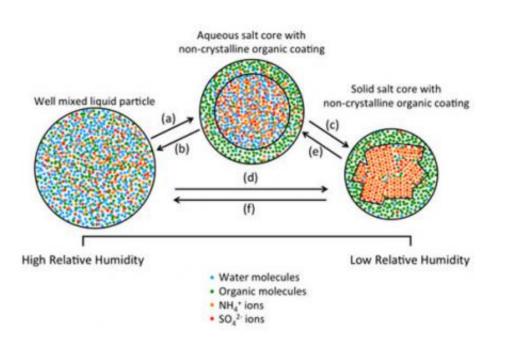


## Tiny airborne pollutants lead double life: study

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Possible phase transitions of particles containing mixtures of organic and inorganic material: a) liquid-liquid phase separation; b) liquid-liquid mixing; c) and d) inorganic efflorescence; and e) and f) inorganic deliquescence. Image courtesy of Lindsay Renbaum-Wolff

University of British Columbia and Harvard researchers have provided visual evidence that atmospheric particles -- which are ubiquitous in the atmosphere, especially above densely populated areas -- separate into distinct chemical compositions during their life cycle.



The observations could have important implications for modelling <u>global</u> <u>climate change</u> and predicting air quality conditions. The <u>tiny particles</u>, which form part of an airborne chemical mix above cities, play a role in pollution by providing a surface for chemical reactions and in climate by reflecting and absorbing <u>solar radiation</u> and by acting as seed surfaces for water condensation and cloud formation.

"We've confirmed experimentally that changes in relative humidity can separate the organic and inorganic material in individual <u>atmospheric</u> <u>particles</u> into distinct liquid phases, much like oil separates from water," says UBC professor Allan Bertram, Director of the collaborative research and training program on <u>atmospheric aerosols</u> at UBC and a principal investigator on the paper.

"Having two liquid phases rather than one can change the rates of chemical reactions on particles, may change the amount of light the particles reflect and absorb, and impact their ability to act as seeds for clouds."

The findings--which used air samples from Atlanta, Georgia, the Harvard Environmental Chamber, and the Pacific Northwest National Laboratory Environmental Chamber--are to published in the <u>Proceedings</u> of the National Academy of Sciences next week.

"I think of it as the beautiful phenomenon when I mixed food coloring, water and vegetable oil in a bottle when I was in grade school," says Harvard researcher Scot Martin, Gordon McKay Professor of Environmental Chemistry at the School of Engineering and Applied Sciences and the Department of Earth and Planetary Sciences.

"More to the point, this phenomenon is really new thinking in the atmospheric sciences, and it completely changes the way we need to think through the reactive chemistry of atmospheric particles, a key



component of urban air quality."

The air in most urban environments contains particles that are mixtures of organic molecules and chemicals like sulfates. When examined, samples from Atlanta revealed distinct liquid phases that were qualitatively similar to idealized particles generated in the laboratory. The liquid-liquid phase separation occurs naturally.

Particulate air pollution is a relatively new area of study, but one of growing concern to researchers, health officials and environmental groups. Increases in the concentration of aerosols are correlated with increased health issues, including cardiopulmonary disorders.

"We need to understand as much as possible about the chemical composition, physical properties and interactions of atmospheric particles if we're going to assess how they impact human health, regional weather patterns, and even global climate change," notes Bertram, with the Department of Chemistry.

## Provided by University of British Columbia

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