment costs several million dollars," notes Stellacci.

Convincing tests in Lake Michigan and Florida

The researchers tested the system in Lake Michigan, near Chicago. Despite the high level of industry in the region, mercury levels were extremely low. "The goal was to compare our measurements to FDA measurements done using conventional methods," explains Stellacci. "Our results fell within an acceptable range."

A mosquito fish from the Everglades in Florida was also tested. This species is not very high on the food chain and thus does not accumulate high levels of mercury in its tissues. "We measured tissue that had been dissolved in acid. The goal was to see if we could detect even minuscule quantities." says Bartosz Grzybowski, Burgess Professor of Chemistry and Director of Non-Equilibrium Energy Research Center at Northwestern University. The United States Geological Survey reported near-identical results after analyzing the same sample.

From quantum to real applications

"I think it is quite incredible," Grzybowski adds, "how the complex principles of quantum tunneling underlying our device translate into such an accurate and practically useful device. It is also notable that our system - through some relatively simple chemical modifications - can be readily adapted to detect other toxic species" Researchers have already demonstrated the detection of cadmium with a very high femtomolar sensitivity.



"With this technology, it will be possible to conduct tests on a much larger scale in the field, or even in fish before they are put on the market," says lead author Eun Seon Cho. This is a necessary public health measure, given the toxic nature of methyl <u>mercury</u> and the extremely complex manner in which it spreads in the environment and accumulates in living tissues.

Provided by Ecole Polytechnique Federale de Lausanne

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