

Water, water everywhere—but is it essential to life?

April 13 2012

Proteins are large organic molecules that are vital to every living thing, allowing us to convert food into energy, supply oxygen to our blood and muscles, and drive our immune systems. Since proteins evolved in a water-rich environment, it is generally thought that they are dependent on water to survive and function.

Proteins consist of one or more polypeptides – chains of amino acids held together by peptide bonds. If a [protein](#) in water is heated to temperatures approaching the boiling point of water, these chains will lose their structure and the protein will denature (unfold).

A classic example of denaturing occurs when an egg is hard-boiled: the structures of the proteins in the egg unfold with temperature and stick together creating a solid. In the egg's case, this process cannot be reversed – however there are many examples where cooling the protein results in refolding of the structure.

Previously, it was thought that water was essential to the refolding process, however the Bristol findings suggest this isn't necessarily the case.

Using a spectroscopic technique called circular dichroism, Dr Adam Perriman of Bristol's School of Chemistry and colleagues have shown that the oxygen-carrying protein myoglobin can refold in an environment that is almost completely devoid of water molecules.

Dr Perriman said: "We achieved this by attaching polymer molecules to the surface of the protein and then removing the [water](#) to give a viscous liquid which, when cooled from a temperature as high as 155°C, refolded back to its original structure.

"We then used the Circular Dichroism beamline (B23) at Diamond Light Source, the UK's national synchrotron science facility in Oxfordshire, to track the refolding of the myoglobin structure and were astounded when we became aware of the extremely high thermal resistance of the new material."

These findings could pave the way for the development of new industrial enzymes where hyper-thermal resistance would play a crucial role, in applications ranging from biosensor development to electrochemical reduction of CO₂ to liquid fuels.

More information: 'Hyper-thermal stability and unprecedented re-folding of solvent-free liquid myoglobin' by Alex Brogan, Giuliano Siligardi, Rohanah Hussain, Adam Perriman and Stephen Mann in *Chemical Science* [DOI: 10.1039/C2SC20143G](https://doi.org/10.1039/C2SC20143G)

Provided by University of Bristol

Citation: Water, water everywhere—but is it essential to life? (2012, April 13) retrieved 27 April 2024 from <https://phys.org/news/2012-04-ndash-essential-life.html>

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.
